

Pollution of Earth's Stratosphere:
Ozone Recovery and Chemistry/Climate Coupling

AOSC 433/633 & CHEM 433

Ross Salawitch

Class Web Site: <http://www.atmos.umd.edu/~rjs/class/spr2015>

Motivating questions:

- a) Levels of CFCs have peaked and are slowly declining: are we seeing a response in total ozone column?**
- b) How might climate change (future variations in temperature *and* / or circulation) driven by rising GHGs affect stratospheric ozone?**
- c) Might climate at the surface be affected by stratospheric ozone?**

Lecture 16
7 April 2015

Recovery of the Ozone Layer

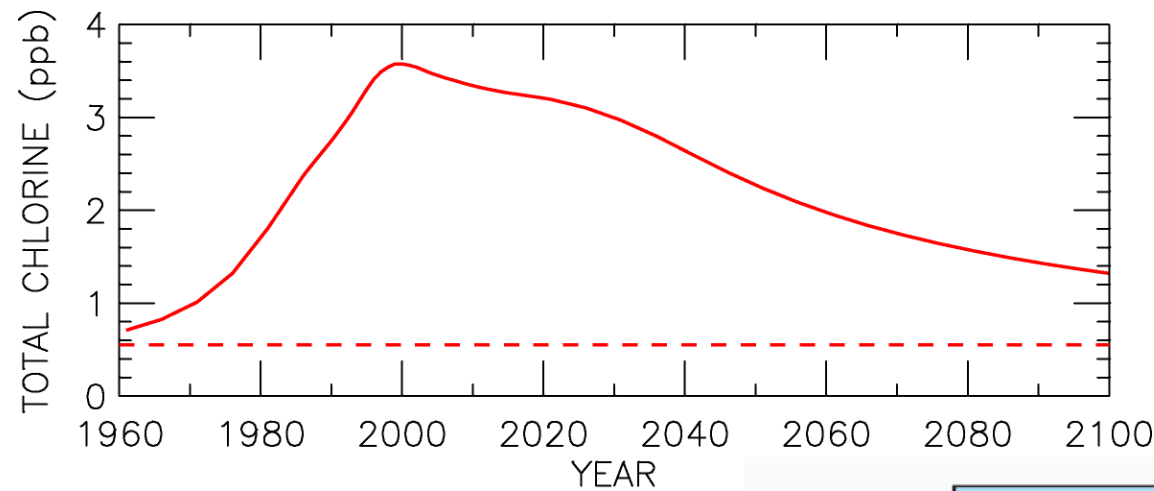


Table 5A-3, WMO/UNEP 2010

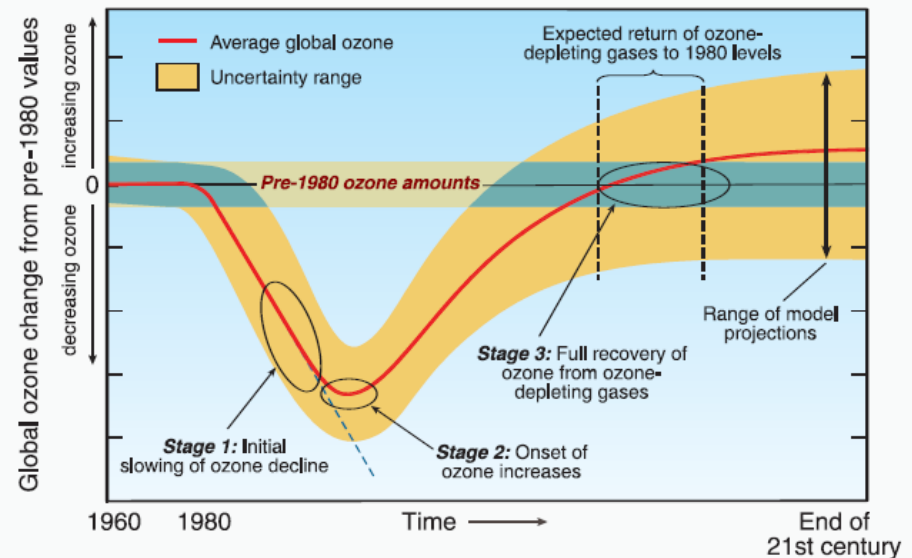
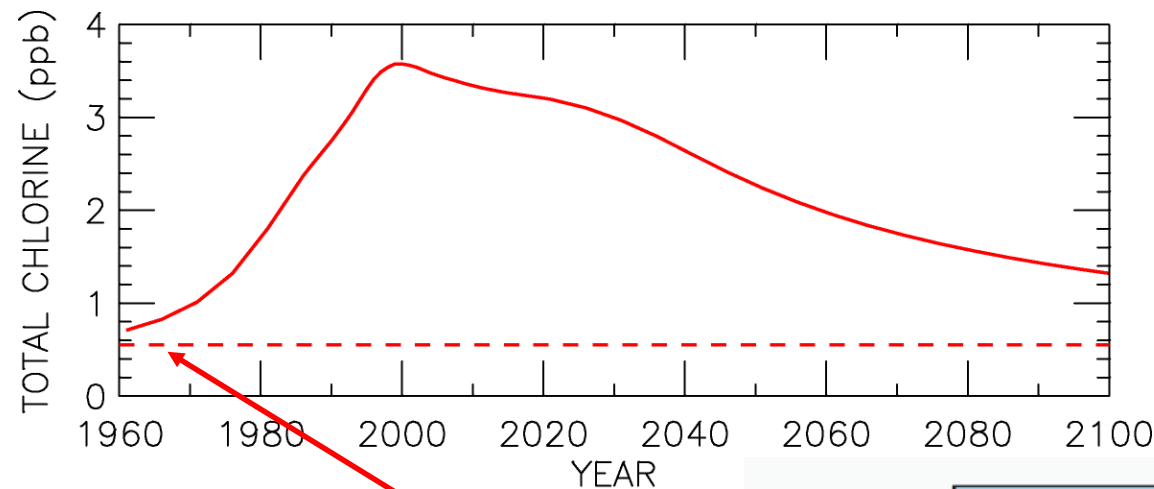


Figure Q19-1. Recovery stages of global ozone. Significant ozone depletion from the release of ozone-depleting gases in human activities first became recognized in the 1980s. The Montreal Protocol provisions are expected to further reduce and eliminate these gases in the atmosphere in the coming decades, thereby leading to the return of ozone amounts to near pre-1980 values. The timeline of the recovery process is schematically illustrated with three stages identified. The large uncertainty range illustrates natural ozone variability in the past and potential uncertainties in global model projections of future ozone amounts. When ozone reaches the full recovery stage, global ozone values may be above or below pre-1980 values, depending on other changes in the atmosphere (see Q20).

Recovery of the Ozone Layer



Why was total chlorine ~0.6 ppb in 1960?

Time series of chlorine content of organic halocarbons that reach the stratosphere. Past values based on direct atmospheric observation. Future values based on projections that include the lifetime for removal of each halocarbon.

Table 5A-3, WMO/UNEP 2010

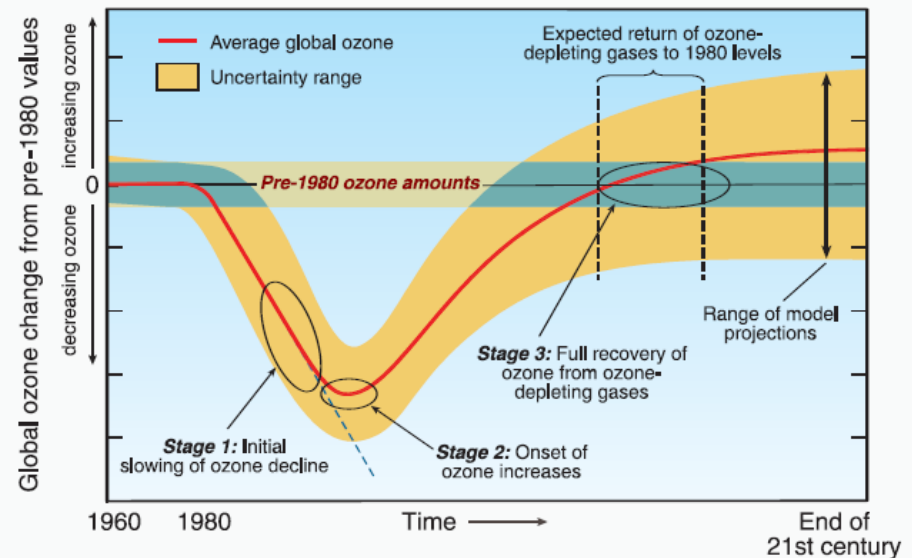
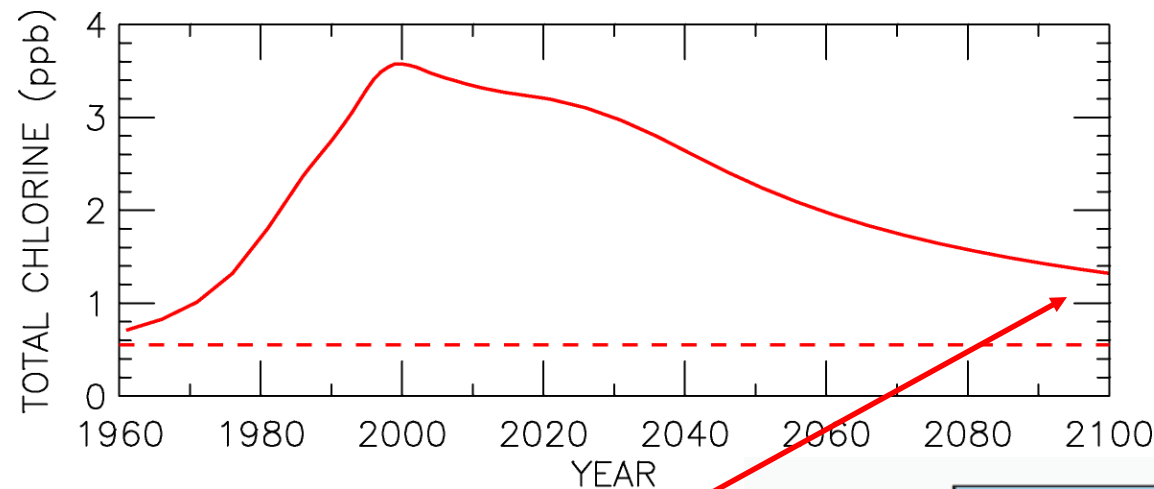


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What very long lived halocarbon is responsible for CCl_y in year 2100 exceeding CCl_y in year 1960 ?

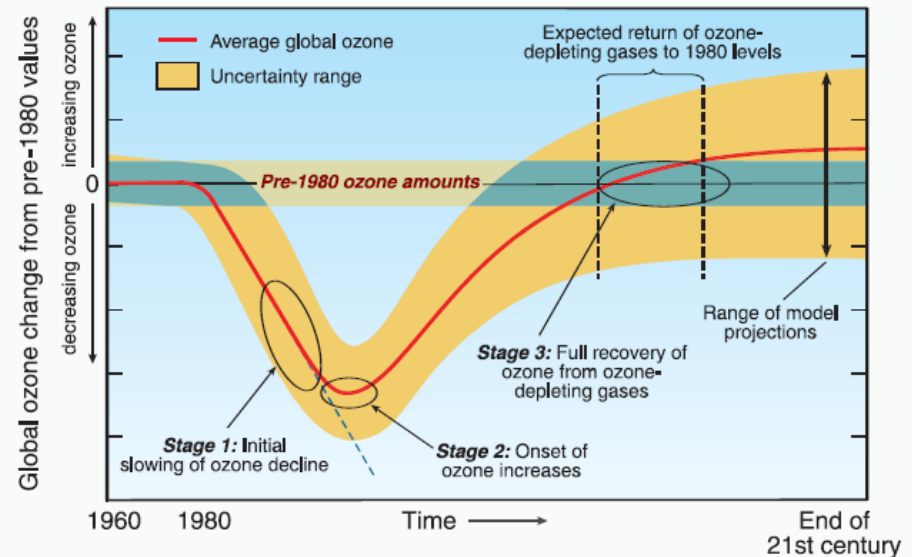
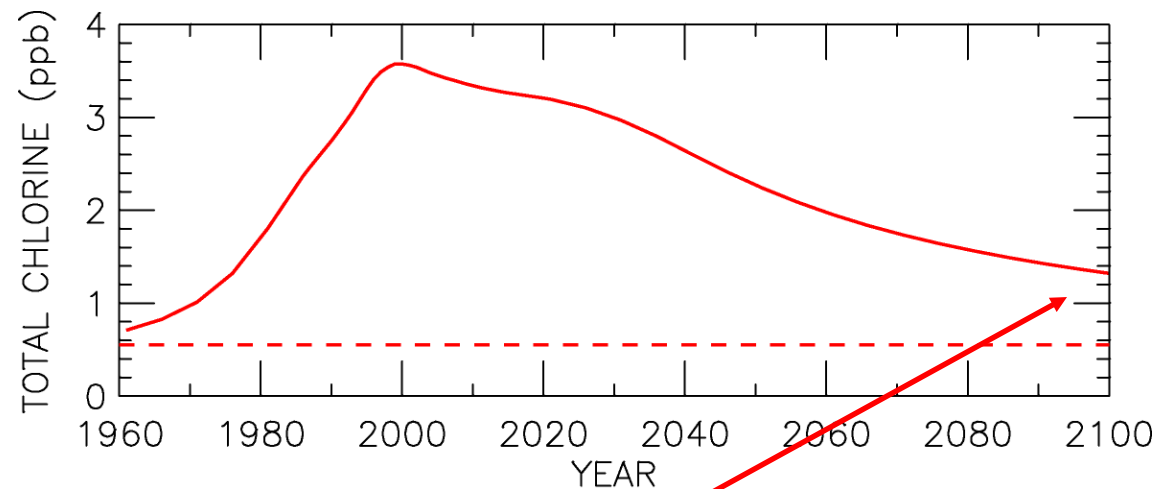


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Recovery of the Ozone Layer



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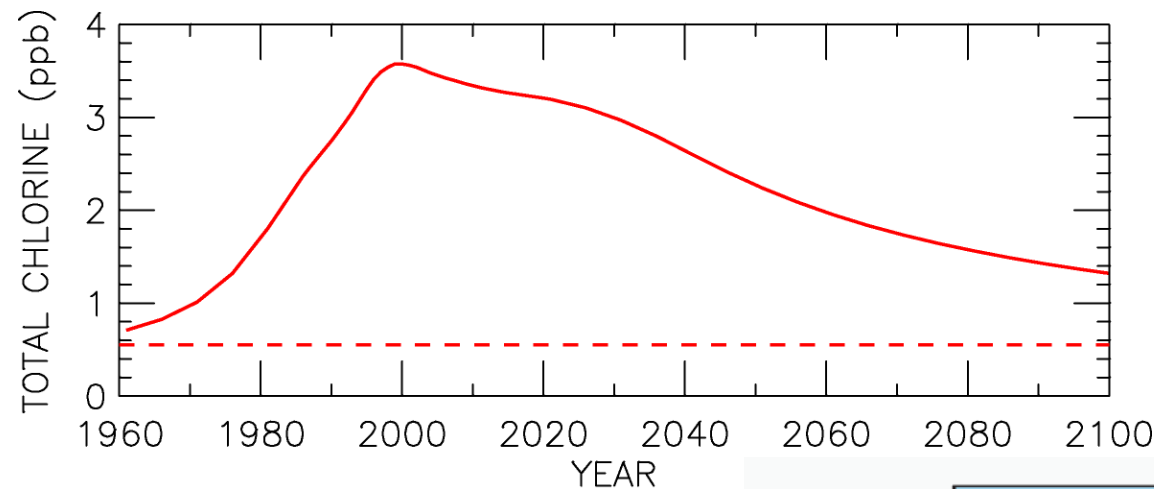
Table 5A-3, WMO/UNEP 2010

Table Q7-1. Atmospheric Lifetimes and Ozone Depletion Potentials of some halogen source & HFC substitute gases.

Gas	Atmospheric Lifetime (years)	Ozone Depletion Potential (ODP) ^c
Halogen source gases		
<i>Chlorine gases</i>		
CFC-11	45	1
CFC-12	100	0.82
CFC-113	85	0.85
Carbon tetrachloride (CCl_4)	26	0.82
HCFCs	1-17	0.01-0.12
Methyl chloroform (CH_3CCl_3)	5	0.16
Methyl chloride (CH_3Cl)	1	0.02
<i>Bromine gases</i>		
Halon-1301	65	15.9
Halon-1211	16	7.9
Methyl bromide (CH_3Br)	0.8	0.66
Very short-lived gases (e.g., CHBr_3)	Less than 0.5	^b very low
Hydrofluorocarbons (HFCs)		
HFC-134a	13.4	0
HFC-23	222	0

Lecture 2, Slide 22

Recovery of the Ozone Layer



Time series of chlorine content of organic halocarbons that reach the stratosphere. Past values based on direct atmospheric observation. Future values based on projections that include the lifetime for removal of each halocarbon.

Table 5A-3, WMO/UNEP 2010

What is especially “odd” about these graphs when scrutinized in tandem ?

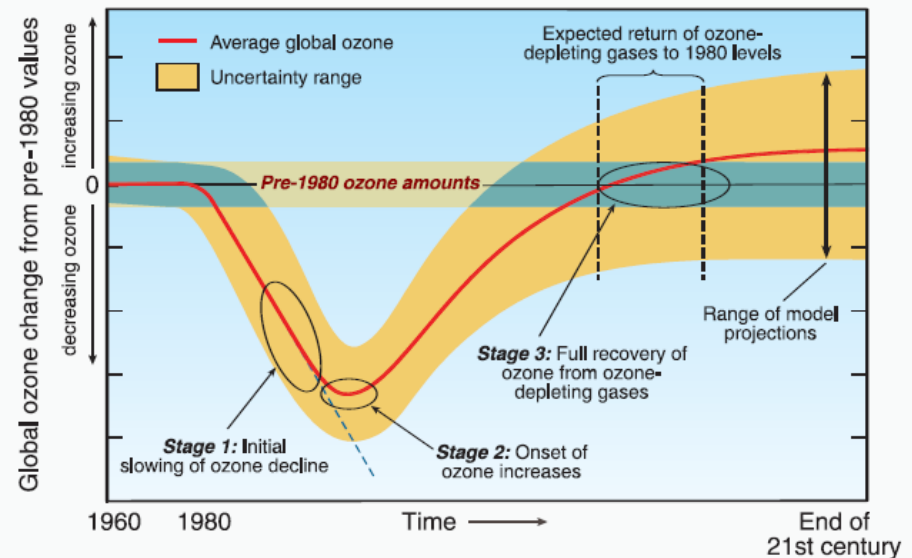
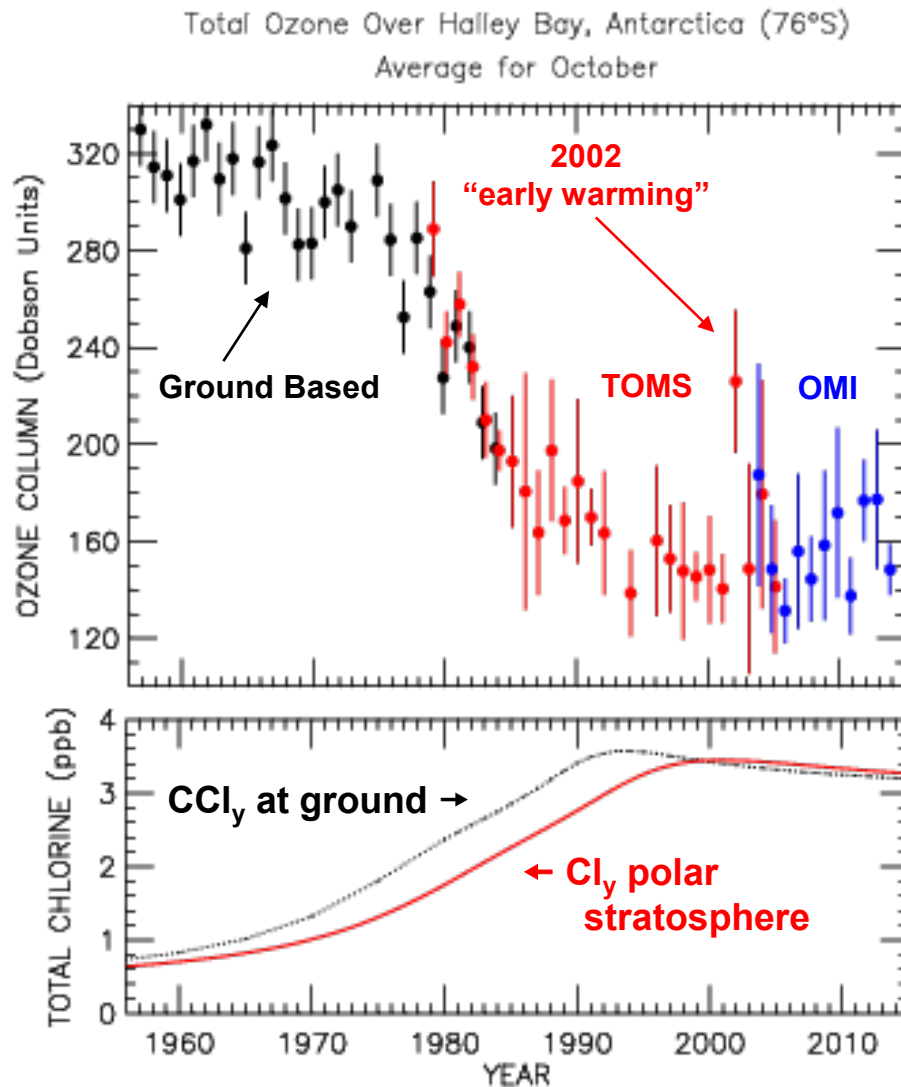


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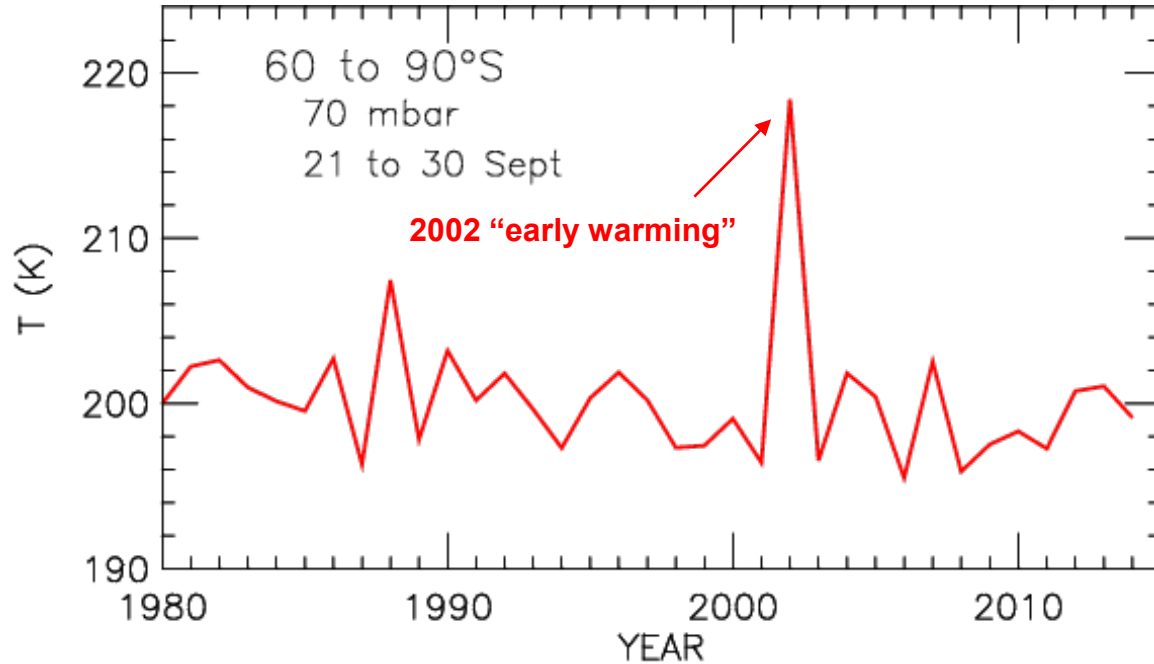
Polar Ozone Loss: Update



How much of this "leveling off" is due to the "leveling off" of halogens ?

Lecture 15, Slide 2
Lecture 15, Slide 21

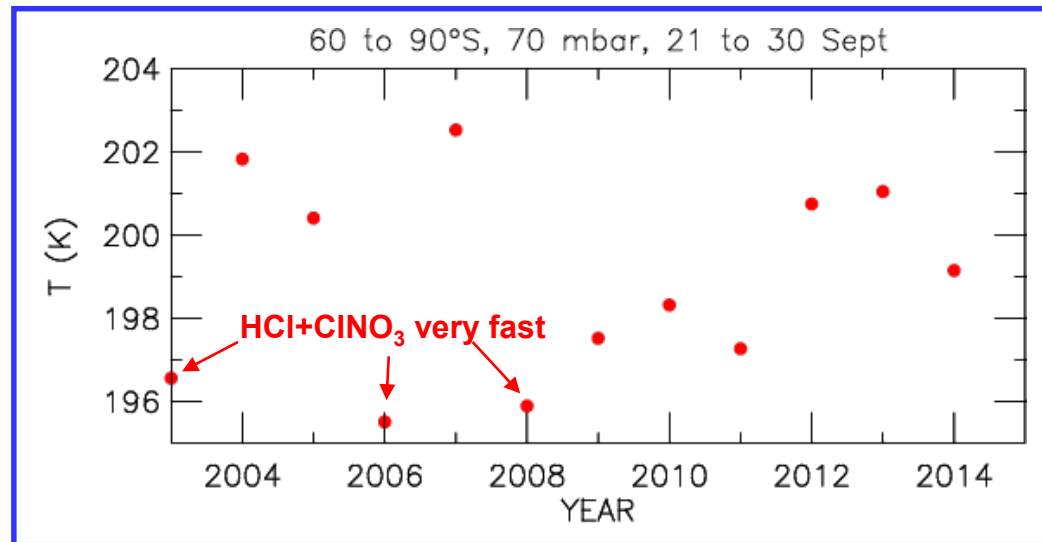
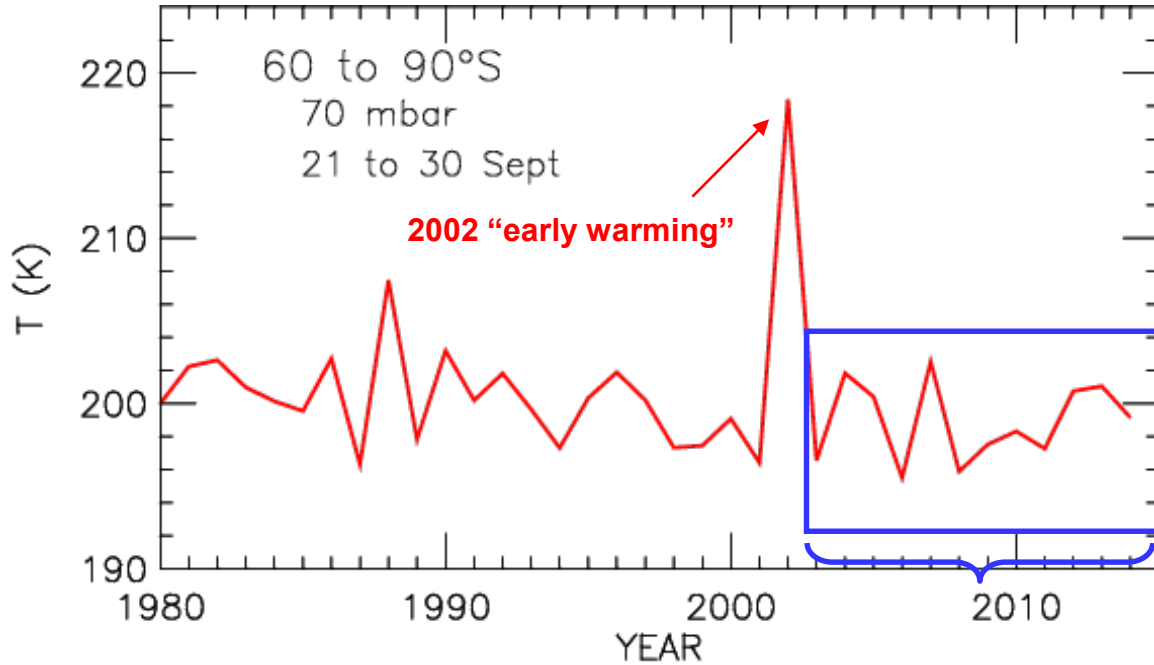
Complication #1



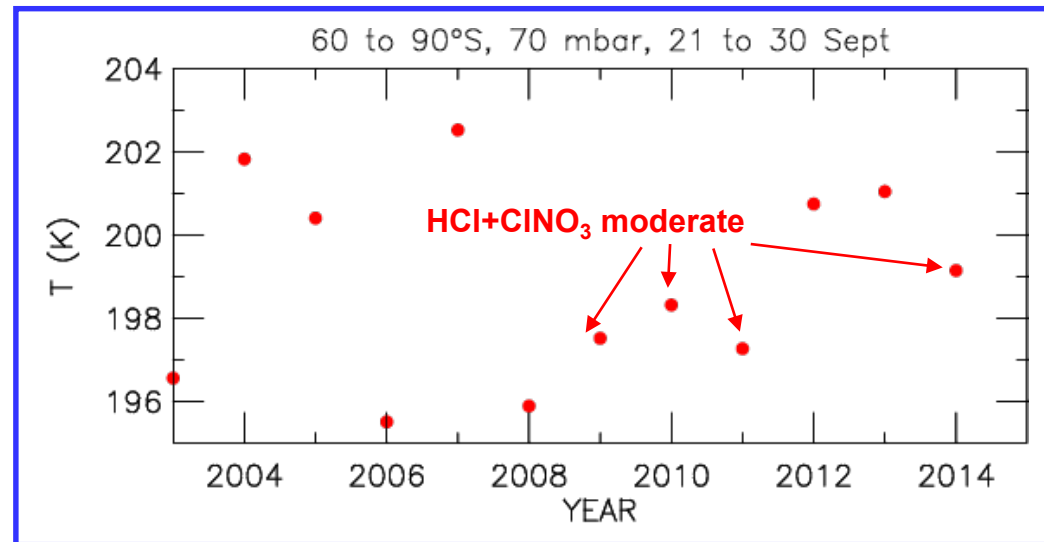
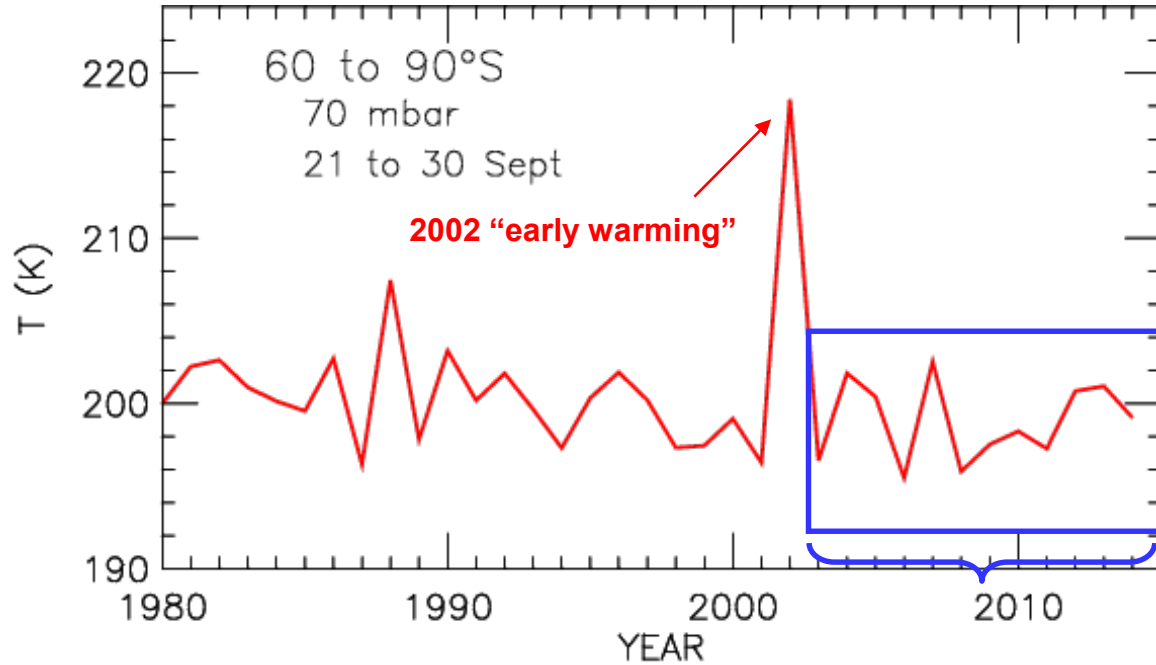
Considerable year to year variability in temperature

Data from http://acdb-ext.gsfc.nasa.gov/Data_services/met/ann_data.html

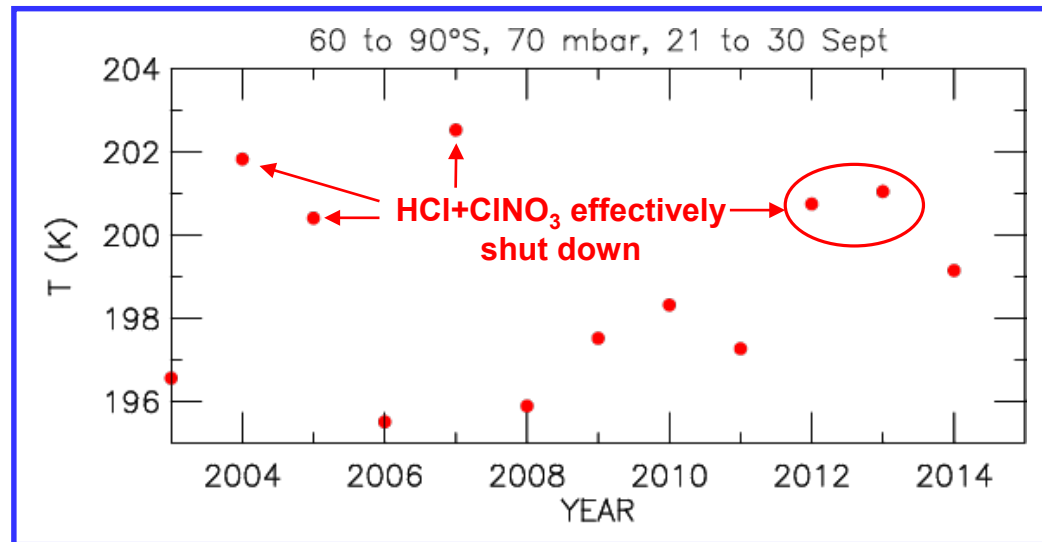
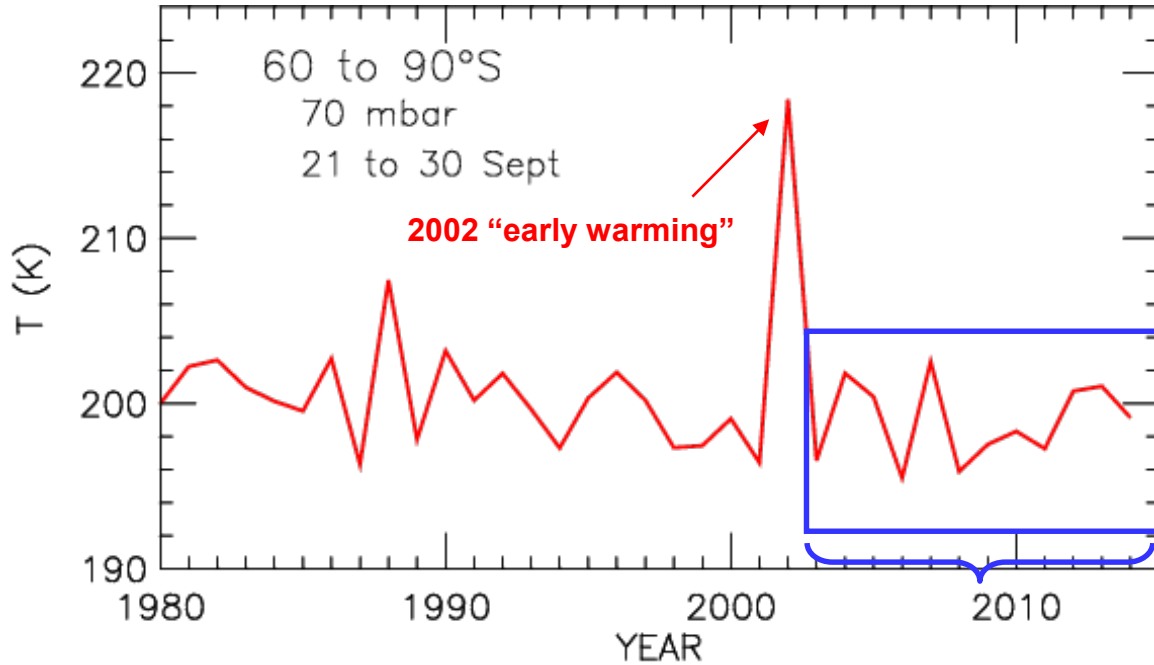
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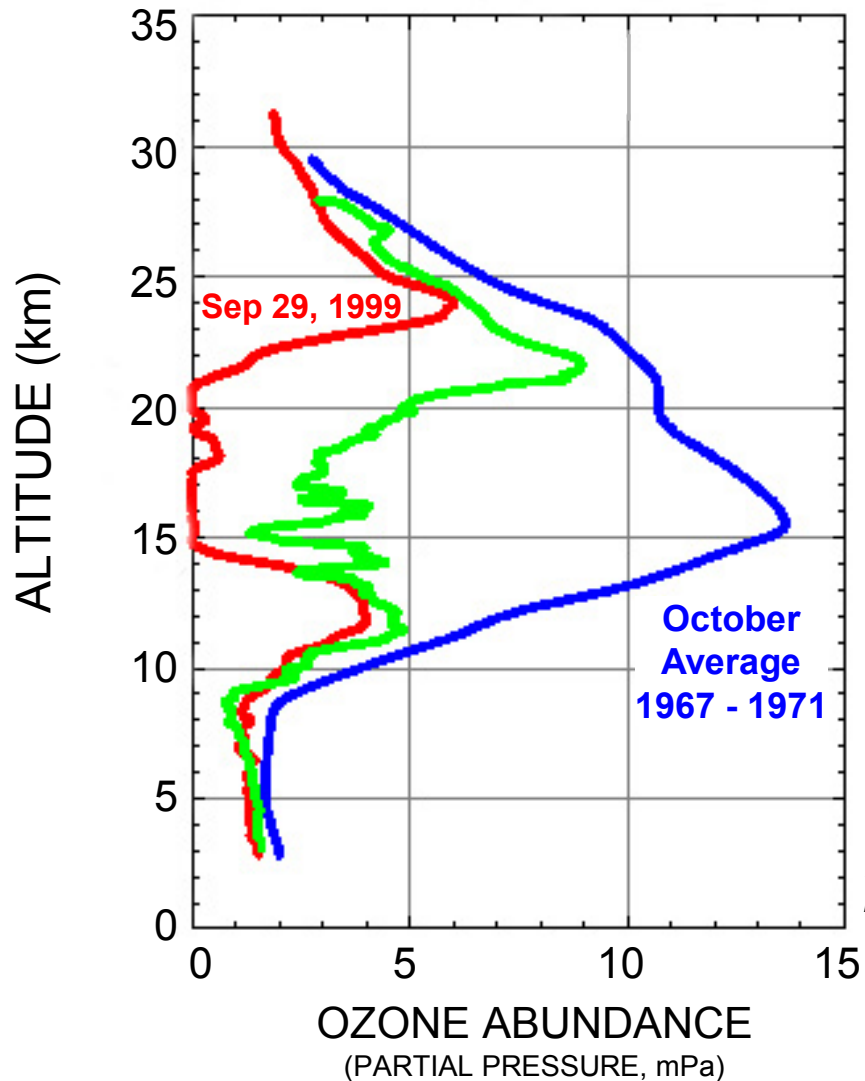
Complication #1



Complication #1



Complication #2

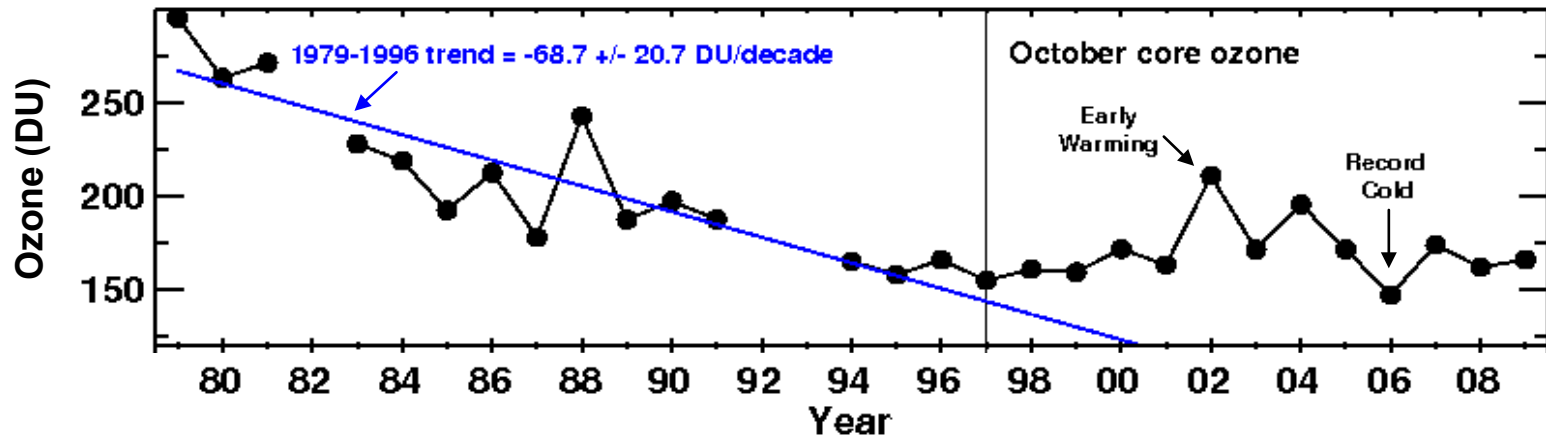


Ozone reaches “zero” over considerable height range.

This “saturation effect” may be the cause of the “leveling off” of the column ozone time series

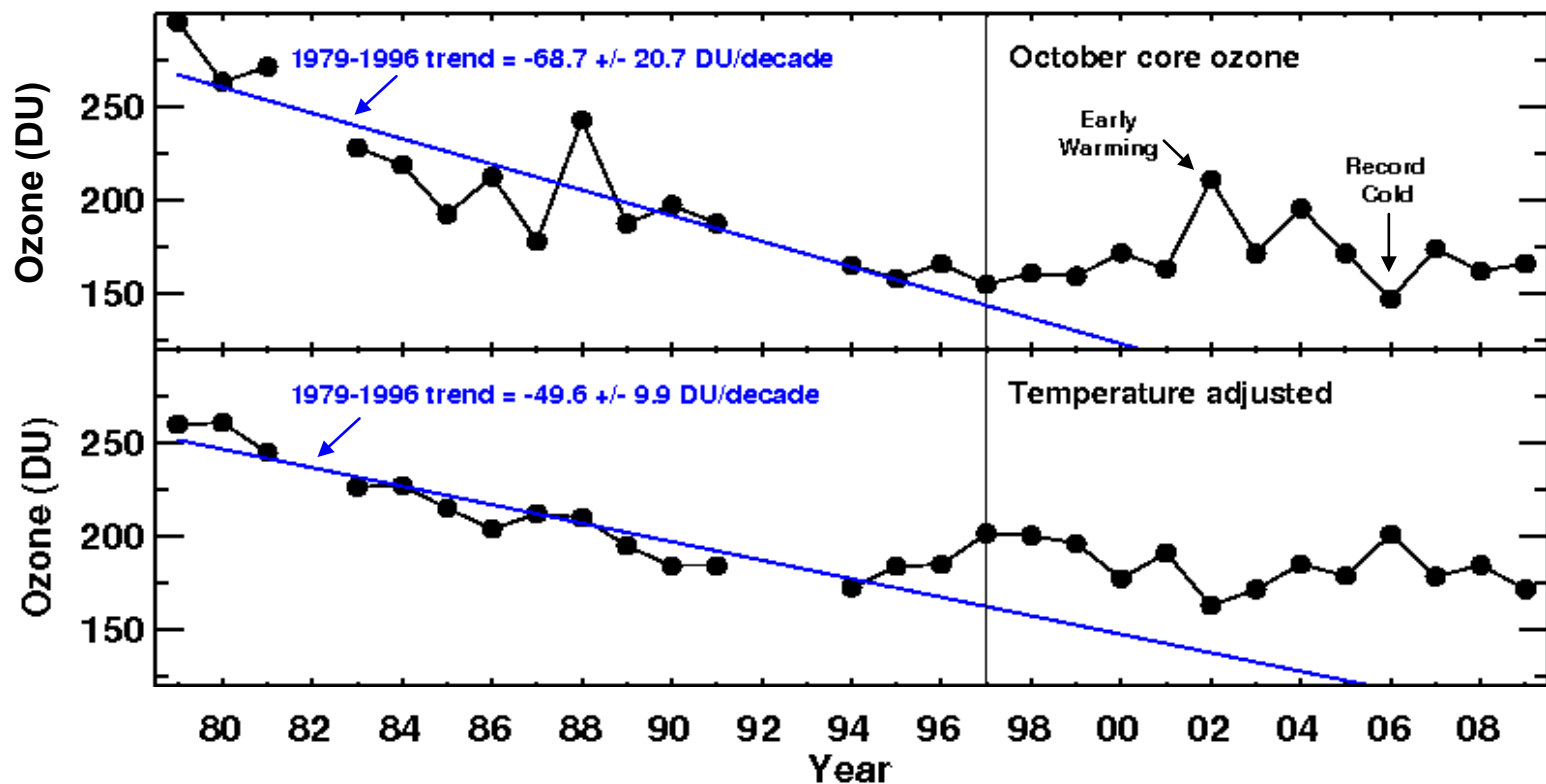
*D. Hofmann,
NOAA CMDL*

Total Ozone Over Antarctica, October



Yang *et al.*, *JGR*, 2008, updated
Figure 2-28, WMO/UNEP 2011

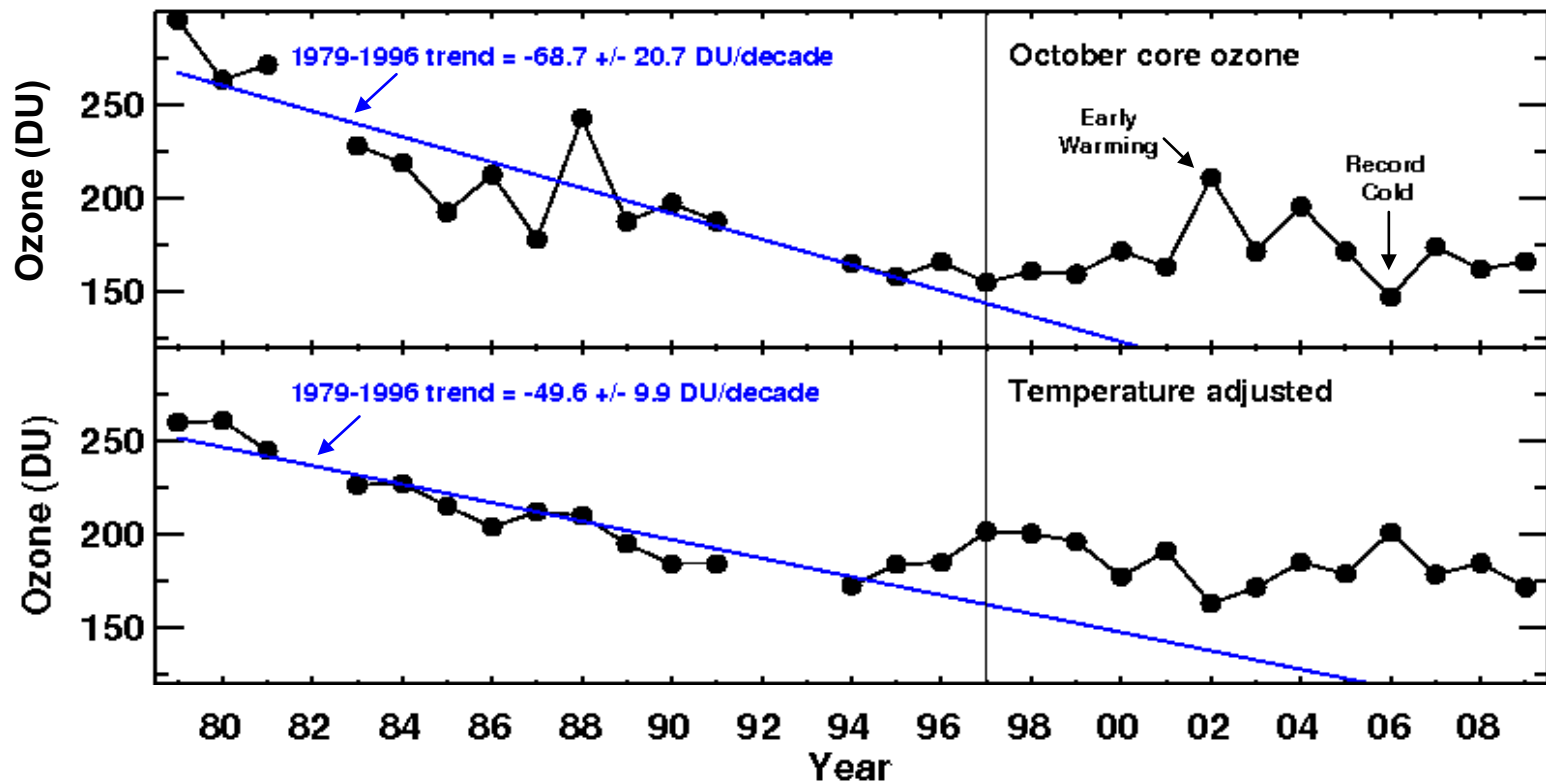
Total Ozone Over Antarctica, October



- Have dealt with **Complication #1** (Year to year variability in T)
- Now, must deal with **Complication #2** (Loss Saturation)

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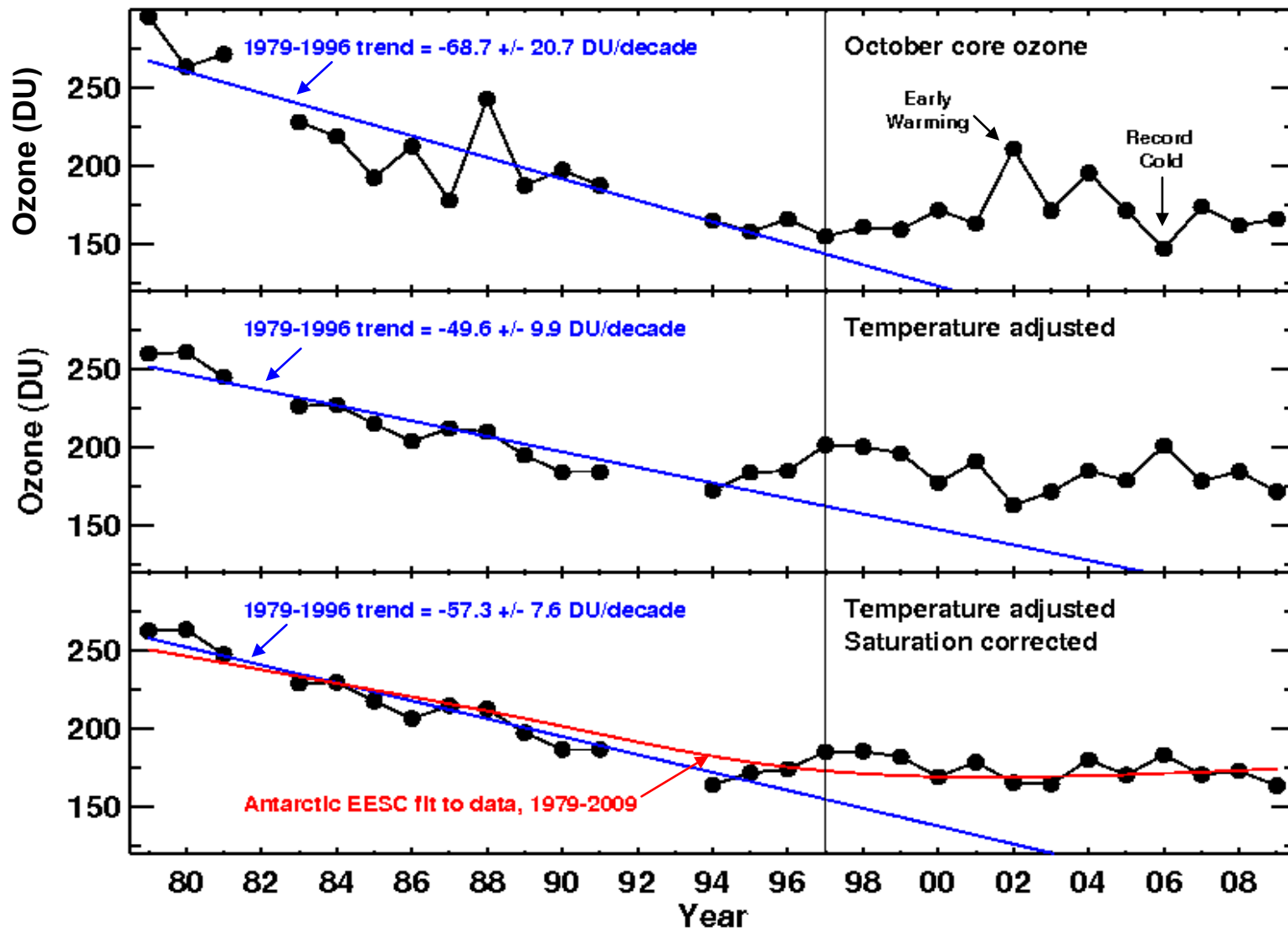
Total Ozone Over Antarctica, October



- Have dealt with Complication #1 (Year to year variability in T)
- Now, must deal with Complication #2 (Loss Saturation)
- In our computer model, we allow ozone in the heart of the ozone depletion region to “go negative”, to assess how much lower column ozone “would have been” without loss saturation

Yang *et al.*, *JGR*, 2008, updated
 Figure 2-28, WMO/UNEP 2011

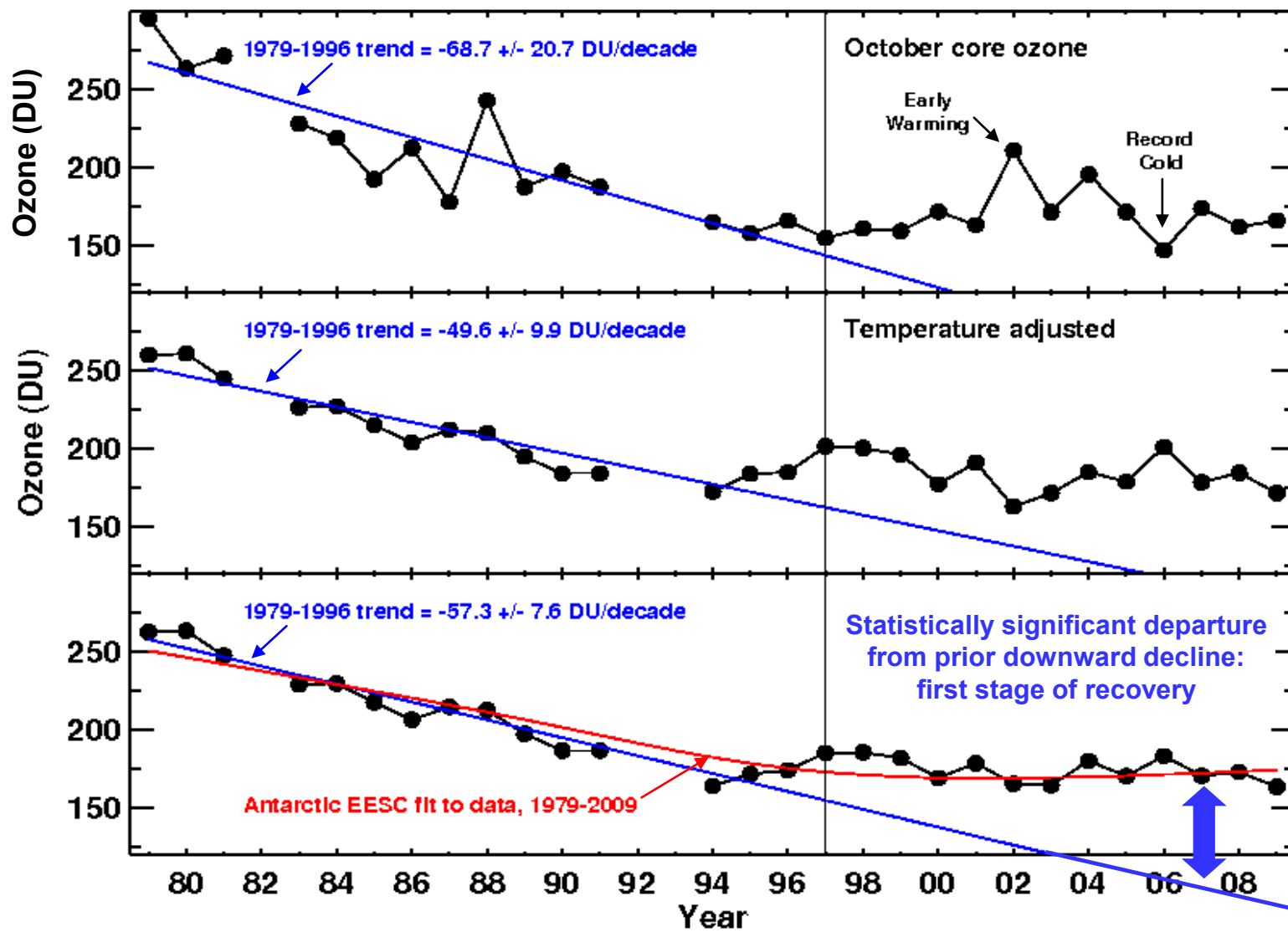
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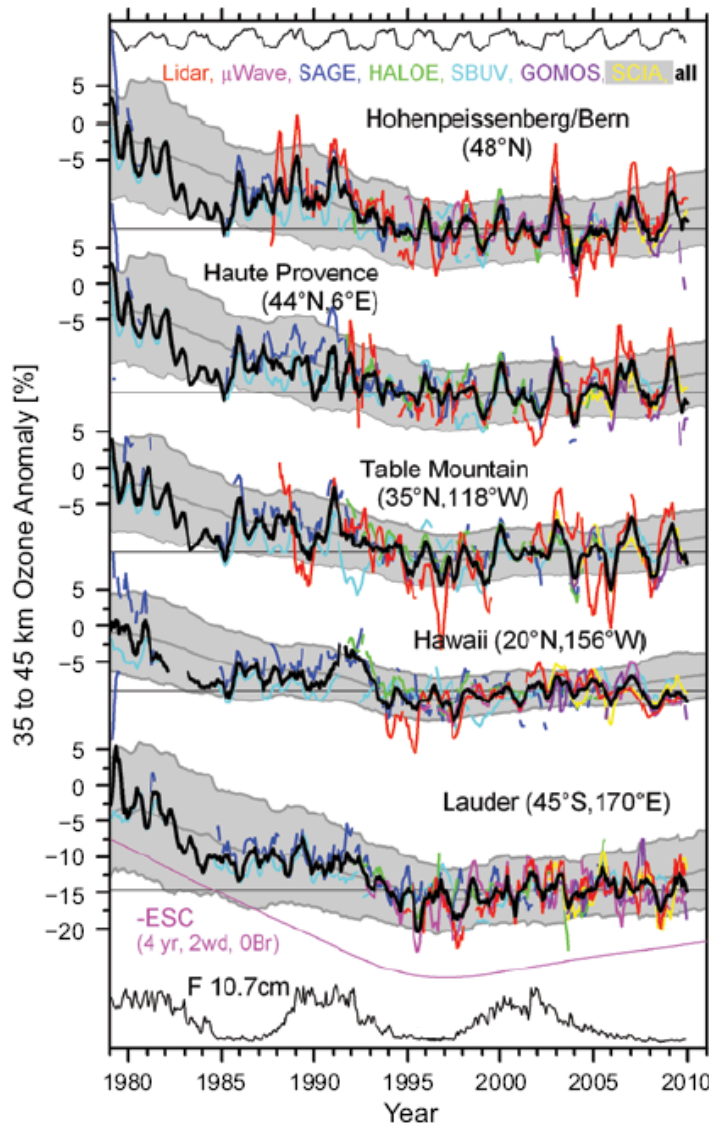
Update to Yang *et al.*, *JGR*, 2008

Figure 2-28, WMO/UNEP 2011

Yang et al. (2008) concluded:

- Antarctic Ozone is in the first stage of recovery due to the leveling off of ozone depleting substances
- In plain English: *chemical loss is not getting any worse*
(use of word “recovery” seems strange, but the community has chosen this word to describe this situation!)
- Yearly variations in Antarctic ozone now driven by meteorology
- Cold winters ⇒ low ozone

Past Trends, Upper Stratospheric Ozone



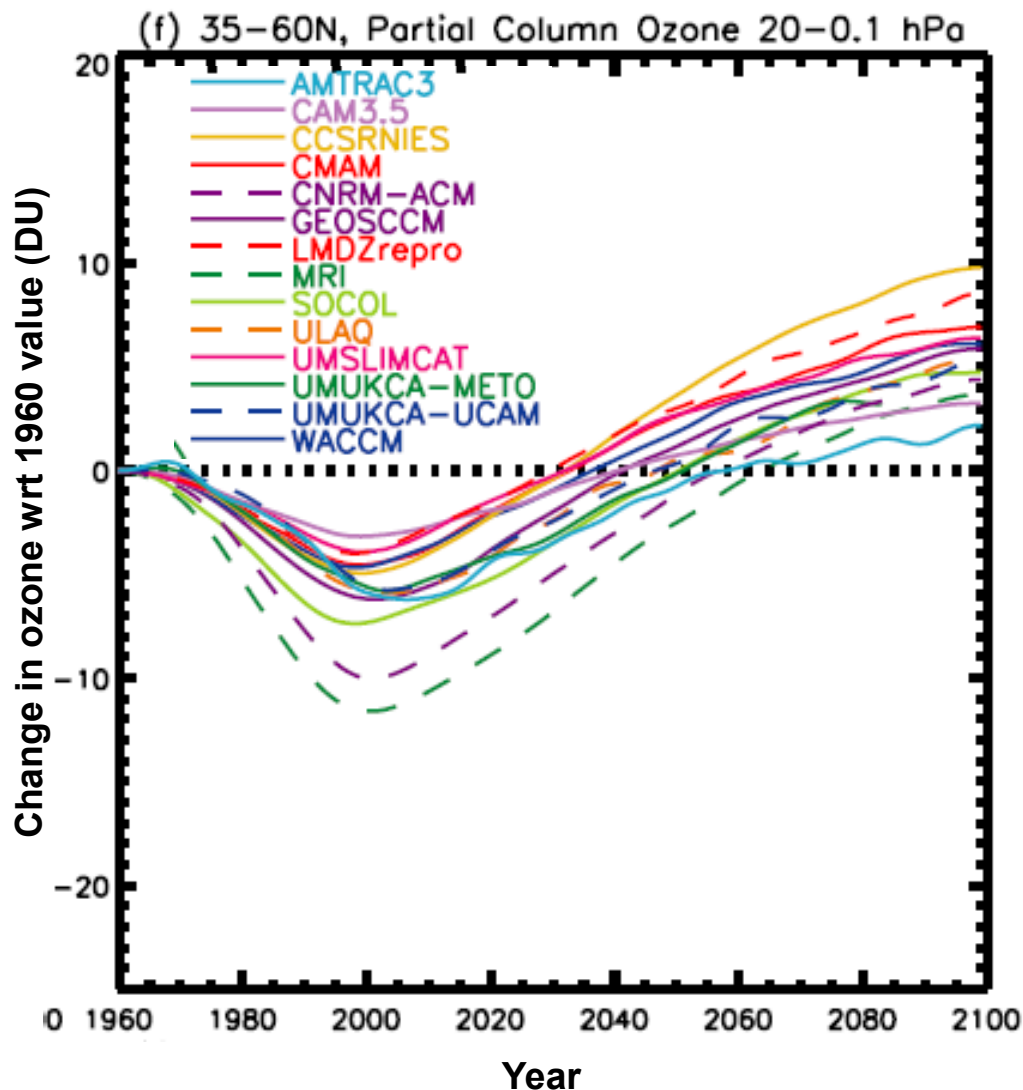
Grey: range of model calculations, where models are forced by rising levels of stratospheric halogens

Trends in ozone at 40 km are “well understood” and **generally follow track time history of stratospheric chlorine loading.**

Lecture 14, Slide 22

Figure 2-5, WMO/UNEP 2011

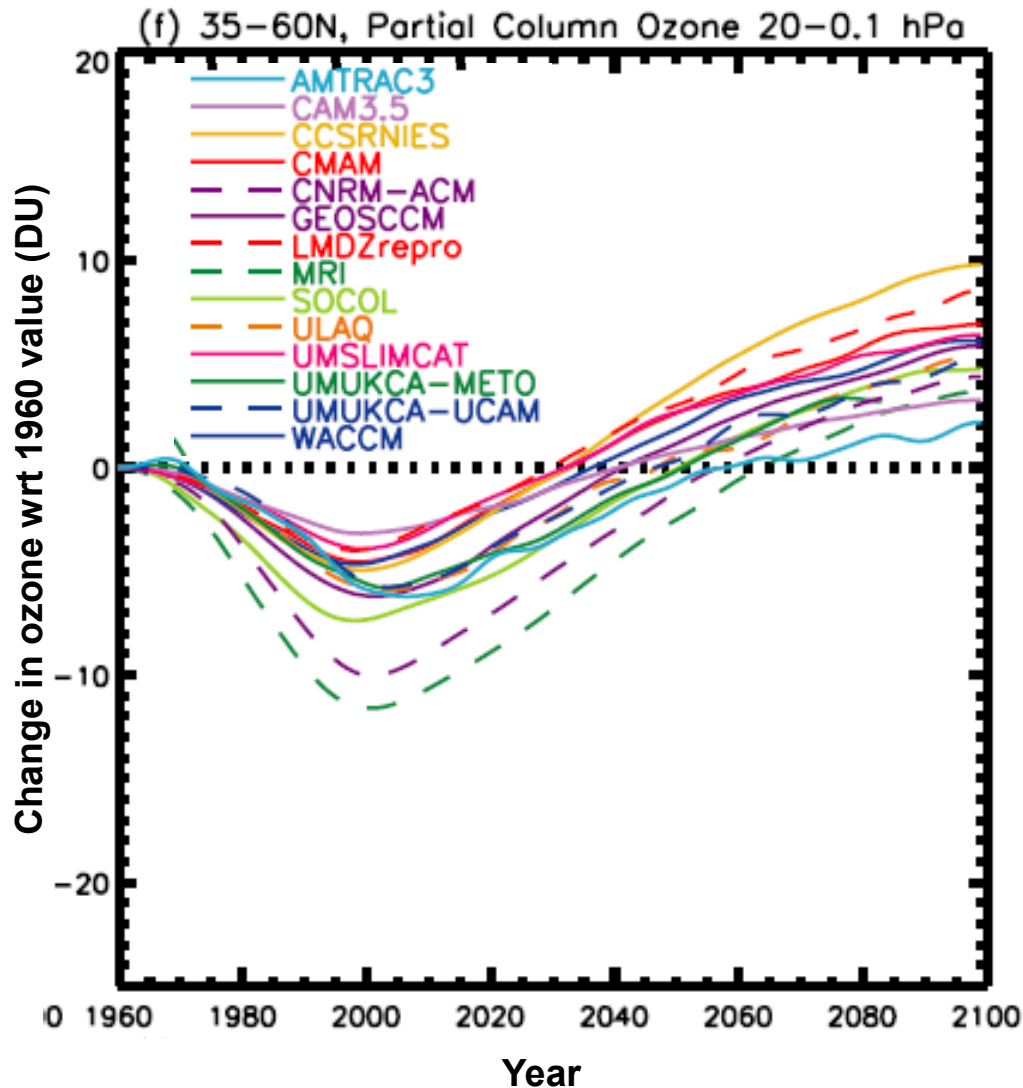
Future Trends, Upper Stratospheric Ozone



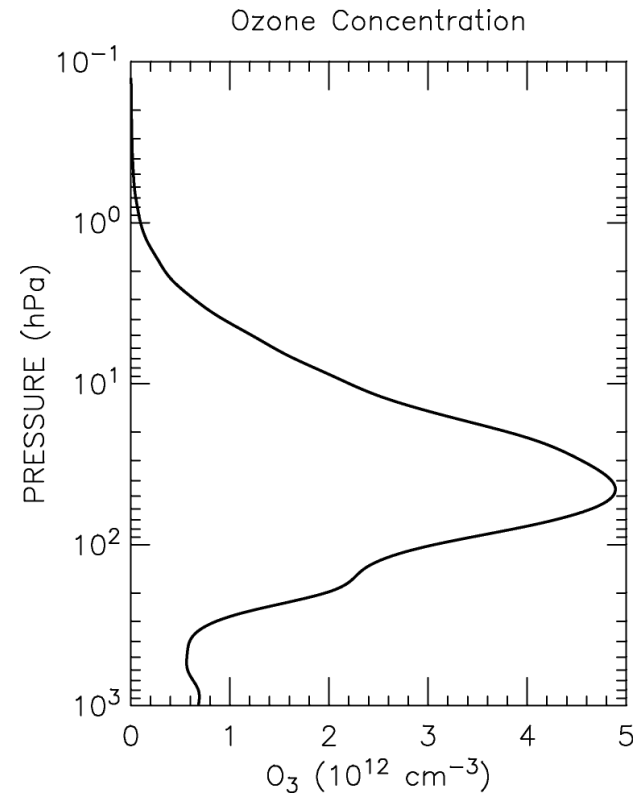
14 coupled chemistry climate models (CCMs) predict upper stratospheric ozone in 2100 will exceed upper stratospheric ozone in 1960

Oman et al., JGR, 2010

Future Trends, Upper Stratospheric Ozone

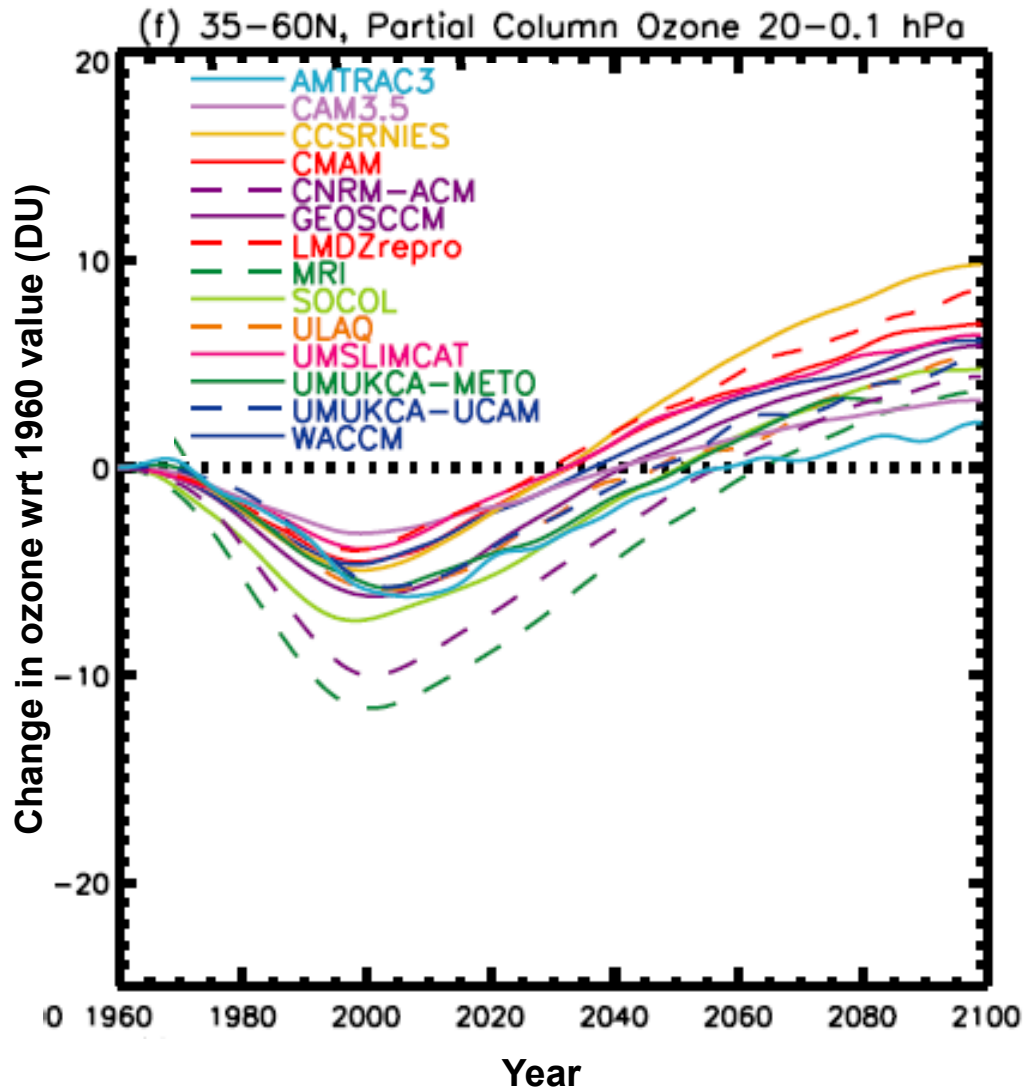


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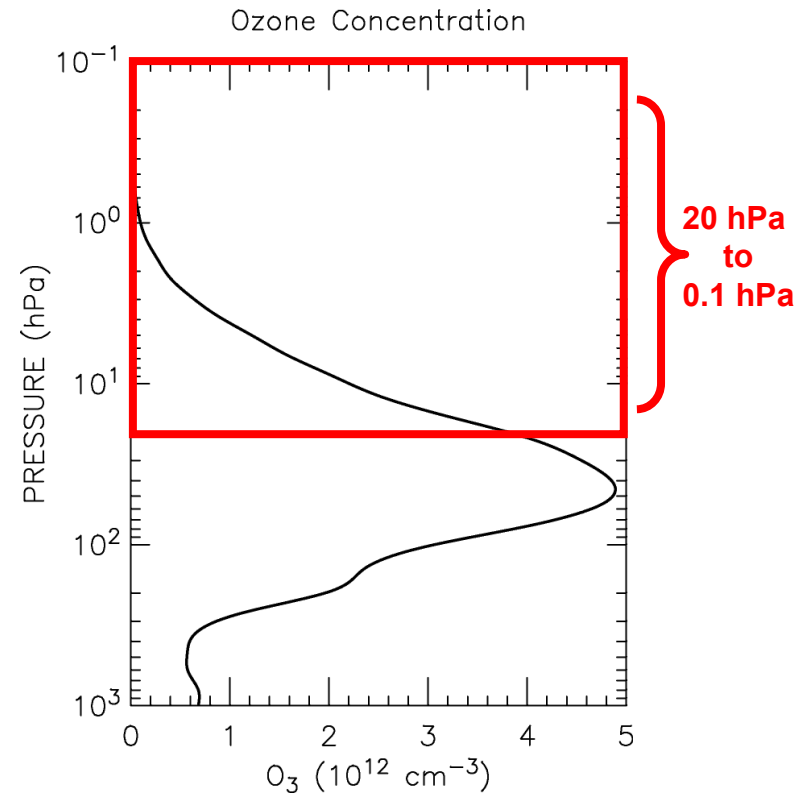


Oman *et al.*, *JGR*, 2010

Future Trends, Upper Stratospheric Ozone



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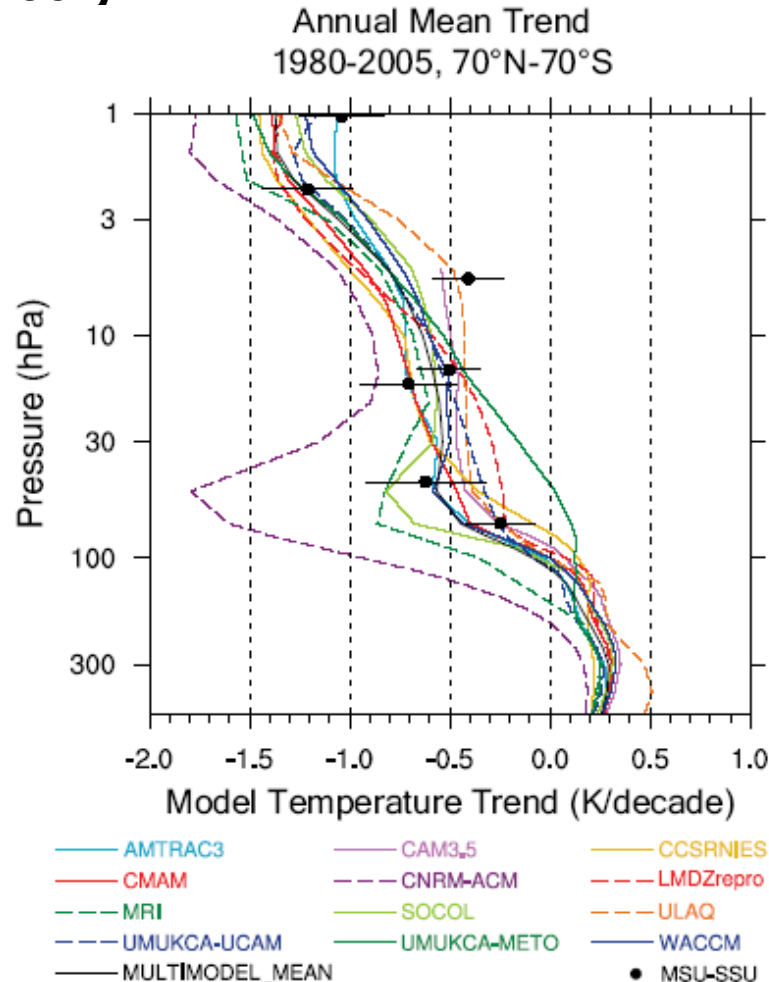


Oman *et al.*, *JGR*, 2010

Climate and Chemistry Coupling

Scientists have long known that rising GHGs leads to cooling of the stratosphere, due to direct radiative effects

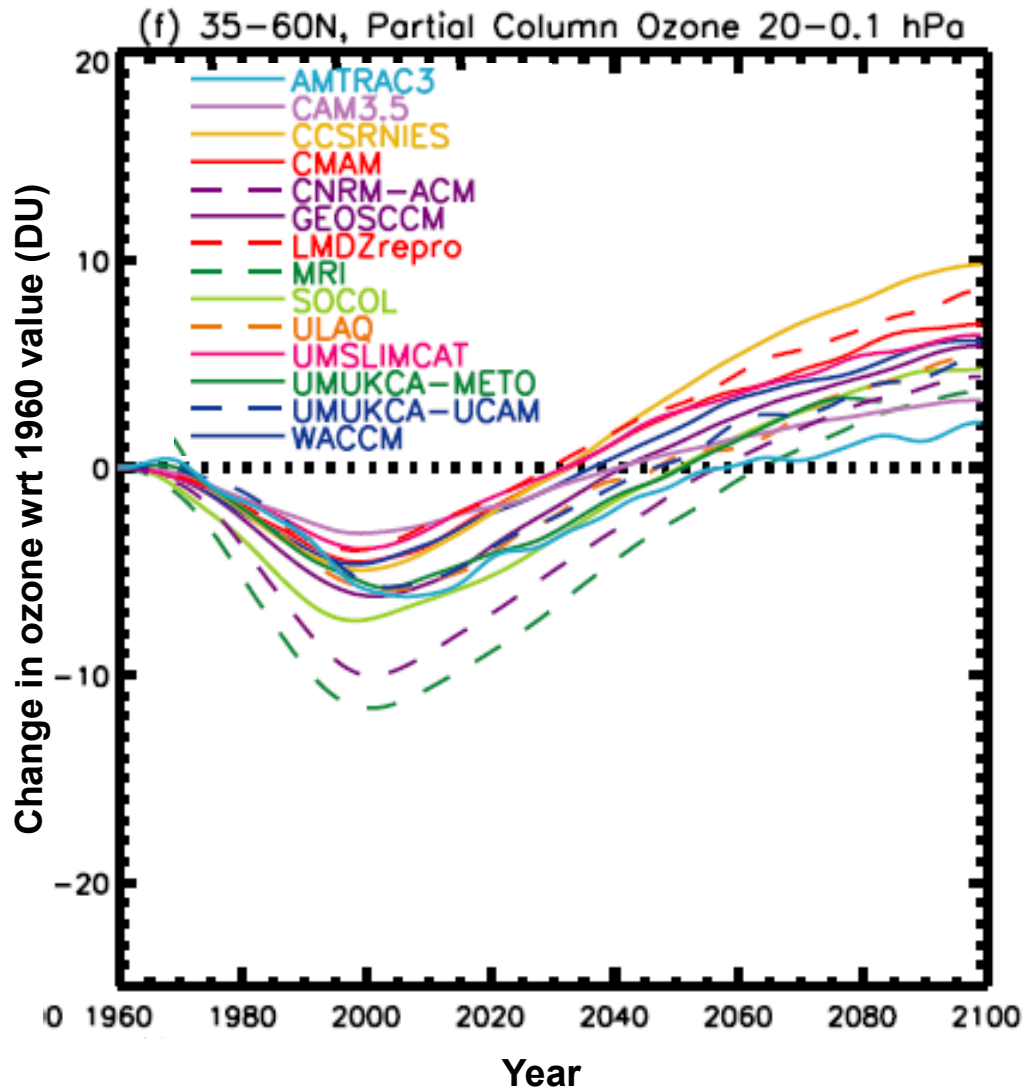
The stratosphere has been cooling past several decades in a manner broadly consistent with theory:



Lecture 15, Slide 30

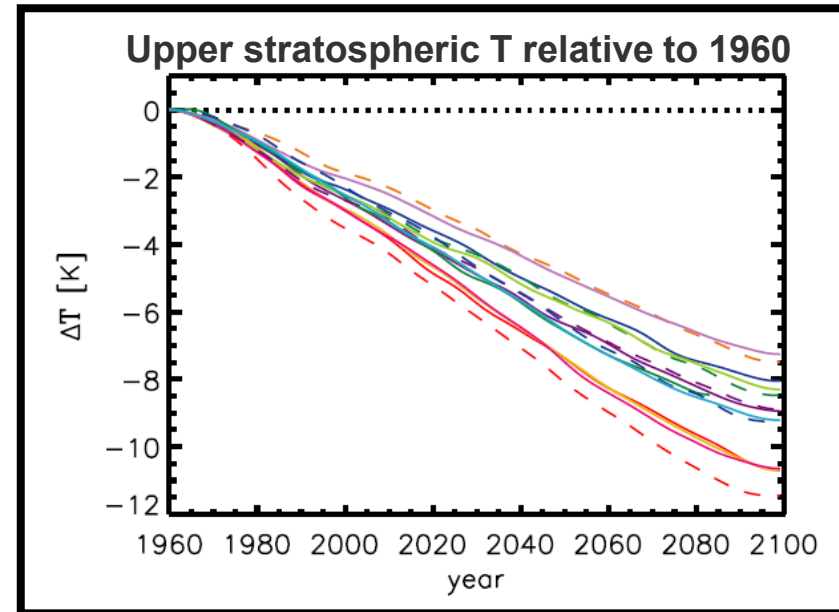
Figure 4-11, WMO/UNEP (2011)

Future Trends, Upper Stratospheric Ozone



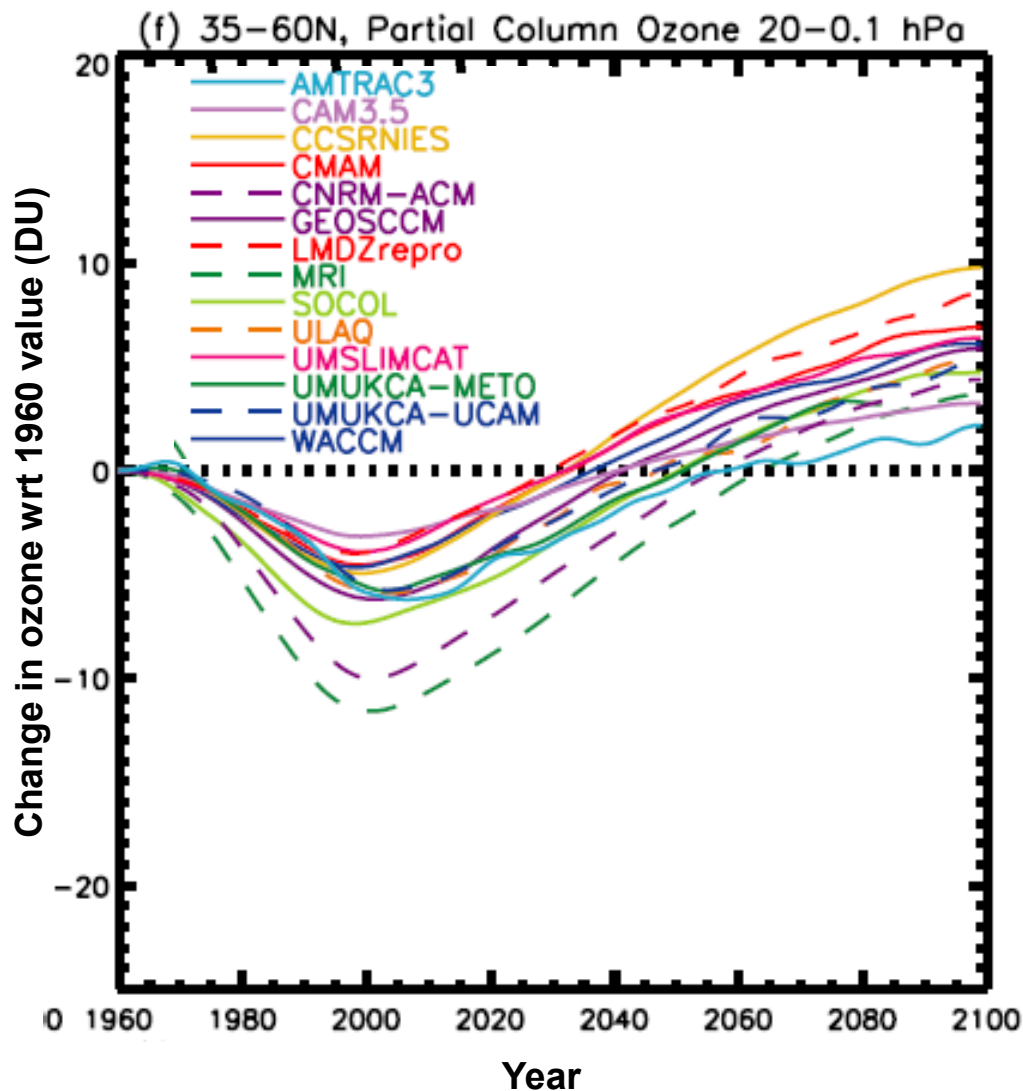
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Due to stratospheric cooling !



Oman *et al.*, *JGR*, 2010

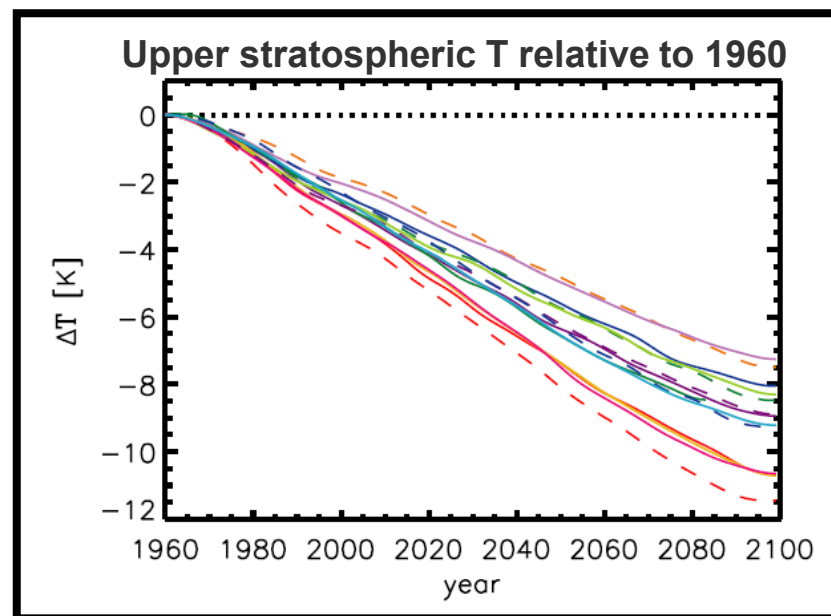
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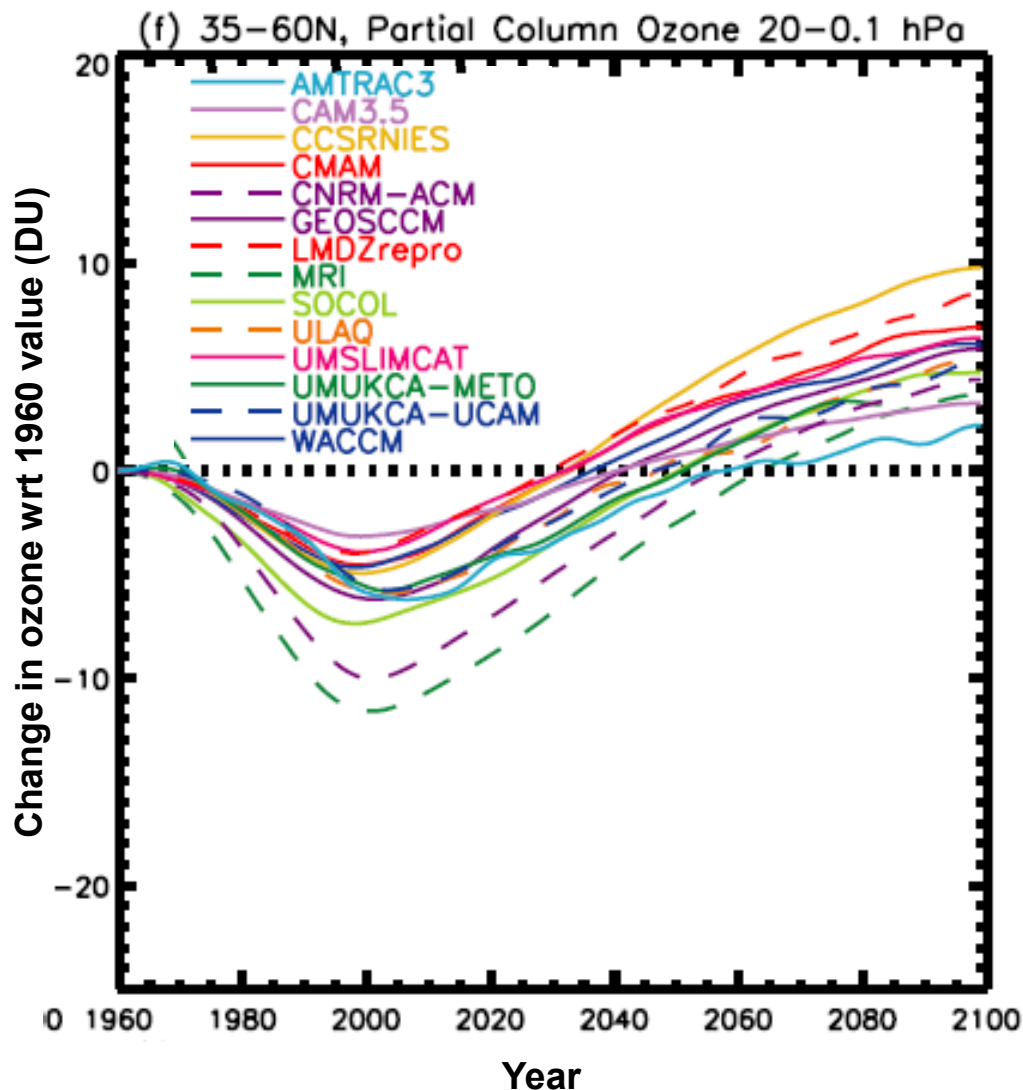
Due to stratospheric cooling !

Why this response of ozone to lower T ?



Oman *et al.*, *JGR*, 2010

Future Trends, Upper Stratospheric Ozone



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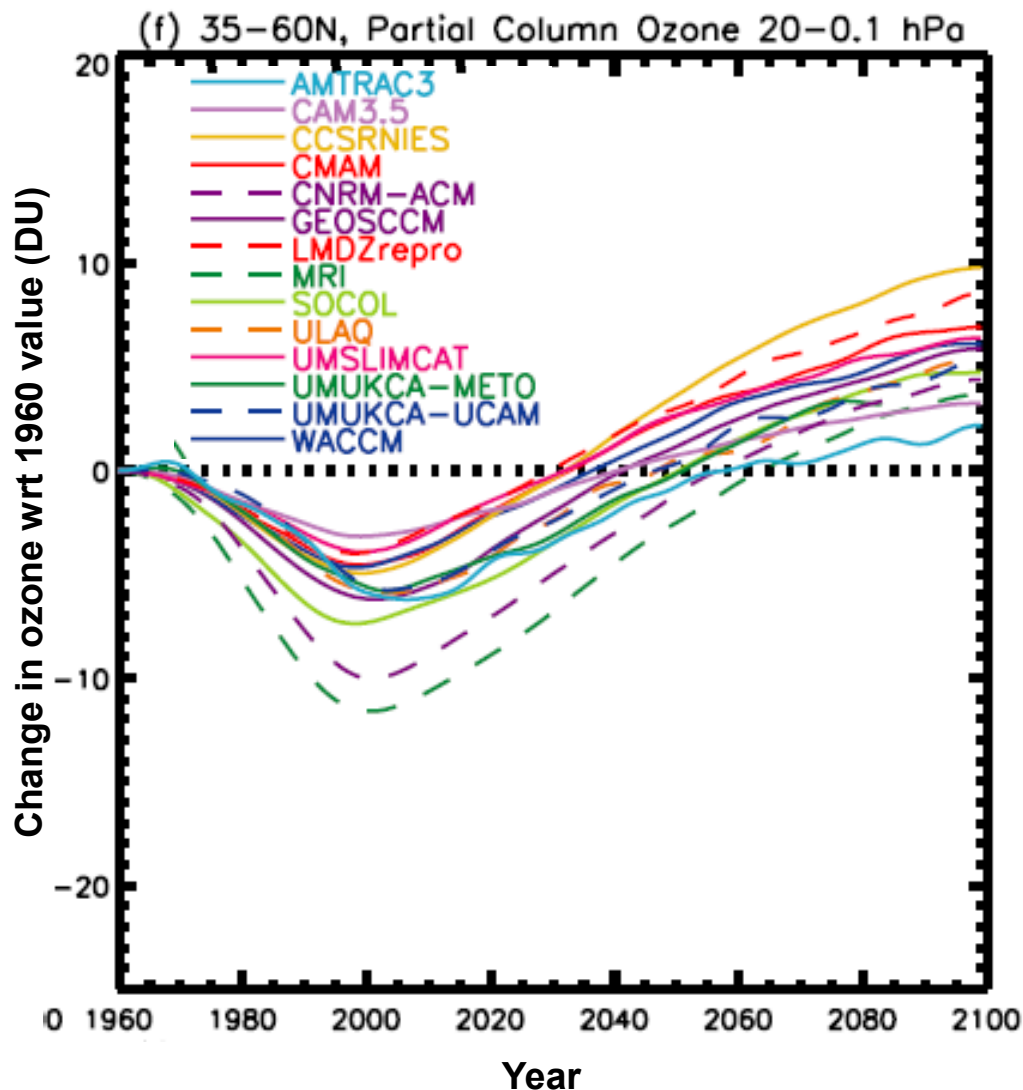
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Gas phase rate constants are sensitive to temperature

Oman *et al.*, *JGR*, 2010

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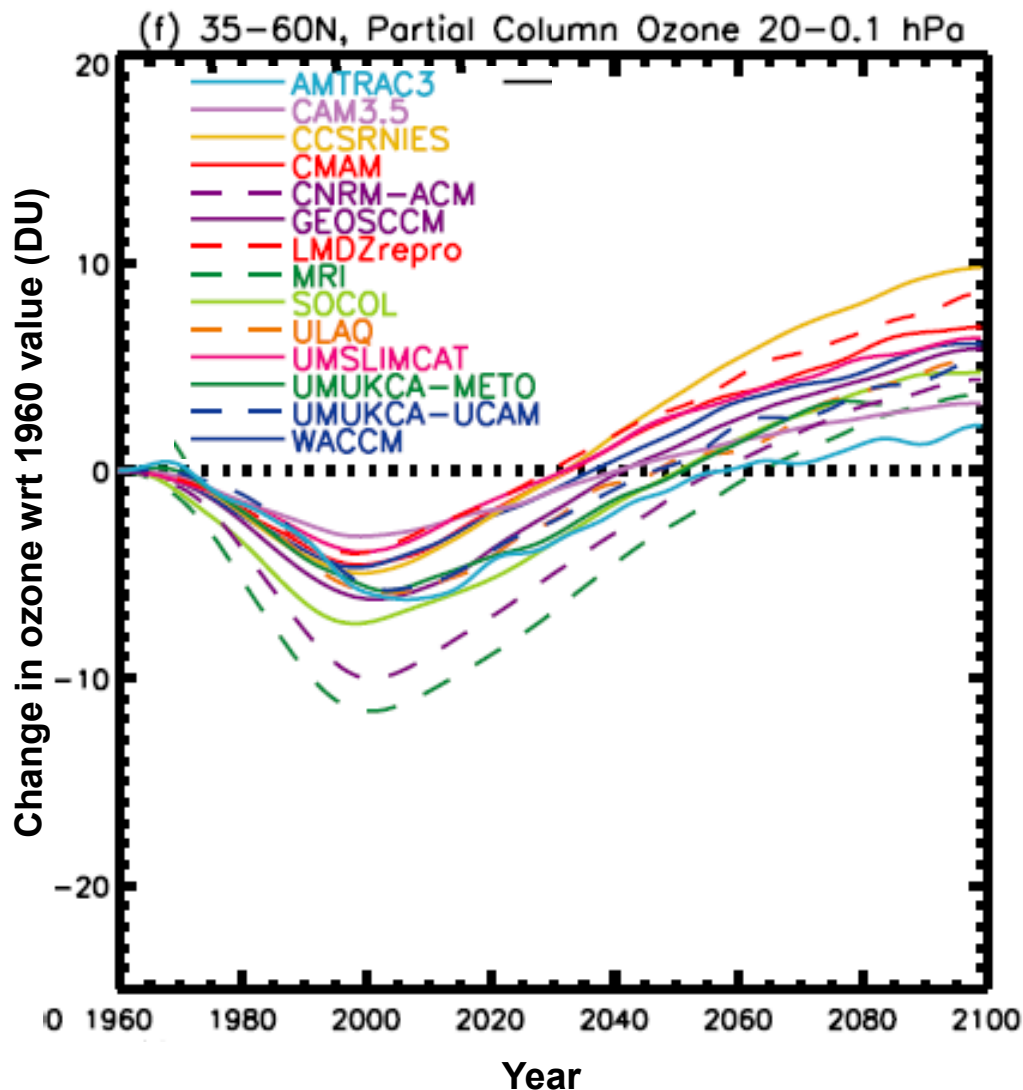
Why this response to lower T ?

Gas phase rate constants are sensitive to temperature

In particular, $O + O_2 + M \rightarrow O_3 + M$ speeds up as T drops

Oman *et al.*, *JGR*, 2010

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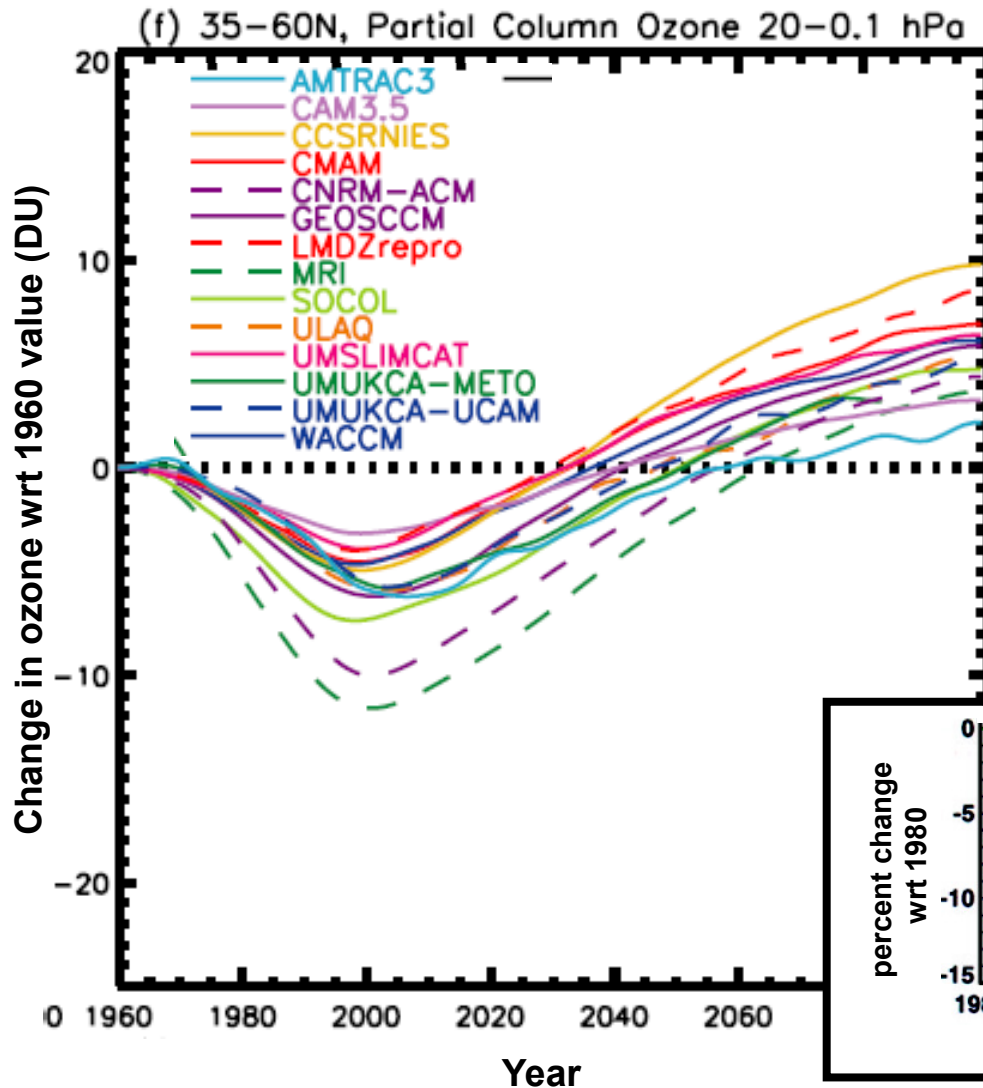
Gas phase rate constants are sensitive to temperature

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How will this affect partitioning of O and O₃ ?

Oman *et al.*, *JGR*, 2010

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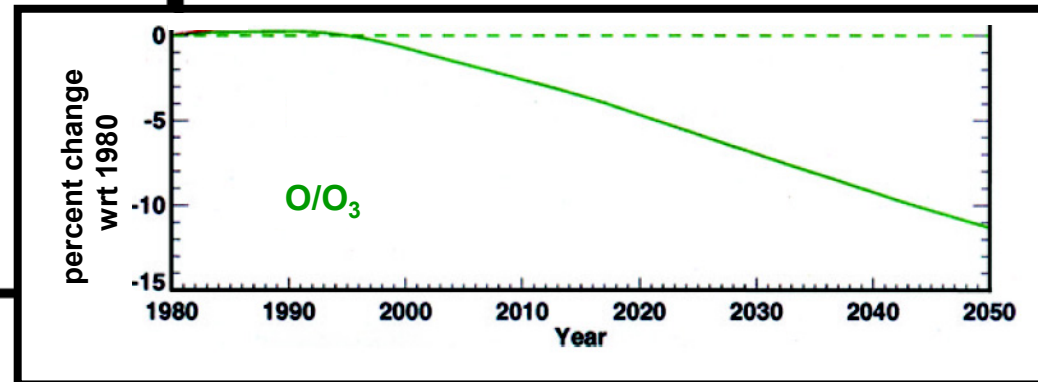
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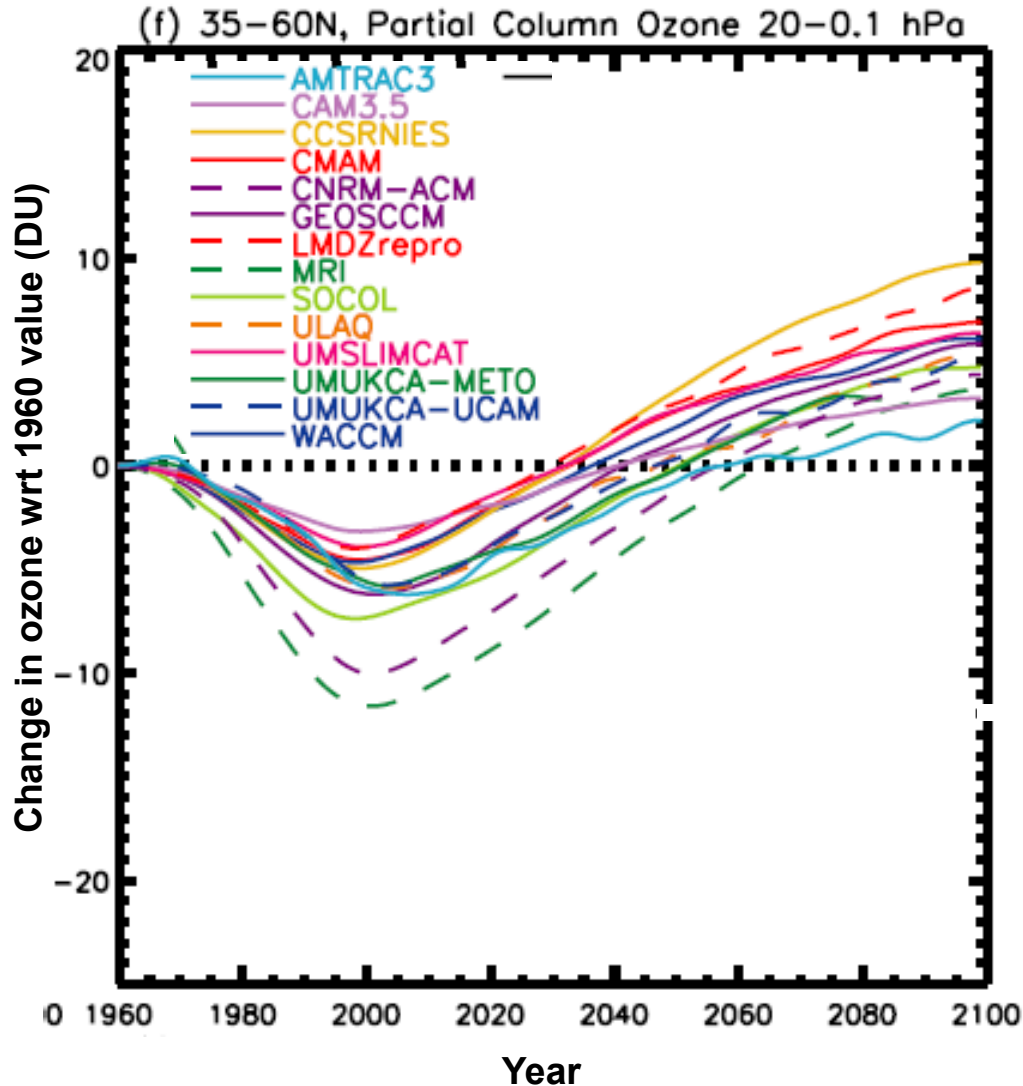
How will this affect partitioning of O and O_3 ?



Oman *et al.*, *JGR*, 2010

Rosenfield *et al.*, *JGR*, 2002

Future Trends, Upper Stratospheric Ozone



Oman et al., JGR, 2010

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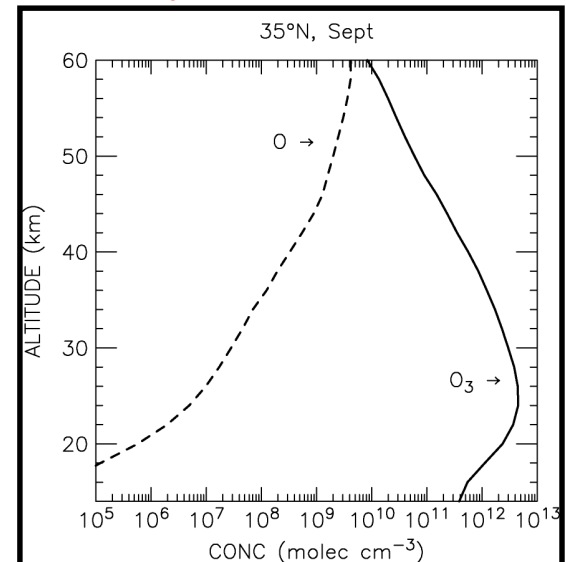
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More Chemistry and Climate Coupling

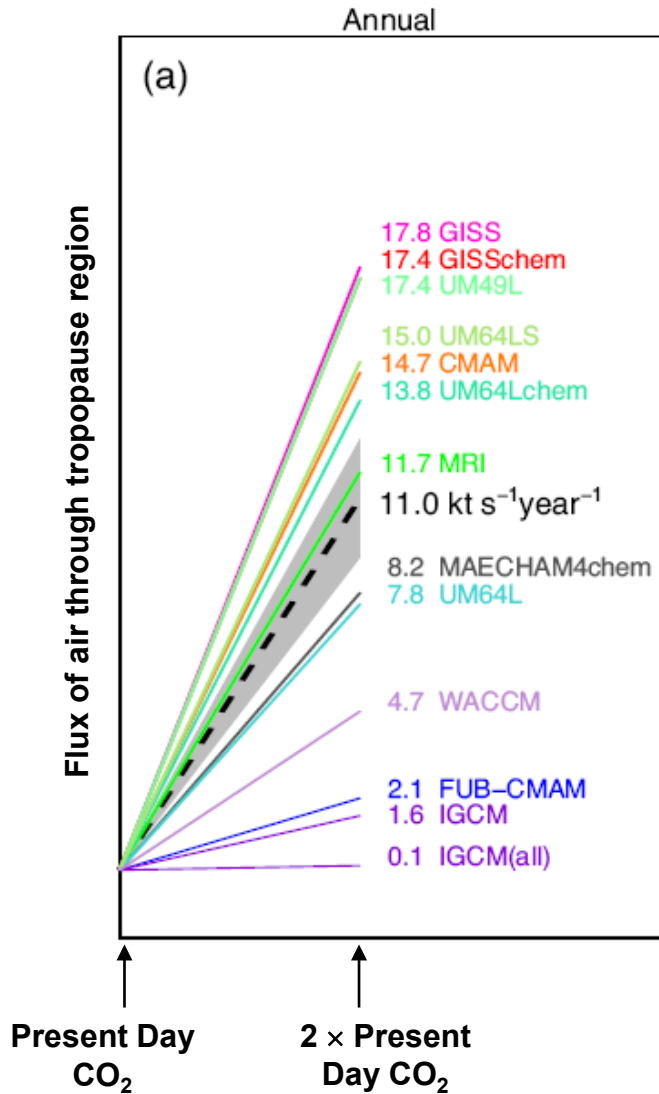


Figure 5-17. Trends in exchange of air from troposphere-to-stratosphere computed by 14 CCMs.

Trends (units of Gg s⁻¹ year⁻¹) are represented by the slope of each line.

Dashed line is the multi-model mean.

After Butchart *et al.*, *Clim. Dyn.*, 2006.

WMO/UNEP Ozone Assessment Report 2007

Brewer-Dobson Circulation

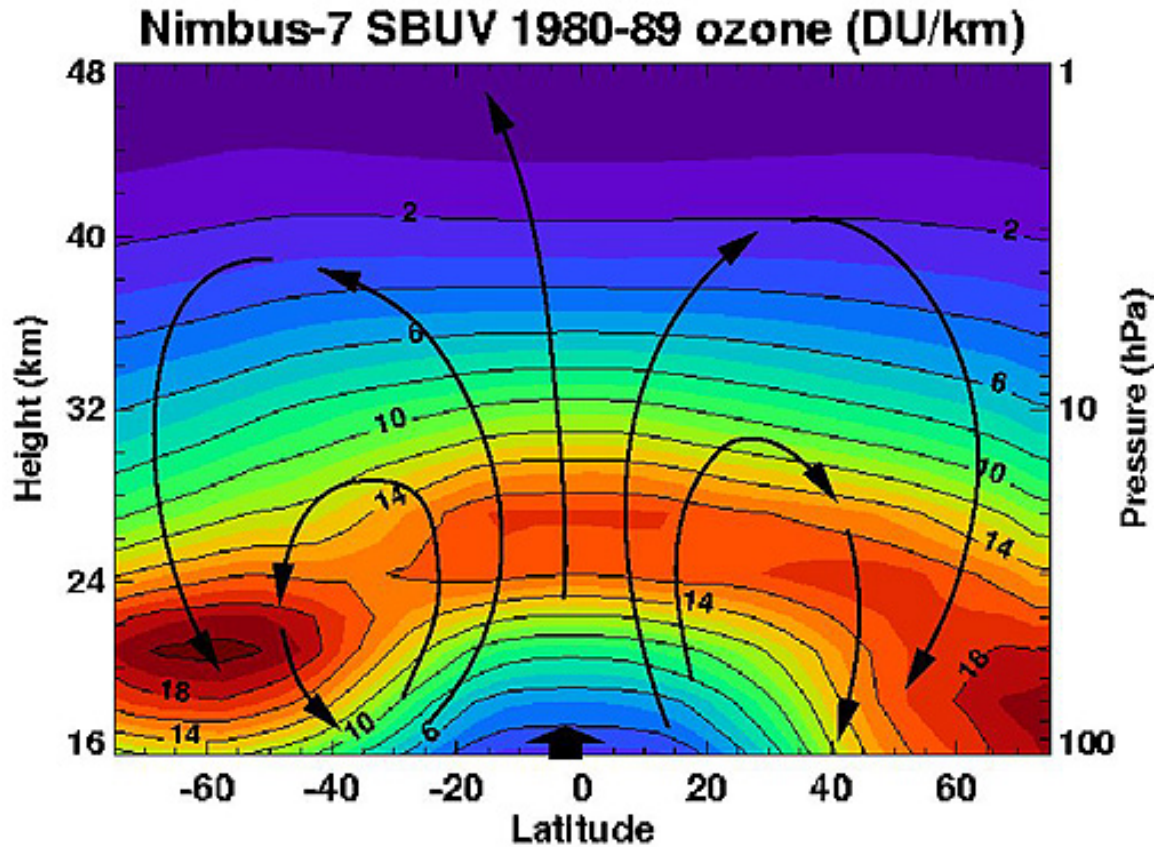


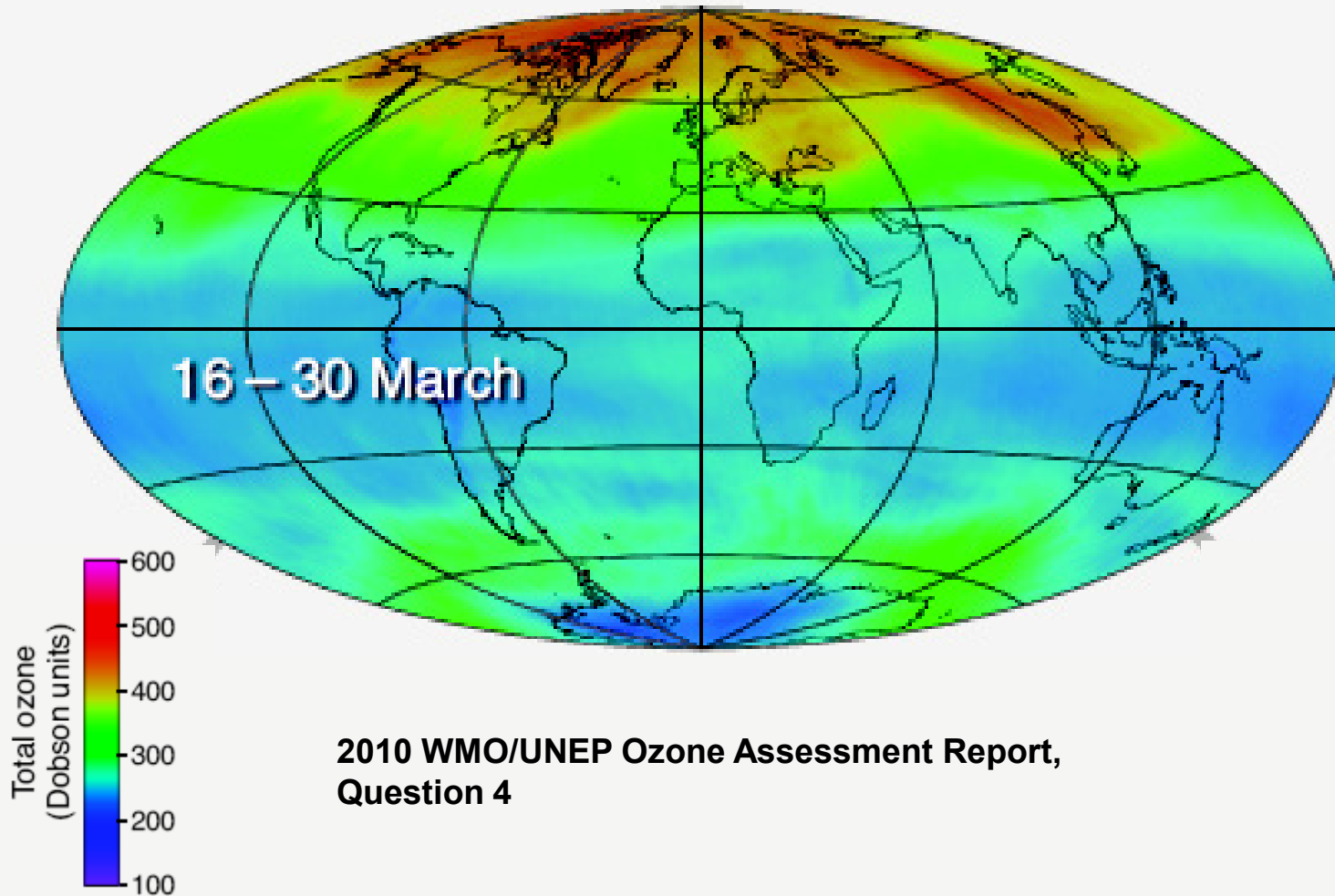
Figure 6.03 Schematic diagram of Brewer-Dobson circulation with seasonally averaged ozone concentration

http://www.ccpo.odu.edu/~lizsmith/SEES/ozone/class/Chap_1/1_Js/1-06.jpg

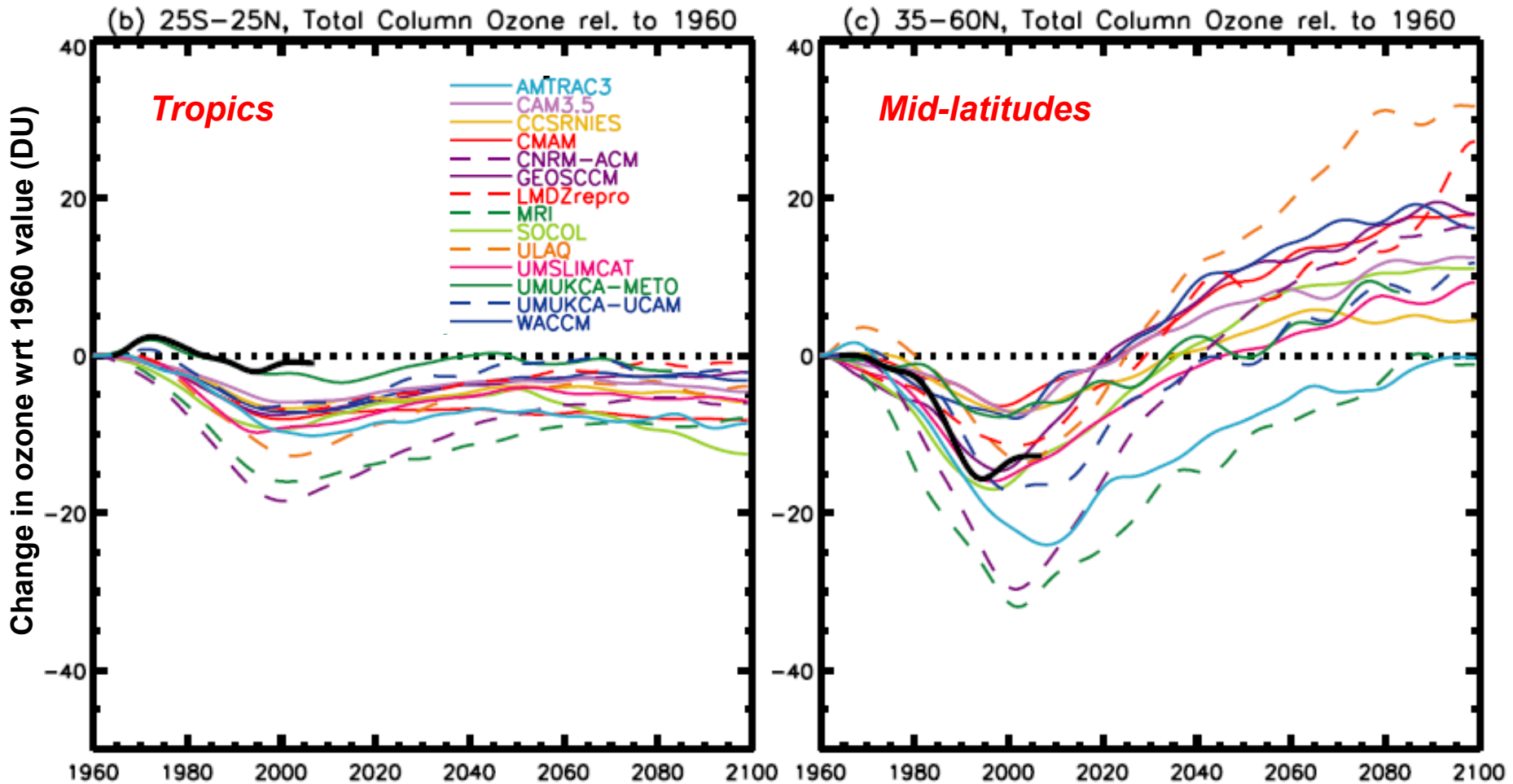
Brewer-Dobson Circulation is a model of atmospheric circulation, proposed by Alan Brewer in 1949 and Gordon Dobson in 1956, that attempts to explain why tropical air has less column ozone than polar air, even though the tropical stratosphere is where most atmospheric ozone is produced

Global Satellite Maps of Total Ozone in 2009

Early spring



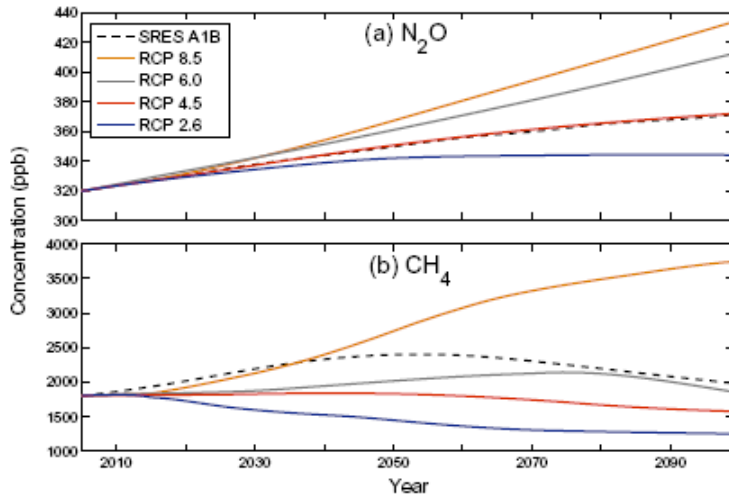
More Chemistry and Climate Coupling



Acceleration of the Brewer-Dobson Circulation causes modeled total ozone column in the tropics to exhibit a sustained, long term decline and modeled total ozone column in the NH to experience a “super recovery”

Oman et al., JGR, 2010

Future Mid-Latitude Ozone: CH₄



Rising CH₄ leads to ozone loss in the upper & lower stratos. by increasing the speed of HO_x mediated loss cycles (blue regions, Fig 6b).

However, there are other processes that result in more ozone (red regions, Fig 6b):

- Rising CH₄ leads to more stratospheric H₂O, cooling this region of the atmosphere, which slows the rate of all ozone loss cycles
- Rising CH₄ speeds up the rate of Cl+CH₄, shifting chlorine from ClO into HCl
- Rising CH₄ leads to more HO₂ in the lowermost stratosphere, where there is sufficient CO to result in production of O₃ by photochemical smog chemistry

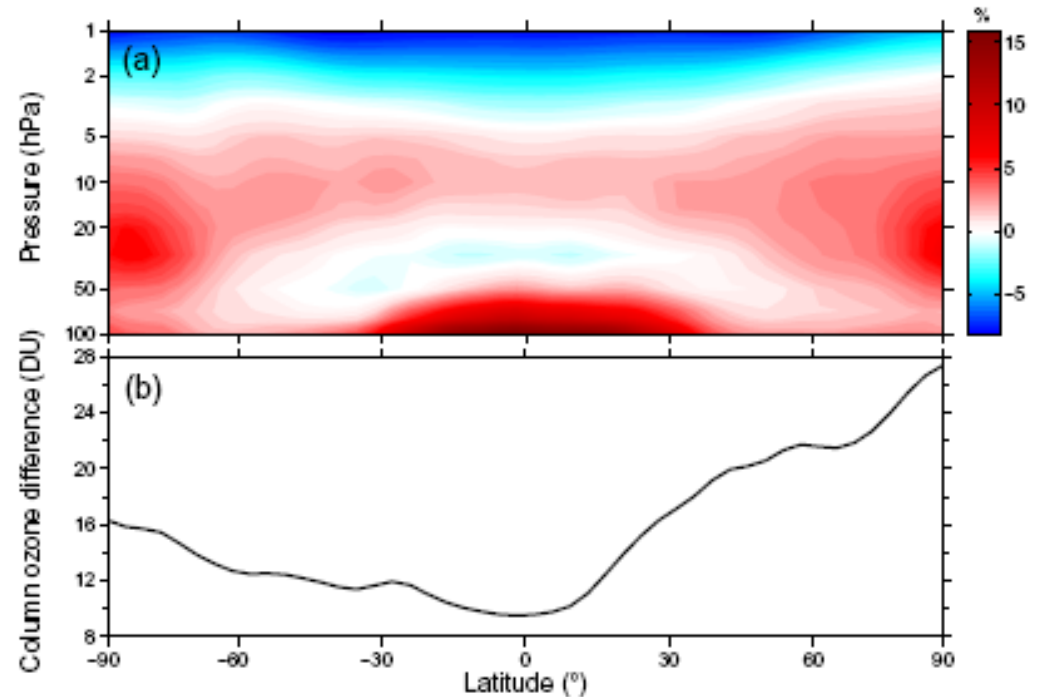
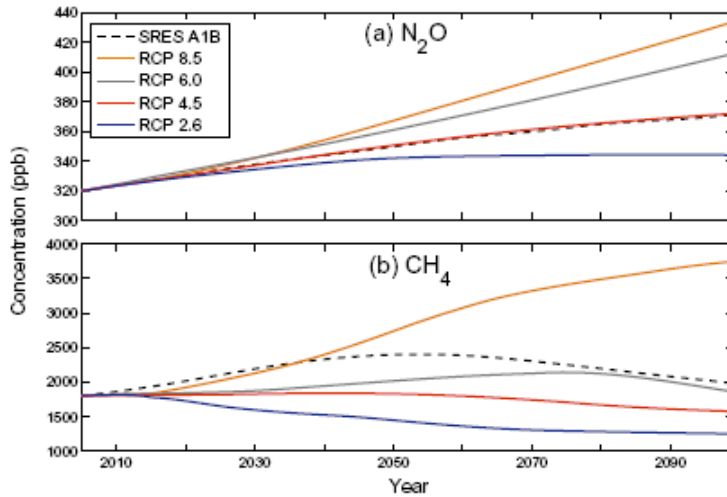


Fig. 6. (a) CH₄-8.5 ozone minus CH₄-2.6 ozone in the 2090s decade, calculated as a percentage of ozone in the CH₄-2.6 simulation. (b) 2090s-decade CH₄-8.5 total column ozone minus CH₄-2.6 total column ozone.

Revell *et al.*, *ACP*, 2012

Future Mid-Latitude Ozone: N₂O



Ozone depleting NO_x cycles speed up with increasing N₂O throughout the middle stratosphere, where these cycles make the largest relative contribution to odd oxygen loss (blue region, Fig 5a).

- As NO₂ increases due to rising N₂O, the abundance of ClO declines, particularly in the lower stratosphere, leading to reduced rates in the total speed of all ozone depleting cycles (red region, Fig 5a); small contrib. to the red region due to production of O₃ by photochemical smog chemistry.

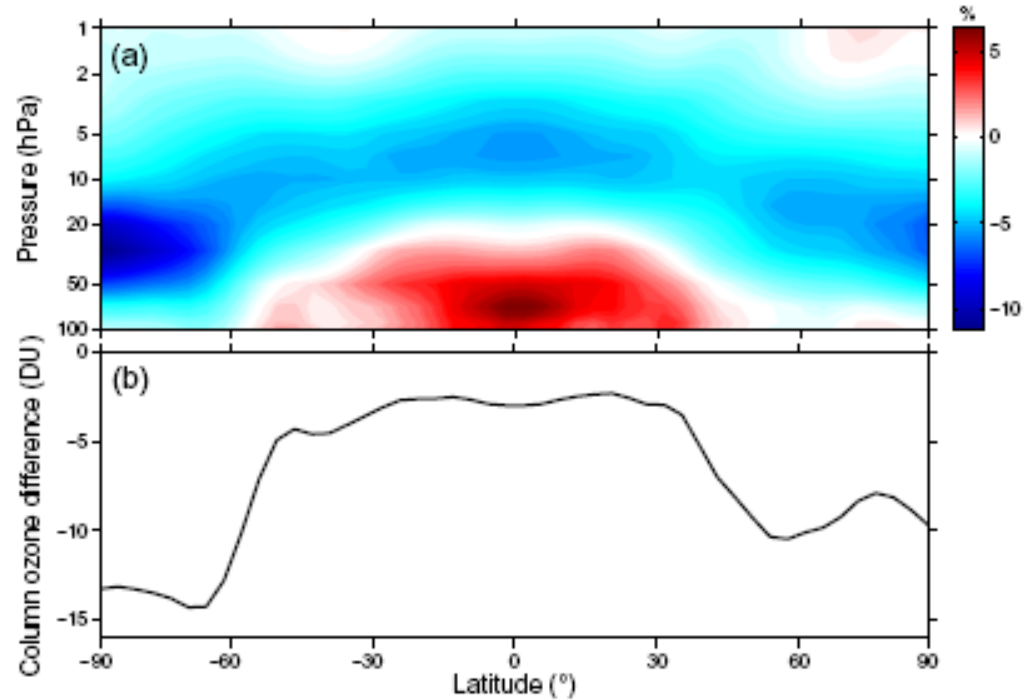
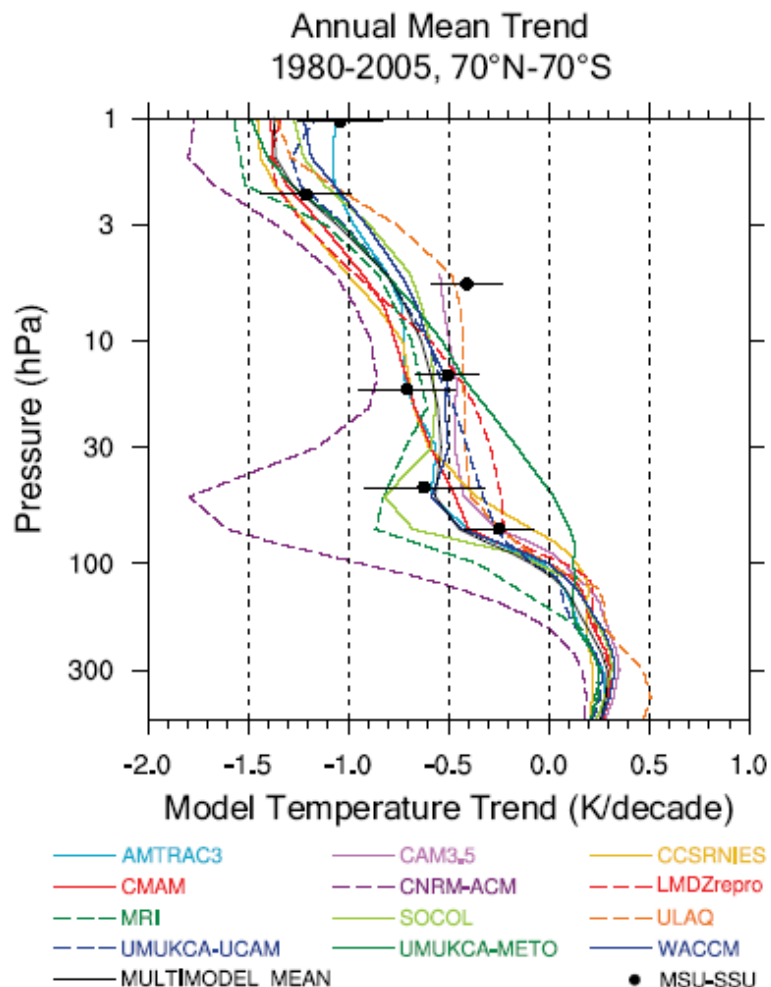


Fig. 5. (a) N₂O-8.5 ozone minus N₂O-2.6 ozone in the 2090s decade, calculated as a percentage of ozone in the N₂O-2.6 simulation. (b) 2090s-decade N₂O-8.5 total column ozone minus N₂O-2.6 total column ozone.

Revell *et al.*, *ACP*, 2012

Future Trends, Stratospheric Ozone

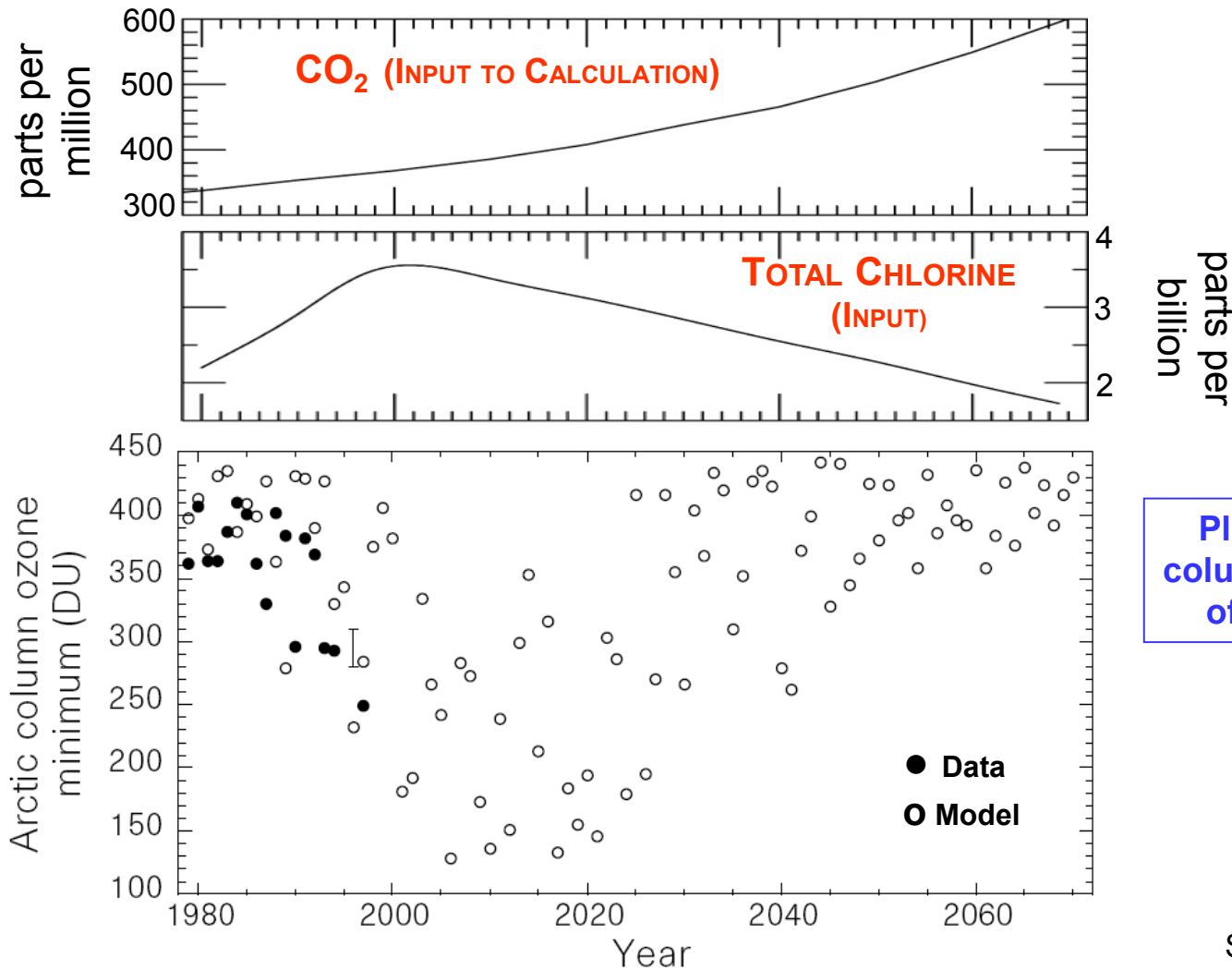


If the stratosphere continues to cool,
for which region of the stratosphere
is ozone “most vulnerable”?

Figure 4-11, WMO/UNEP (2011)

Lecture 15, Slide 30

Arctic Ozone Loss - Climate and Chemistry Coupling

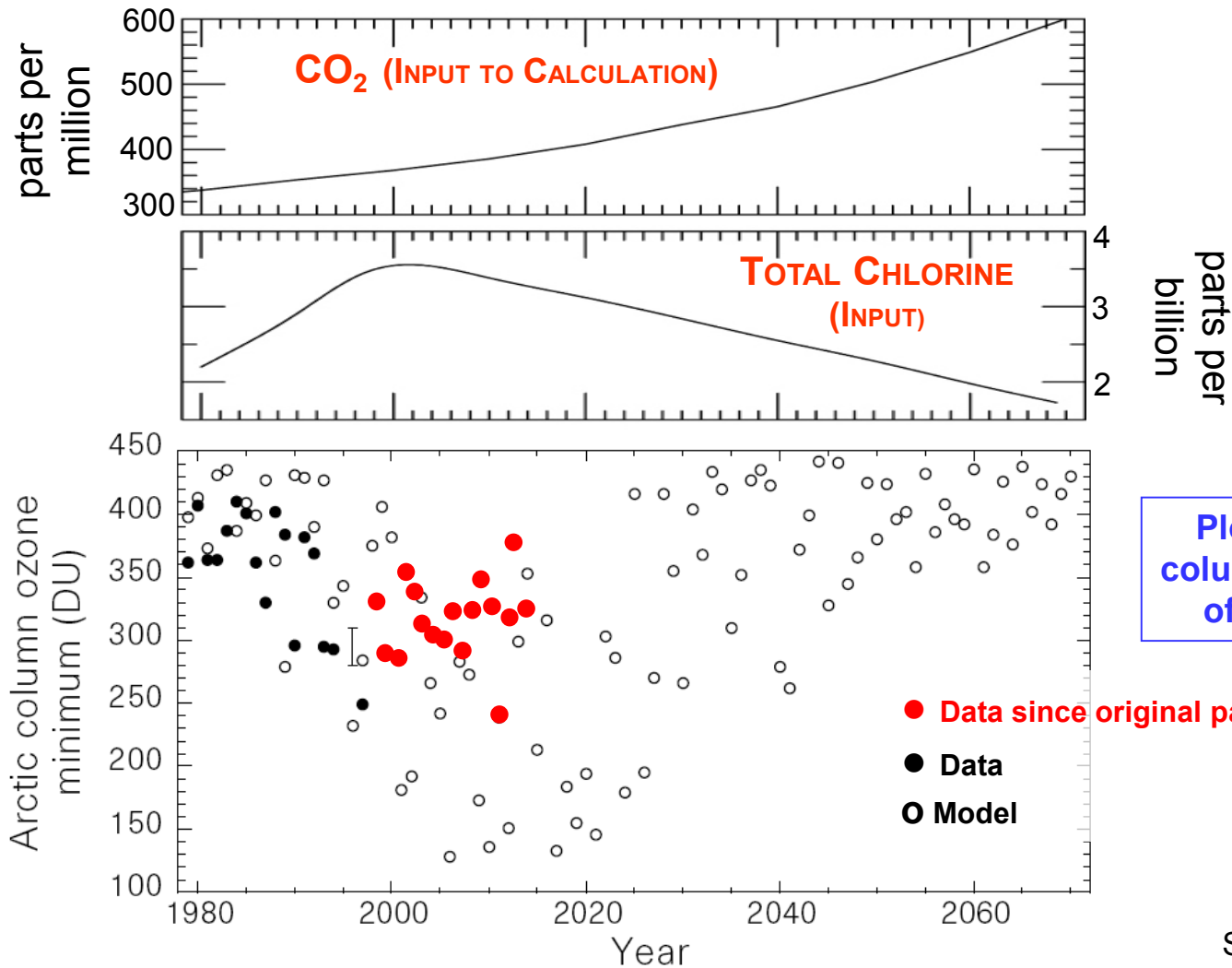


Shindell et al., *Nature*, 1998

Shindell et al. (1998) postulated that rising GHGs would lead to more stable polar vortex circulations, resulting in maximum loss of Arctic ozone in the decade 2010 to 2019.

Driving factor is a decrease in the poleward propagation of planetary waves i.e., dynamics

Arctic Ozone Loss - Climate and Chemistry Coupling



Shindell et al., *Nature*, 1998

Arctic column ozone has not reached the very deep minima predicted by Shindell et al., even in 2011

Arctic Ozone Loss - Climate and Chemistry Coupling

March O₃ Column 60°N–90°N

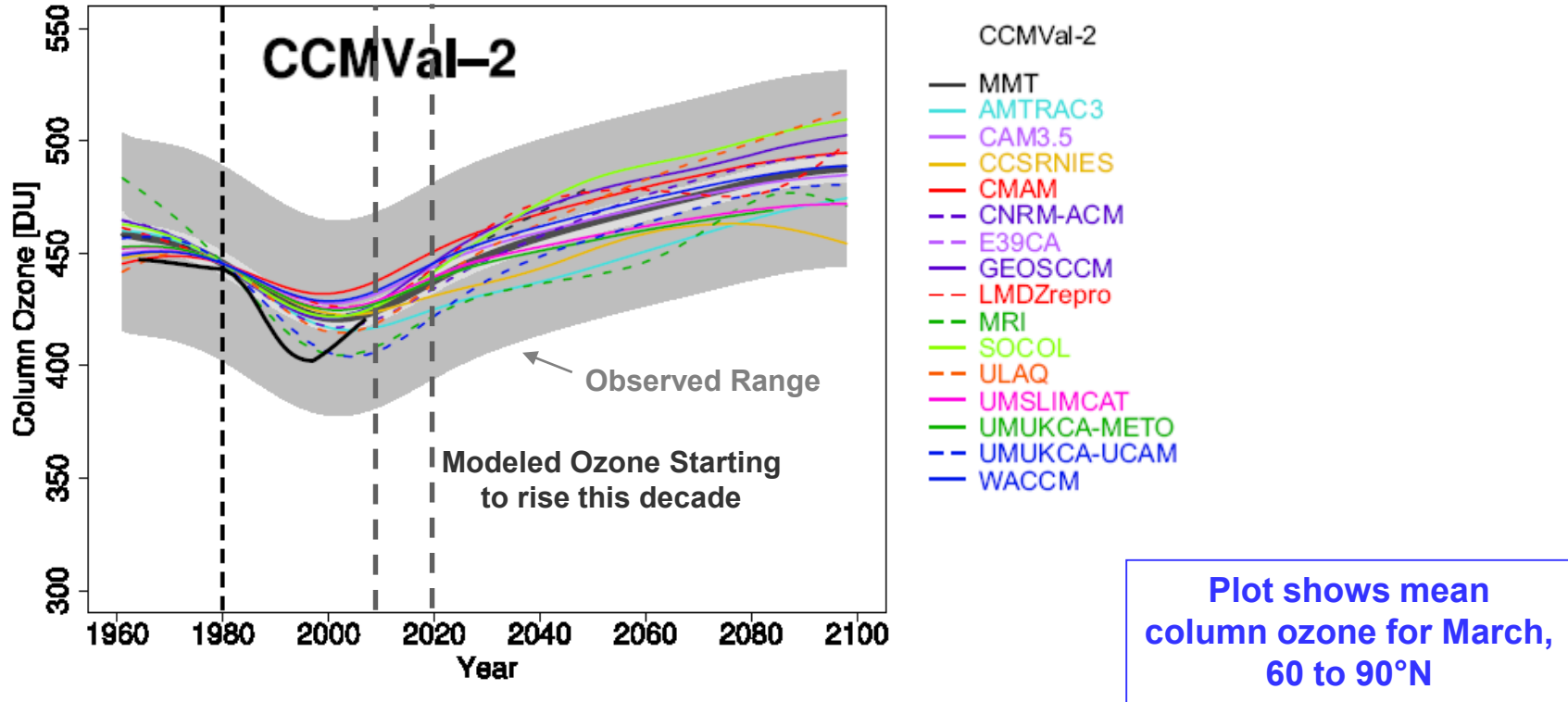


Figure 9.11, CCMVal-2

http://www.atmosp.physics.utoronto.ca/SPARC/ccmval_final/index.php

Latest generation of chemistry climate models do not reproduce the results of Shindell et al. (1998)
No consensus on how Arctic ozone will be affected by climate change.

Arctic Ozone Loss - Climate and Chemistry Coupling

March O₃ Column 60°N–90°N

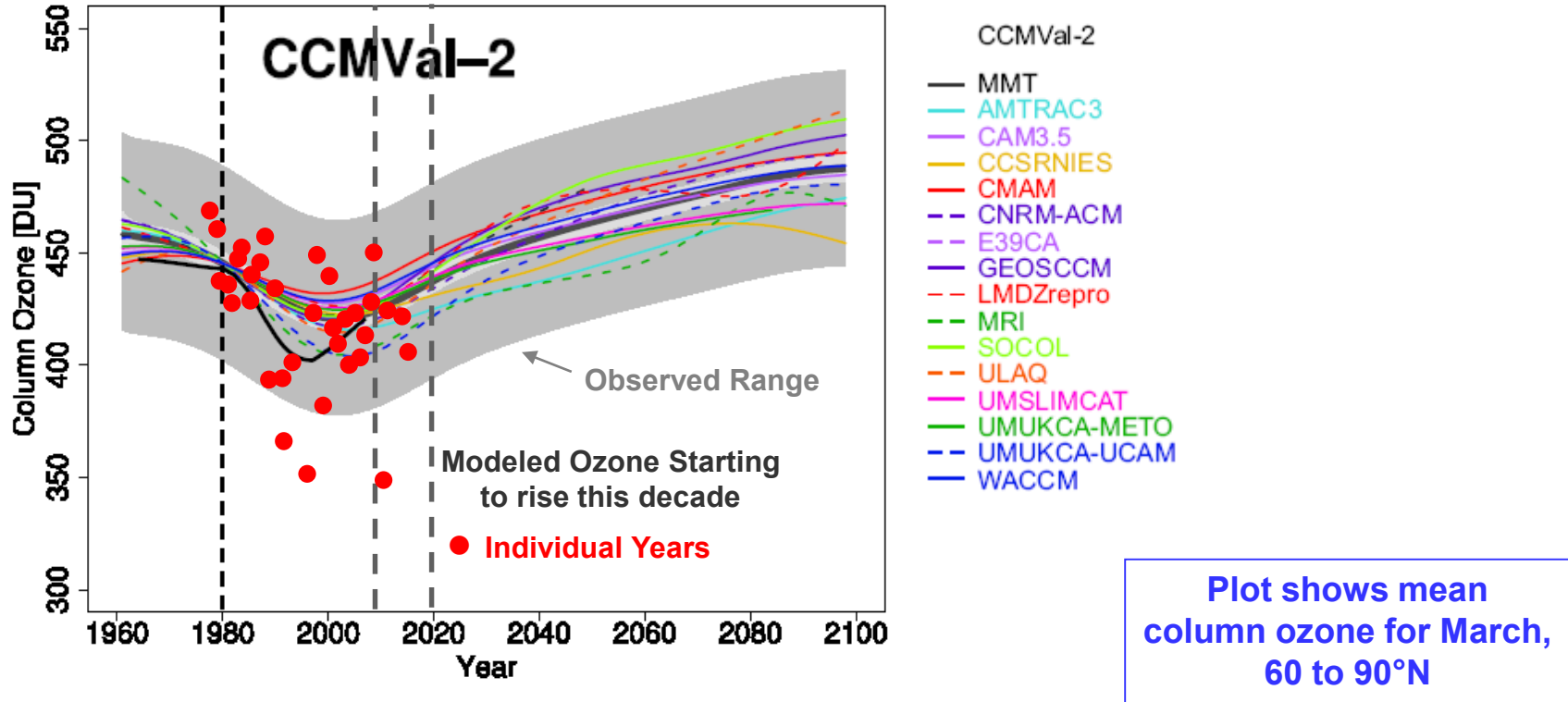


Figure 9.11, CCMVal-2

http://www.atmosp.physics.utoronto.ca/SPARC/ccmval_final/index.php

Data for individual years suggest latest generation of models in need of major improvement

Arctic Ozone Loss and Climate Change

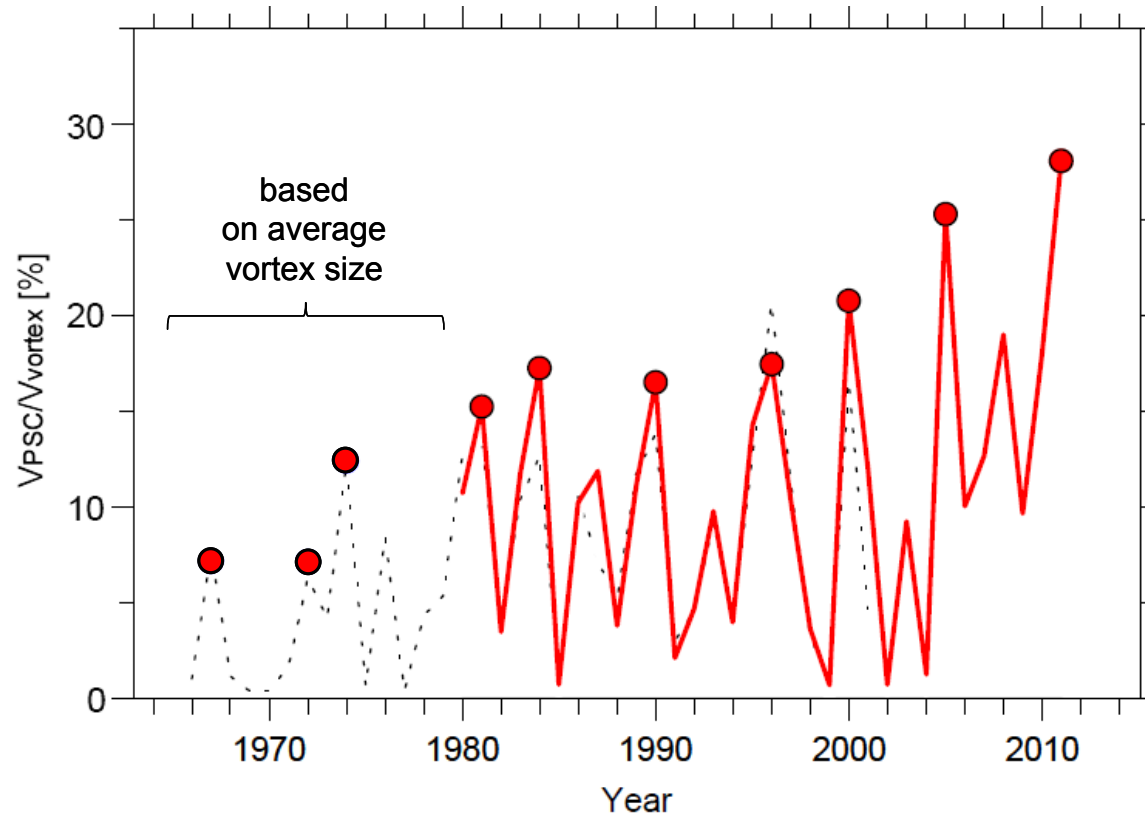


Figure 2-16, WMO/UNEP (2011)

Updated for Arctic winter 2011
and **normalized for vortex area**

- Factor of three increase in the maximum of V_{PSC} over the past four decades
- Coldest Arctic winters may be getting colder !!!
- Cause uncertain: might be due to increased radiative efficiency of vortex during dynamically quiet years

Arctic Ozone Loss and Climate Change

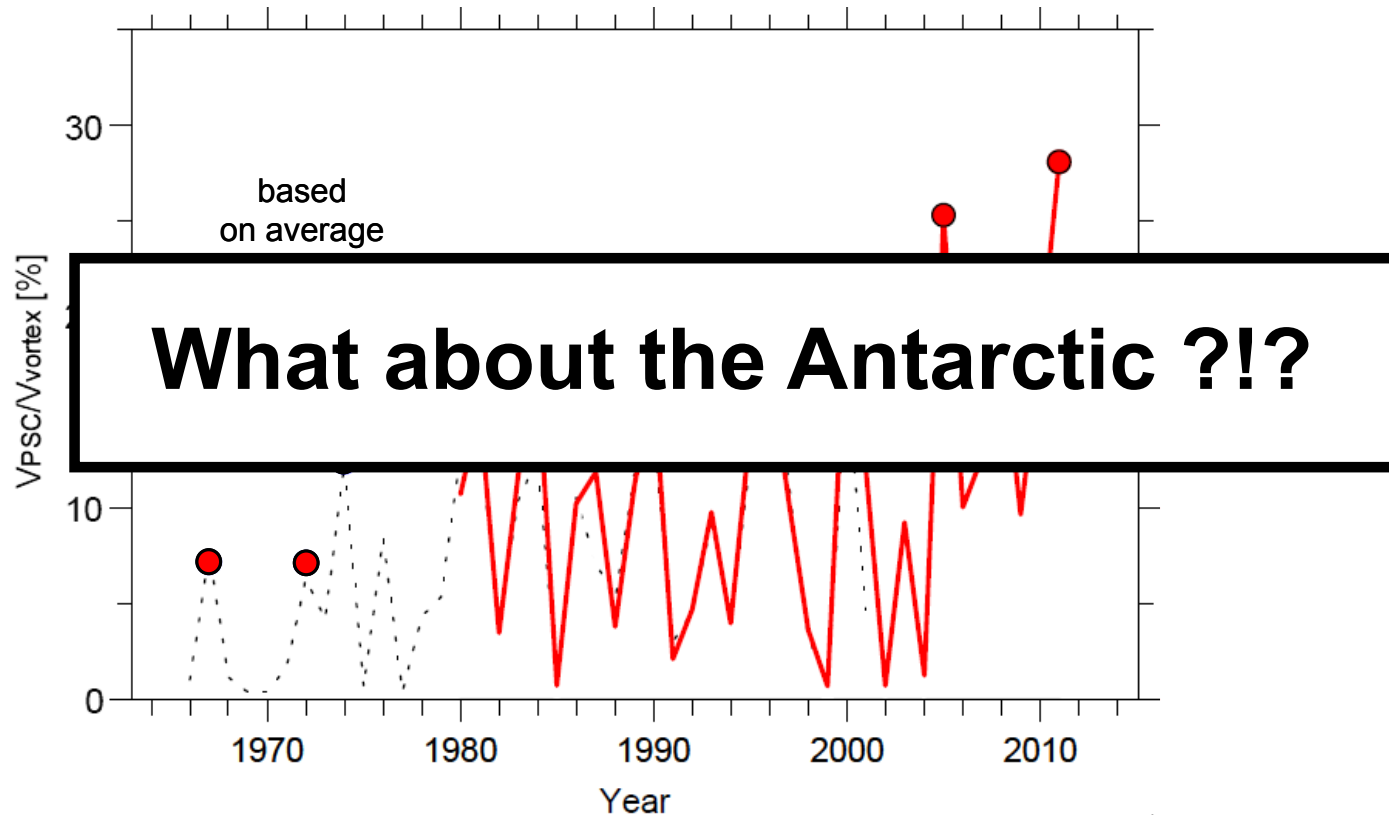
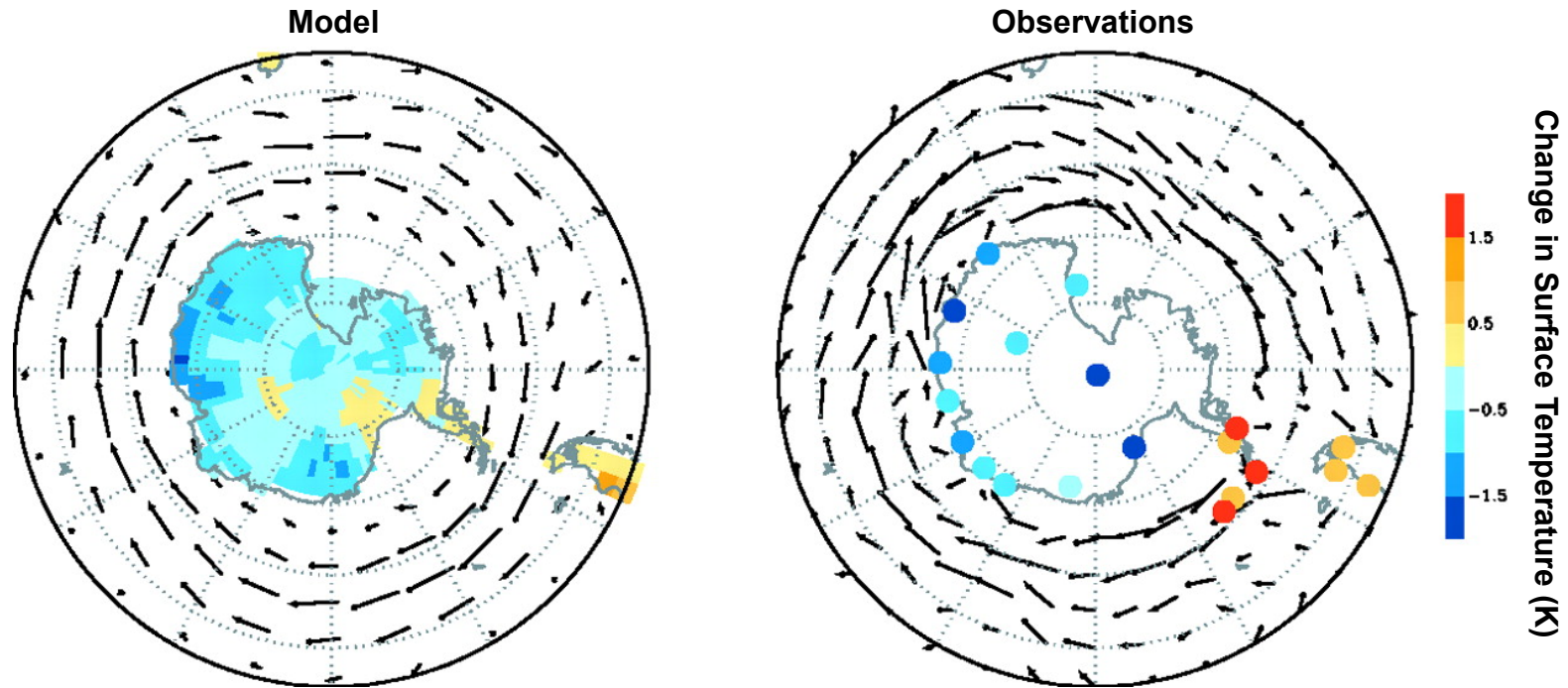


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The Ozone Hole may have shielded the Antarctic surface from warming!



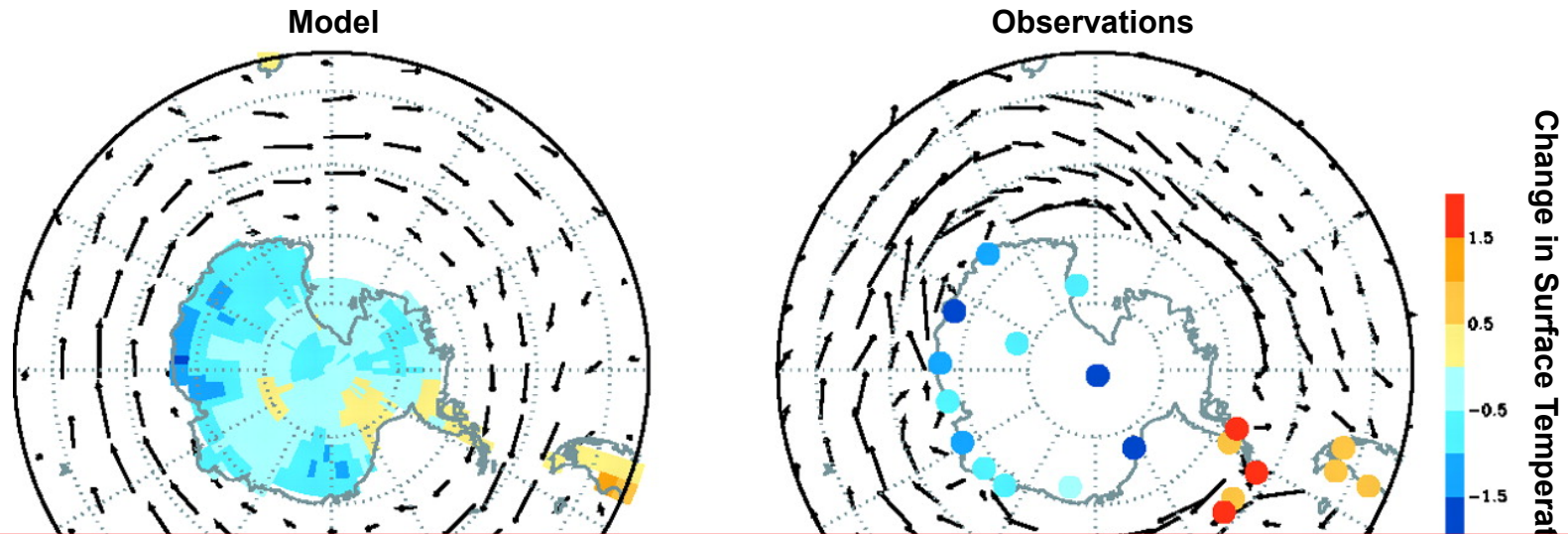
Simulated and observed changes in surface temperature (K) and wind speed, 1969 to 2000, averaged over December to May. The longest wind vector corresponds to 4 m/s.

Gillett and Thompson, *Science*, 2003

As ozone depletion occurs:

The positive phase of the southern annular mode (SAM) increases, causing Antarctic surface westerlies to intensify, resulting in cooling of Antarctic continent

The Ozone Hole may have shielded the Antarctic surface from warming!



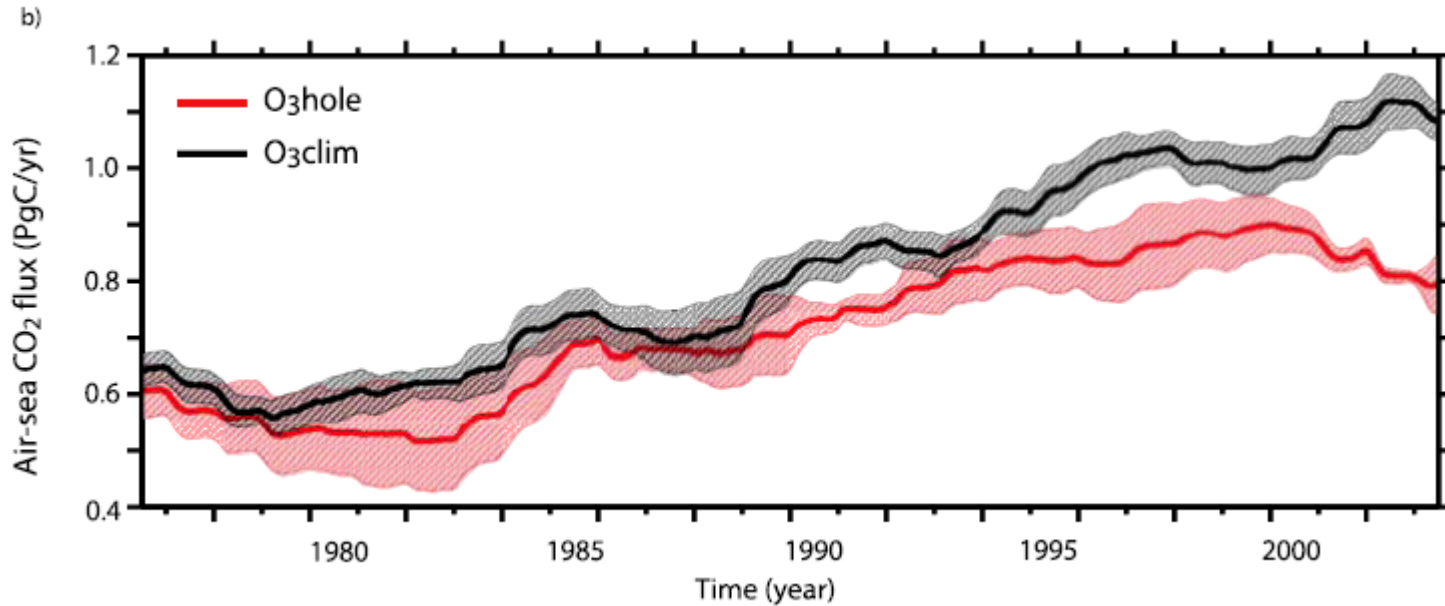
SAM: difference in zonal mean sea-level pressure between 40°S and 65°S. The pattern associated with SAM is a nearly annular pattern with a large low pressure anomaly centered on the South Pole and a ring of high pressure anomalies at mid-latitudes. The SAM effects storm tracks, precipitation patterns, etc.

http://www.climate.be/textbook/chapter5_node6.html

As ozone depletion occurs:

The positive phase of the southern annular mode (SAM) increases, causing Antarctic surface westerlies to intensify, resulting in cooling of Antarctic continent

The Ozone Hole may have lead to increased ventilation of CO₂ from southern ocean



(b) Integrated air to sea CO₂ flux (south of 40°S) showing stratospheric ozone depletion (O₃hole) significantly reduces CO₂ uptake (relative to O₃clim), and is strongly correlated with changes in $\Delta p\text{CO}_2$.

Lenton *et al.*, *GRL*, 2009

As ozone depletion occurs:

The positive phase of the southern annular mode (SAM) increases, causing Antarctic surface westerlies to intensify, resulting in increased ventilation of CO₂ from southern ocean

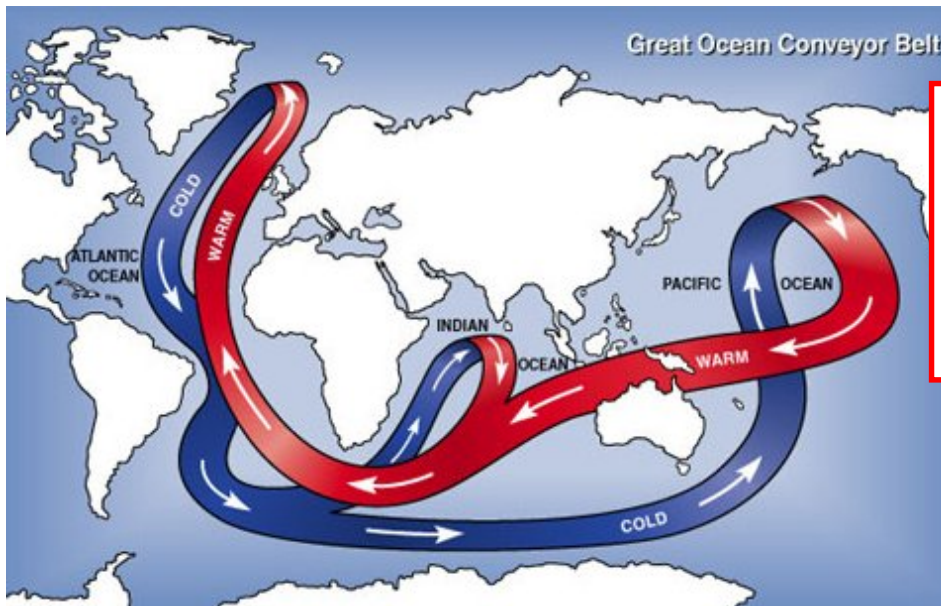
Uptake of Atmospheric CO₂ by Oceans

– Solubility Pump:

- More CO₂ can dissolve in cold polar waters than in warm equatorial waters. As major ocean currents (e.g. the Gulf Stream) move waters from tropics to the poles, they are cooled and take up atmospheric CO₂
- Deep water forms at high latitude. *As deep water sinks, ocean carbon (ΣCO_2) accumulated at the surface is moved to the deep ocean interior.*

– Biological Pump:

- Ocean biology limited by availability of nutrients such as NO₃⁻, PO₄⁻, and Fe²⁺ & Fe³⁺. Ocean biology is never carbon limited.
- Detrital material “rains” from surface to deep waters, *contributing to higher CO₂ in intermediate and deep waters*



In the model, this elevated oceanic CO₂ is returned to the atmosphere due to stronger winds, which lead to more ocean turbulence ... all due to the Antarctic ozone hole !

Lecture 5, Slide 25

http://science.nasa.gov/headlines/y2004/05mar_arctic.htm

Chemistry Climate Coupling

CCMs (chemistry climate models): developed to quantify impacts of climate change on stratospheric ozone and impacts of ozone depletion/recovery on climate:

As GHGs rise:

- 1. Brewer-Dobson circulation predicted to accelerate leading to:
 - a) less ozone in tropical lower stratosphere (“permanent depletion”)**
 - b) more ozone in mid-latitude lower stratosphere (“super recovery”)****
- 2. Upper stratosphere cools, slowing down rate limiting steps for ozone loss and therefore leading to “super recovery”**
- 3. Eventually, CH₄ and N₂O will drive future levels of ozone**

Data analysis suggests “coldest Arctic winters getting colder”:

- 1. Possibly due to rising GHGs**
- 2. Not represented by CCMs**

As Antarctic ozone depletion occurs:

The positive phase of the southern annular mode (SAM) increases, causing Antarctic surface westerlies to intensify, resulting in:

- 1. Cooling of Antarctic continent**
- 2. Increased ventilation of CO₂ from southern ocean**