

# Pollution of Earth's Troposphere: Acid Rain & Aerosols

## AOSC 433/633 & CHEM 433

Ross Salawitch

**Class Web Sites:** <http://www.atmos.umd.edu/~rjs/class/spr2015>

**Lecture 13**  
**26 March 2015**

# Overview of Aerosols

- Aerosols aka particulate matter (PM)
- Size generally ranges from 0.005  $\mu\text{m}$  to 100  $\mu\text{m}$  diameter
- Can be liquid or solid
- Dust: solid, produced by grinding or crushing operation
- Fumes: formed by condensation of gases
- Smoke or soot: carbon particles resulting from incomplete combustion
- SOA: secondary organic aerosol, formed by condensation of decomposition products of VOCs (volatile organic compounds) including isoprene ( $\text{C}_5\text{H}_8$ ) which is mainly biogenic and benzene ( $\text{C}_6\text{H}_6$ ) which is mainly anthropogenic
- PM can be emitted directly as carbonaceous material (primary pollutant) or formed in atmosphere upon condensation/transformation of gaseous emissions of  $\text{SO}_2$ ,  $\text{NO}_x$ , and  $\text{NH}_3$ 
  - Eastern US: sulfates dominate due to greater reliance on coal-fired power plants
  - Western US: carbon and nitrates dominate due to agriculture & transportation

# Overview of Aerosols

- Health effects driven by size and chemical composition
- Smaller particles most hazardous
- Benzene-like compounds called polycyclic aromatic hydrocarbons (PAH) most hazardous



<http://www.barnesandnoble.com/w/polycyclic-aromatic-hydrocarbons-pierre-a-haines>

- Fall speed of aerosols varies as (diameter)<sup>2</sup>  
2  $\mu\text{m}$  diameter particle has residence time in 1 km of atmosphere of 2 months, if removed by only gravitational settling  
⇒ small particles are suspended in the atmosphere until removed by \_\_\_\_\_ ?

# Health Effects of Air Pollution

International New York Times

## Air Pollution Raises Stroke Risk

By NICHOLAS BAKALAR MARCH 24, 2015 4:30 PM 7 Comments



Air pollution — even for just one day — significantly increases the risk of stroke, a large review of studies has found.

Researchers pooled data from 103 studies involving 6.2 million stroke hospitalizations and deaths in 28 countries.

The analysis, [published online in BMJ](#), found that all types of pollution except ozone were associated with increased risk for stroke, and the higher the level of pollution, the more strokes there were.

Daily increases in pollution from nitrogen dioxide, sulfur dioxide, carbon monoxide and particulate matter were associated with corresponding increases in strokes and hospital admissions. The strongest associations were apparent on the day of exposure, but increases in particulate matter had longer-lasting effects.

The exact reason for the effect is unclear, but studies have shown that air pollution can constrict blood vessels, increase blood pressure and increase the risk for blood clots. Other research has tied air pollution to a higher risk of heart attacks, stroke and other ills.

<http://well.blogs.nytimes.com/2015/03/24/air-pollution-raises-stroke-risk>

## BMJ: British Medical Journal

### Short term exposure to air pollution and stroke: systematic review and meta-analysis

Anoop S V Shah,<sup>1</sup> Kuan Ken Lee,<sup>1</sup> David A McAllister,<sup>2</sup> Amanda Hunter,<sup>1</sup> Harish Nair,<sup>2</sup> William Whiteley,<sup>3</sup> Jeremy P Langrish,<sup>1</sup> David E Newby,<sup>1</sup> Nicholas L Mills<sup>1</sup>

<sup>1</sup>BHF/University Centre for Cardiovascular Science, University of Edinburgh, Edinburgh EH16 4SB, UK

<sup>2</sup>Centre of Population Health Sciences, University of Edinburgh, Edinburgh, UK

<sup>3</sup>Centre for Clinical Brain Sciences, University of Edinburgh, Edinburgh, UK

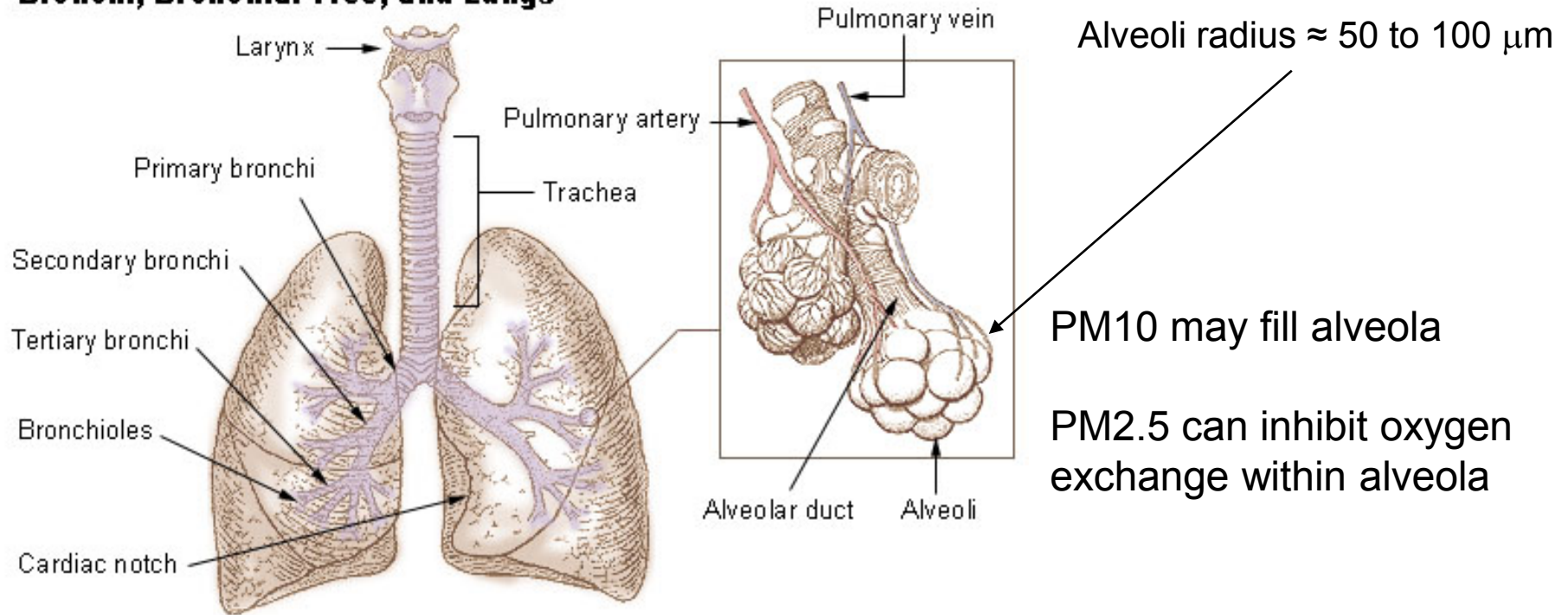
Admission to hospital for stroke or mortality from stroke was associated with an increase in concentrations of carbon monoxide (relative risk 1.015 per 1 ppm, 95% confidence interval 1.004 to 1.026), sulphur dioxide (1.019 per 10 ppb, 1.011 to 1.027), and nitrogen dioxide (1.014 per 10 ppb, 1.009 to 1.019). Increases in PM<sub>2.5</sub> and PM<sub>10</sub> concentration were also associated with admission and mortality (1.011 per 10  $\mu\text{g}/\text{m}^3$  (1.011 to 1.012) and 1.003 per 10  $\mu\text{g}/\text{m}^3$  (1.002 to 1.004), respectively).

Gaseous and particulate air pollutants have a marked and close temporal association with admissions to hospital for stroke or mortality from stroke. Public and environmental health policies to reduce air pollution could reduce the burden of stroke.

The lead author, Dr. Anoop Shah, a lecturer in cardiology at the University of Edinburgh, said that there was little an individual can do when air pollution spikes. “If you’re elderly, or have co-morbid conditions, you should stay inside,” he said. But policies leading to cleaner air would have the greatest impact, he said. “It’s a question of getting cities and countries to change.”

# Health Effects of Aerosols

## Bronchi, Bronchial Tree, and Lungs



Exposure to elevated levels of particulate matter leads to increase risk of respiratory illnesses, cardiopulmonary disease, ischemic heart disease, and heart attacks

# Health Effects of Aerosols

## Assessment of Public Health Risks Associated with Atmospheric Exposure to PM<sub>2.5</sub> in Washington, DC, USA

Natasha A. Greene<sup>1\*</sup>, and Vernon R. Morris<sup>1,2</sup>

<sup>1</sup>Program in Atmospheric Sciences, Howard University, Washington, DC 20059, USA

<sup>2</sup>Department of Chemistry, Howard University, Washington, DC 20059, USA

Our findings show that there are significant risks of ward-specific pediatric asthma emergency room visits (ERV). Results also illustrate lifetime excess lung cancer risks, exceeding the  $1 \times 10^{-6}$  threshold for the measured levels of particulate matter and heavy metals (chromium and arsenic) on behalf of numerous subpopulations in the DC selected wards.

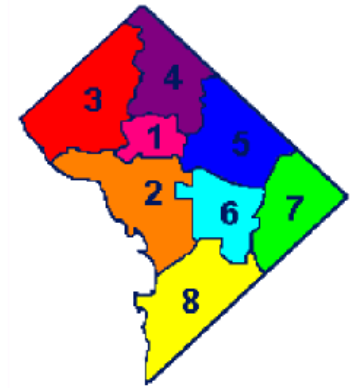


Figure 1: Washington, DC Wards Schematic

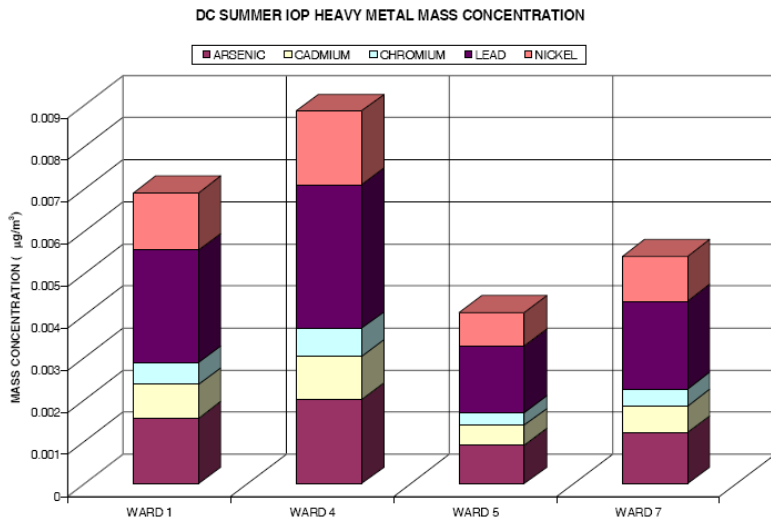


Figure 5: Heavy Metal Content of Fine PM for Summer IOP

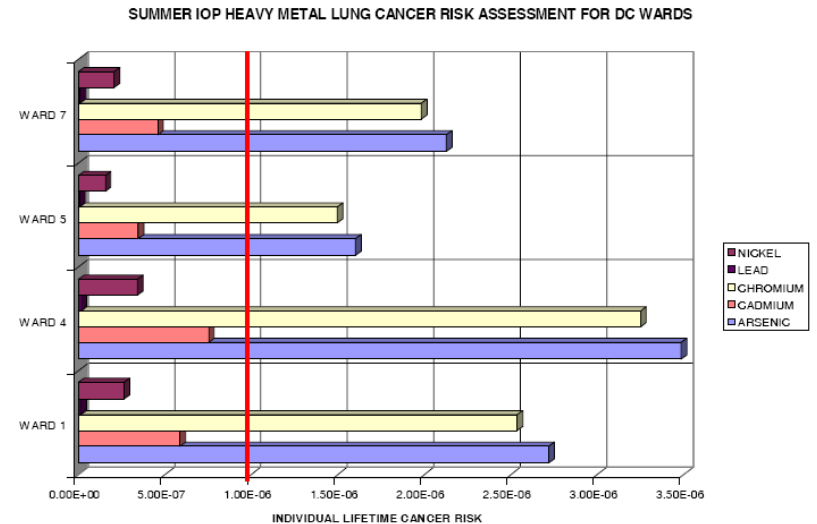
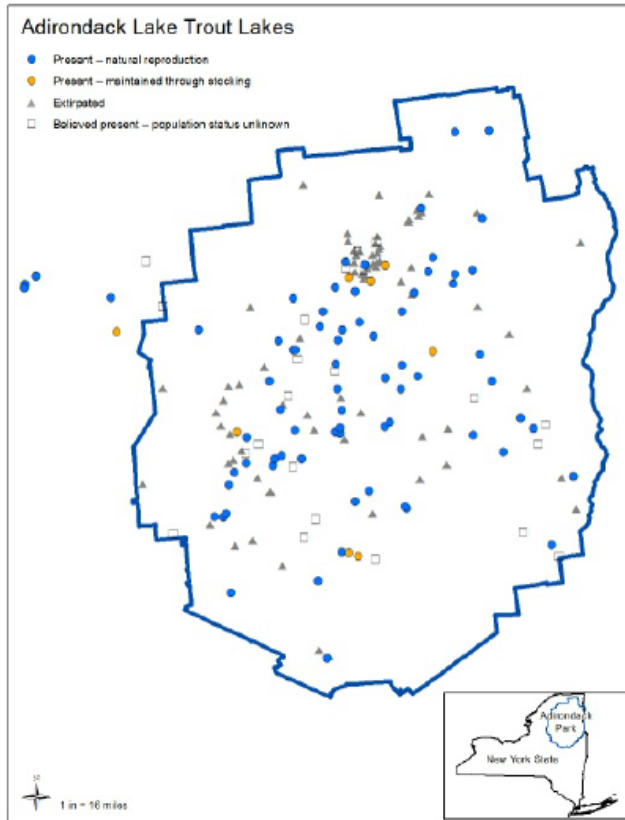


Figure 10: Summer IOP Lifetime Excess Lung Cancer Risk by DC Wards.

# Lake Acidification

## Adirondack Park, New York



- Largest American park outside of Alaska (9,300 square miles)
- Suffered worse damage due to acid rain than any other region in the U.S.
- 700 lakes had become too acidic to support native aquatic species
- Considerable recent progress after extensive legislative battles:

The EPA states that from 1990 to 2013, there was a seventy-seven percent decrease in sulfur dioxide emissions and a forty-nine percent decrease in total nitrogen oxide emissions.

Charles Driscoll is a professor at Syracuse University who has been studying acid rain in the Adirondacks for decades. Driscoll noted that because of the reductions that many lakes are now once again supporting species like brook trout. However, he also said that some lakes will take centuries to recover.

“We’ve seen a partial recovery, but there is still quite a bit of damage, particularly on soils and streams,” Driscoll said. “I think that we’re part way there ... but we need additional reductions to more fully recover.”

<http://www.nature.org/ourinitiatives/regions/northamerica/unitedstates/newyork/adirondacks-lake-trout-report-december-2014.pdf>

See also <http://www.adirondackalmanack.com>

# Cultural Degradation

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In 1944



At present

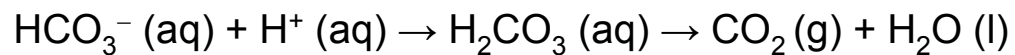
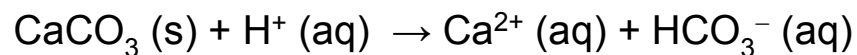
Figure 6.22, Chemistry in Context.  
Limestone statue of George Washington, NYC

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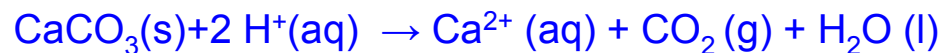


Figure 6.24, Chemistry in Context.  
Mayan art, Mexico.

Marble limestone, composed mainly of calcium carbonate ( $\text{CaCO}_3$ ), slowly dissolves in the presence of hydrogen ion:



or:

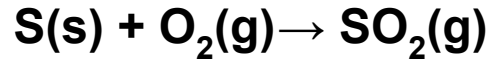




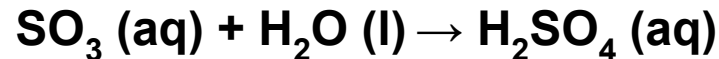
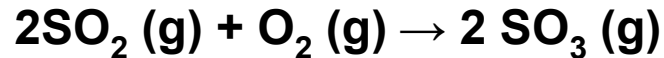
# Acid Rain: SO<sub>2</sub>

**Chemical formula of coal: C<sub>135</sub>H<sub>96</sub>O<sub>9</sub>NS (S varies with coal type)**

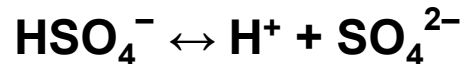
**Combustion of leads to release of sulfur dioxide (SO<sub>2</sub>)**



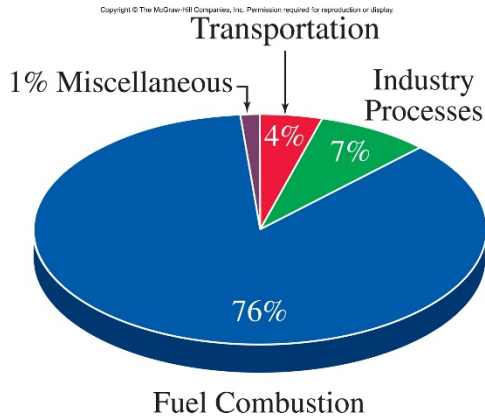
**SO<sub>2</sub> reacts with O<sub>2</sub> to form sulfur trioxide (SO<sub>3</sub>)**



**Followed by:**



# SO<sub>2</sub> Sources (US)



Primary source of SO<sub>2</sub> is fuel combustion; emissions from this sector are decreasing.

Emissions from transportation are small and largely unchanged.

Figure 6.14, Chemistry in Context. US SO<sub>2</sub> emission sources, 2007

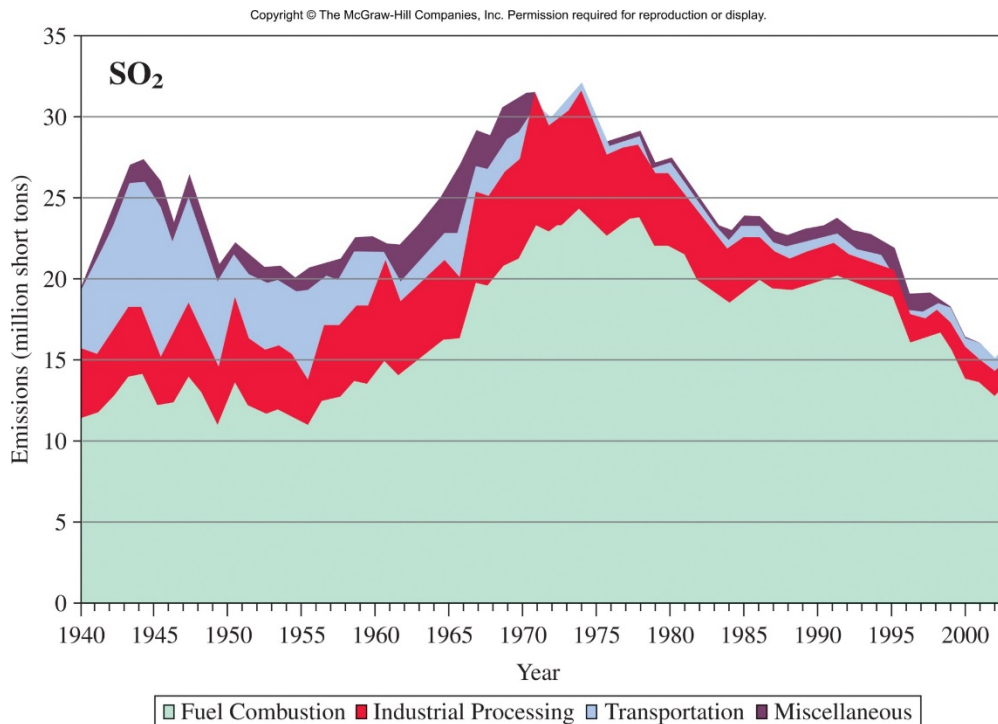
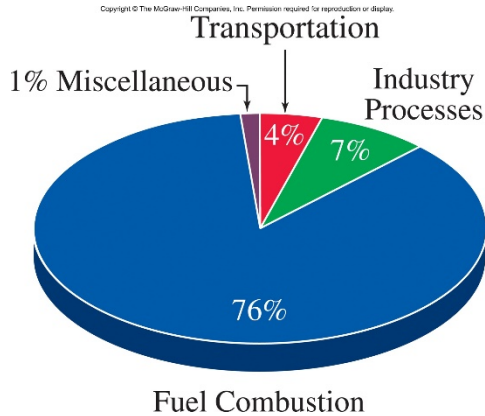


Figure 6.21, Chemistry in Context. US SO<sub>2</sub> emissions, 1940 to 2003

# SO<sub>2</sub> Sources (US)

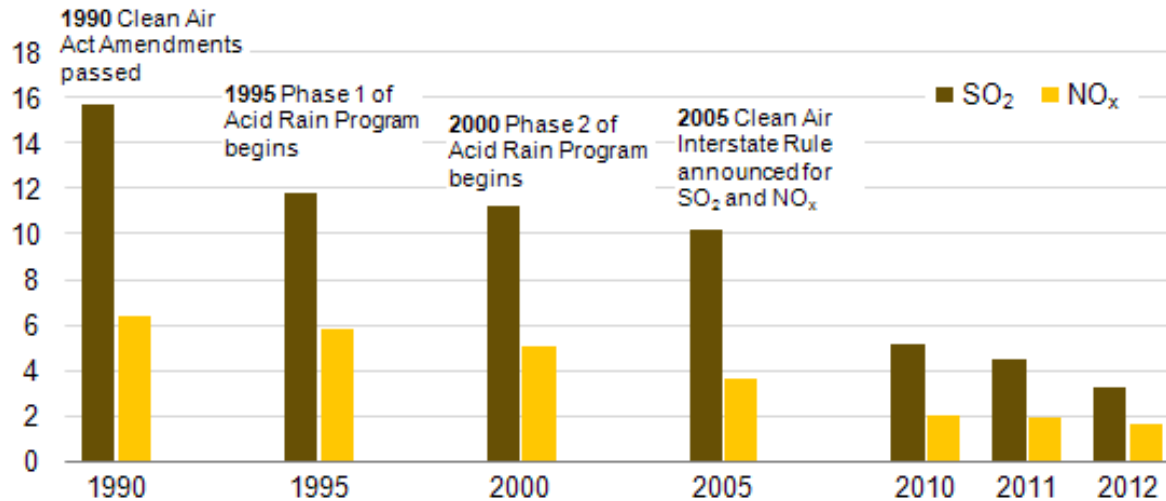


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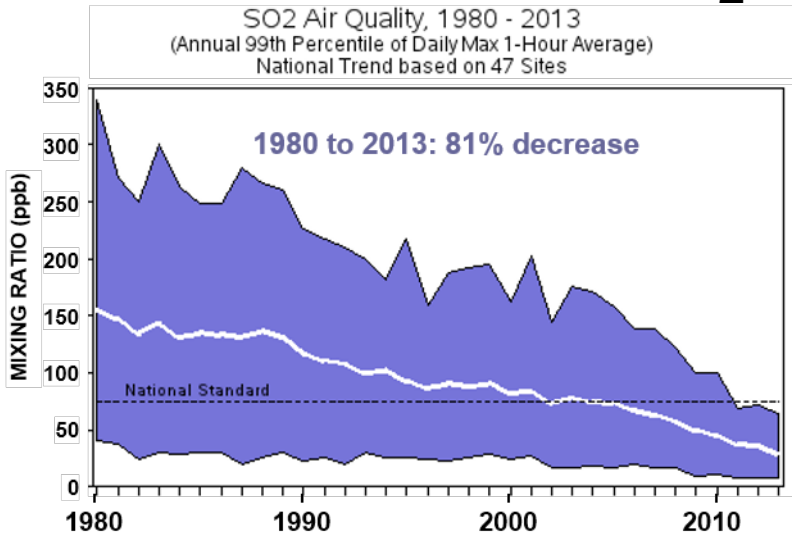
Figure 6.14, Chemistry in Context. US SO<sub>2</sub> emission sources, 2007

## SO<sub>2</sub> and NO<sub>x</sub> emissions from the electric power sector million short tons



<http://www.eia.gov/todayinenergy/detail.cfm?id=10151>

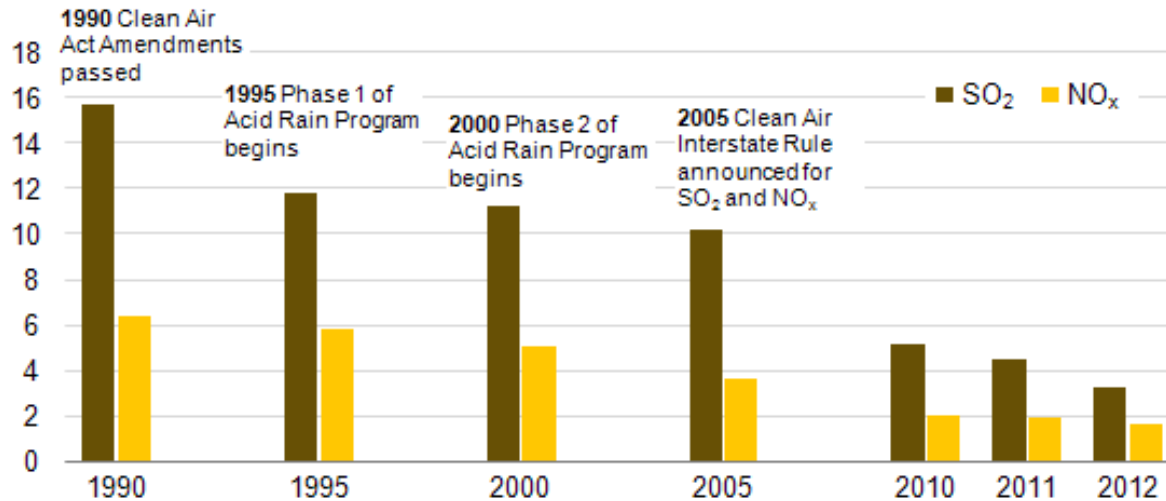
# SO<sub>2</sub> Sources (US)



Observed SO<sub>2</sub> dropping, largely in compliance with NAAQS 1 hr standard of 75 ppb

<http://www.epa.gov/airtrends>

SO<sub>2</sub> and NO<sub>x</sub> emissions from the electric power sector  
million short tons



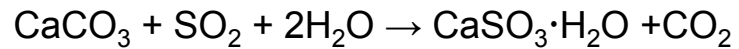
<http://www.eia.gov/todayinenergy/detail.cfm?id=10151>

# Removal of SO<sub>2</sub> from Power Plants

## SO<sub>2</sub> Control: Flue Gas Desulphurization



Pulverized limestone (CaCO<sub>3</sub>) is mixed with water to make a slurry sprayed into flue gas, resulting in:



Cost on order \$200 million per unit

Another technology using lime, CaO, exists but is not in widespread use due to high cost of lime

## Md Coal Plants with Capacity over 400 Mw

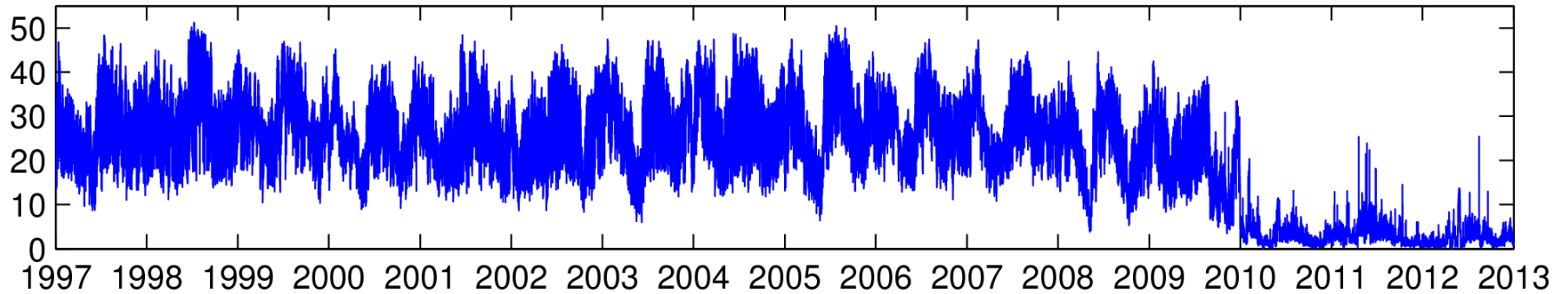
Plant	County	Capacity, MW	Year Built	SCR	FGD
Brandon Shores	Anne Arundel	1273	1984, 1991	Partial	Yes
Morgantown	Charles	1252	1970, 1971	Yes	Yes
Chalk Point	Prince Georges	728	1964, 1965	No	Yes
Dickerson	Montgomery	588	1959, 1960, 1962	No	Yes
Herbert Wagner	Anne Arundel	977	1959, 1966	Partial	No
Crane	Baltimore	400	1961, 1963	Partial	No

Note: A 7<sup>th</sup> coal plant, R. Paul Smith Power Station in Williamsport (near Hagerstown), closed on 1 Sept 2012

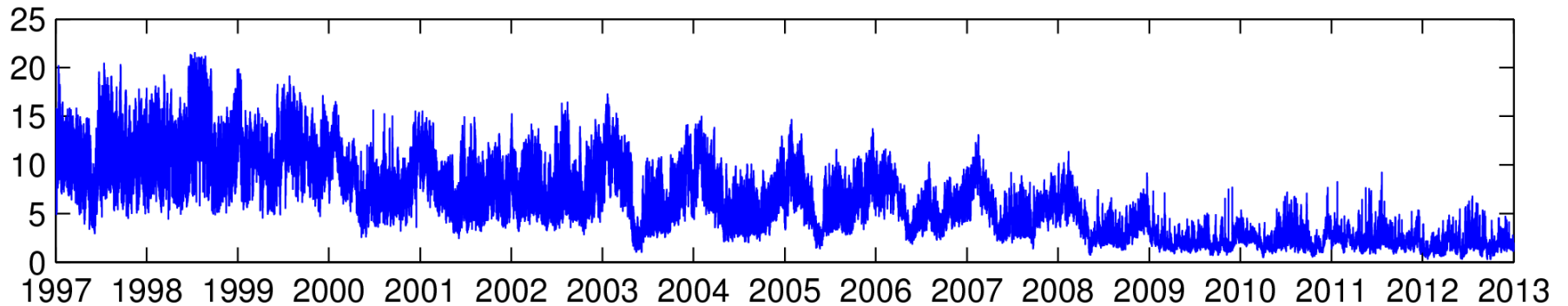
Sources: [http://www.sourcewatch.org/index.php/Maryland\\_and\\_coal](http://www.sourcewatch.org/index.php/Maryland_and_coal)  
<http://raven-power.com/plants/brandon-shores>  
<http://www.industcards.com/st-coal-usa-md.htm>

# Maryland Trends

SO<sub>2</sub>, 10<sup>3</sup> kg/hr



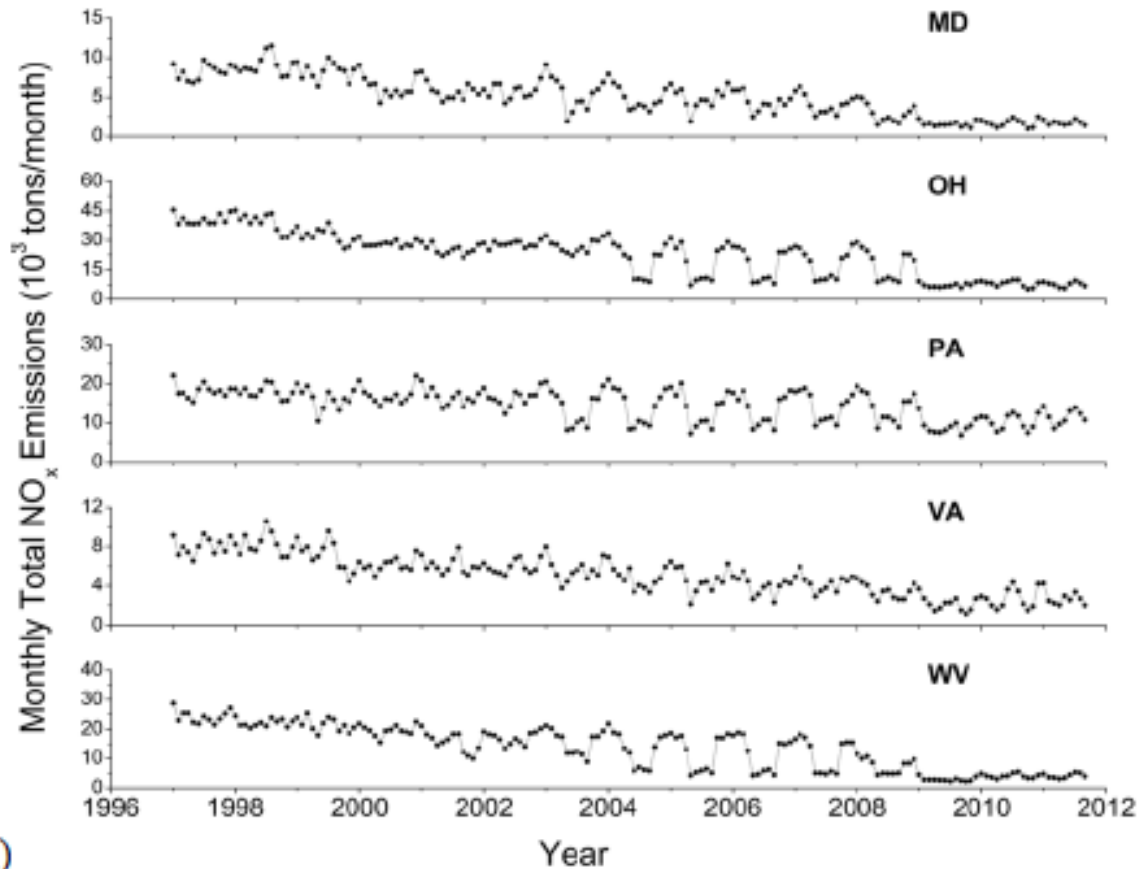
NO<sub>x</sub>, 10<sup>3</sup> kg/hr



Courtesy: K. Vinnikov

# Trends in power plant NO<sub>x</sub> emission, region

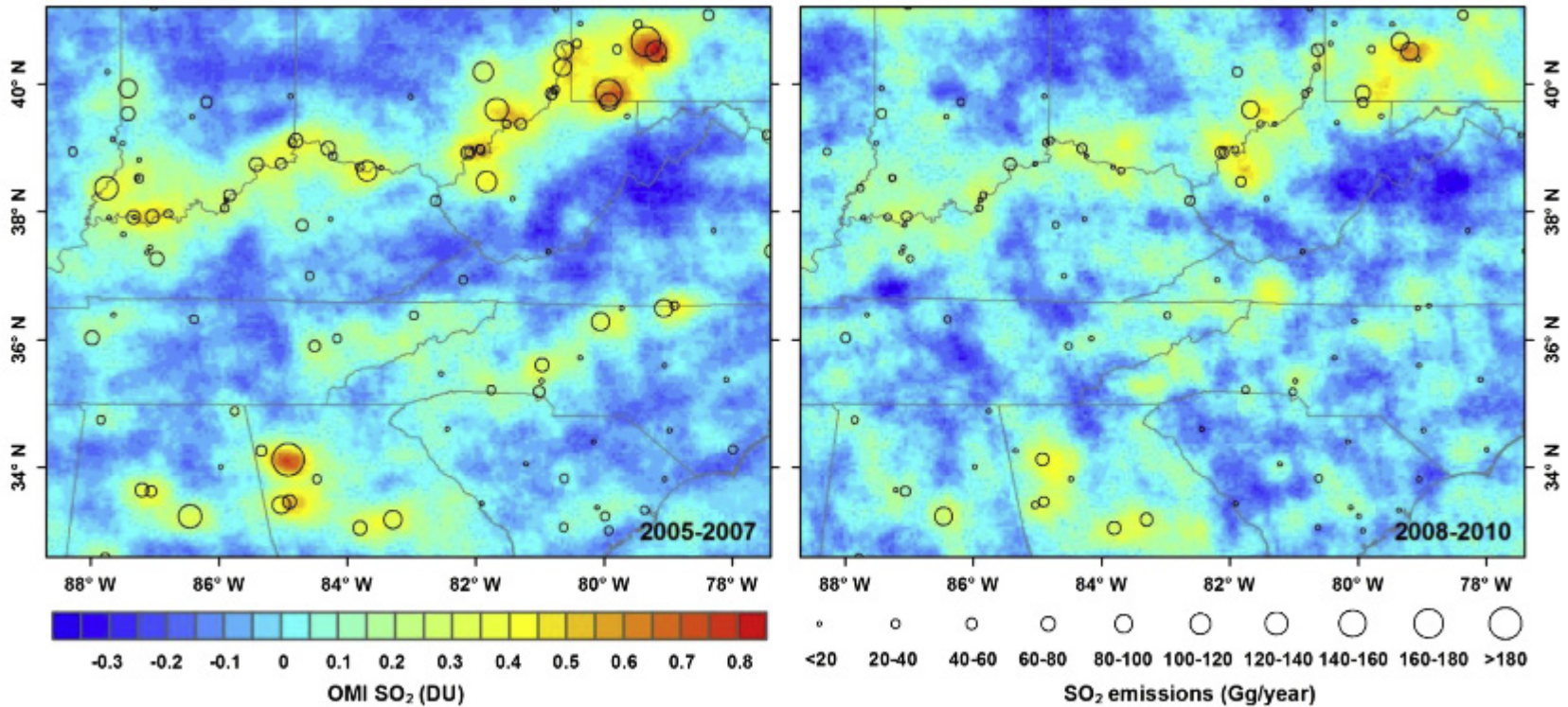
H. He et al.: Trends in emissions and concentrations of air pollutants



He et al., *ACP*, 2013



# SO<sub>2</sub> From Space (US)

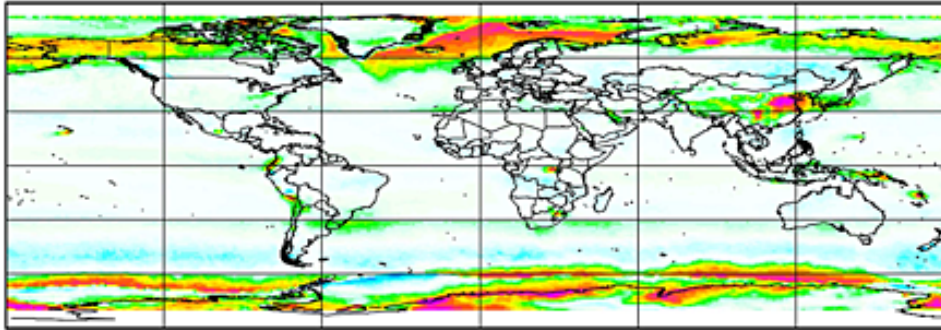


**Fig. 4.** Mean SO<sub>2</sub> burdens over the Ohio River Basin for 2005–2007 (left) and 2008–2010 (right) measured by OMI, confirming a substantial reduction in SO<sub>2</sub> pollution around the largest coal-fired power plants, as a result of the implementation of SO<sub>2</sub> emission control measures (adapted from NASA Earth Observatory, as reported in Fioletov et al., 2011).

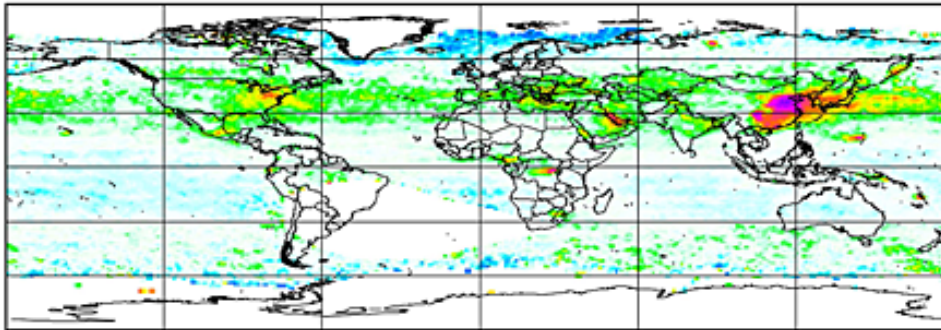
Streets et al., Atmos. Envir., 2013

# SO<sub>2</sub> From Space (Global)

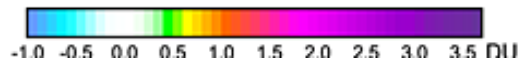
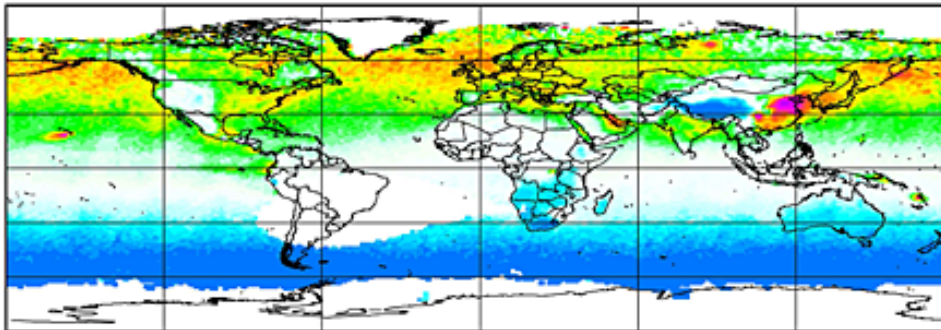
OMI, 2005-2010



SCIAMACHY, 2005-2009



GOME 2 DLR, 2007-2010



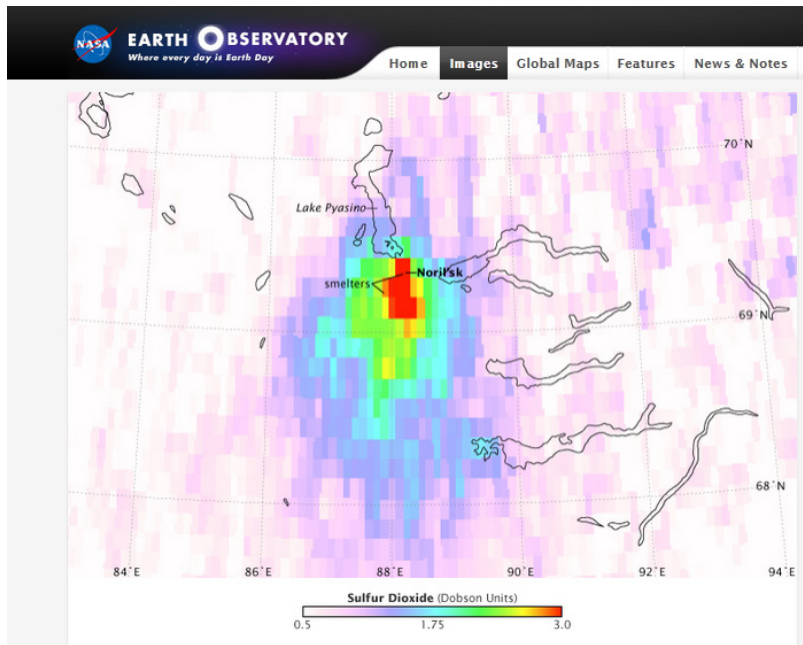
Fioletov et al., JGR, 2013

# SO<sub>2</sub> From Space (Norilsk, Russia)



Copper and nickel smelters in Norilsk, Russia are largest anthropogenic point source of SO<sub>2</sub>

[http://news.bbc.co.uk/1/hi/in\\_pictures/6529225.stm](http://news.bbc.co.uk/1/hi/in_pictures/6529225.stm)



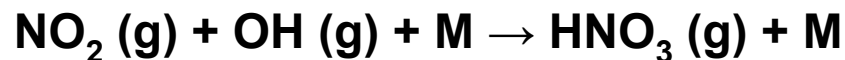
Enhanced SO<sub>2</sub> in this region readily apparent from space

<http://earthobservatory.nasa.gov/IOTD/view.php?id=36063>

# Acid Rain: NO<sub>x</sub>

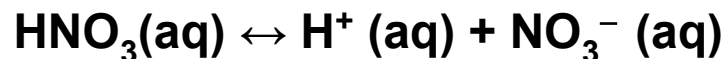
**NO<sub>x</sub> plays major role in tropospheric O<sub>3</sub> formation.**

**In Lecture 12, we emphasize the critical importance of radical termination:**



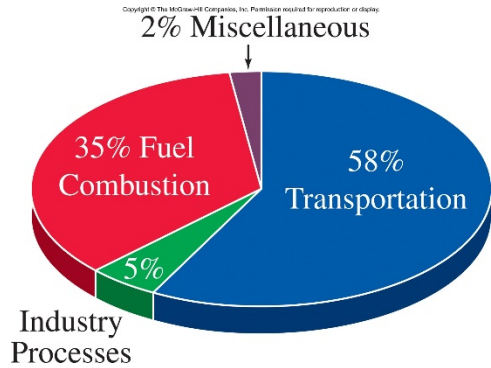
**Nitric acid, HNO<sub>3</sub>, is soluble. Hence, in the presence of droplets, HNO<sub>3</sub> (g) can become HNO<sub>3</sub> (aq)**

**HNO<sub>3</sub> (aq) then dissociate:**



**and well “oops, we did it again”**

# NO<sub>x</sub> Sources (US)



Primary source of NO<sub>2</sub> is transportation; EPA inventory suggests emissions from this sector are holding steady, whereas the UMD Atmos Chem group believes emission in the mid-Atlantic have fallen dramatically (Anderson et al., Atmos Environ, 2014)

Emissions from fuel combustion primary driver of inventory decline

Figure 6.16, Chemistry in Context. US NO<sub>x</sub> emission sources, 2007

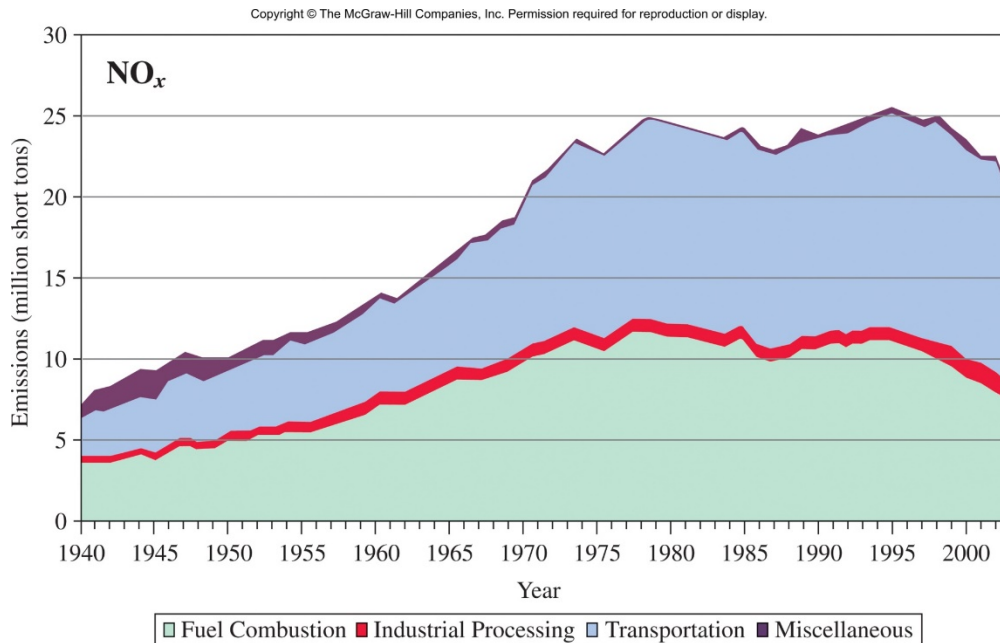
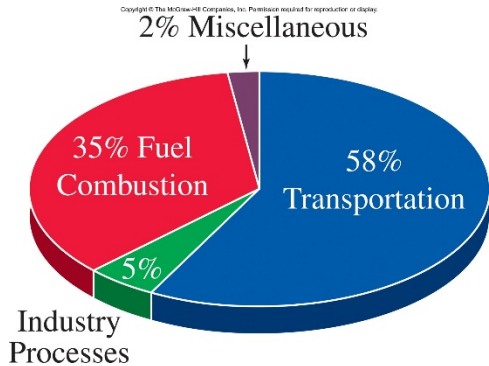


Figure 6.21, Chemistry in Context. US NO<sub>x</sub> emissions, 1940 to 2003

# NO<sub>x</sub> Sources (US)

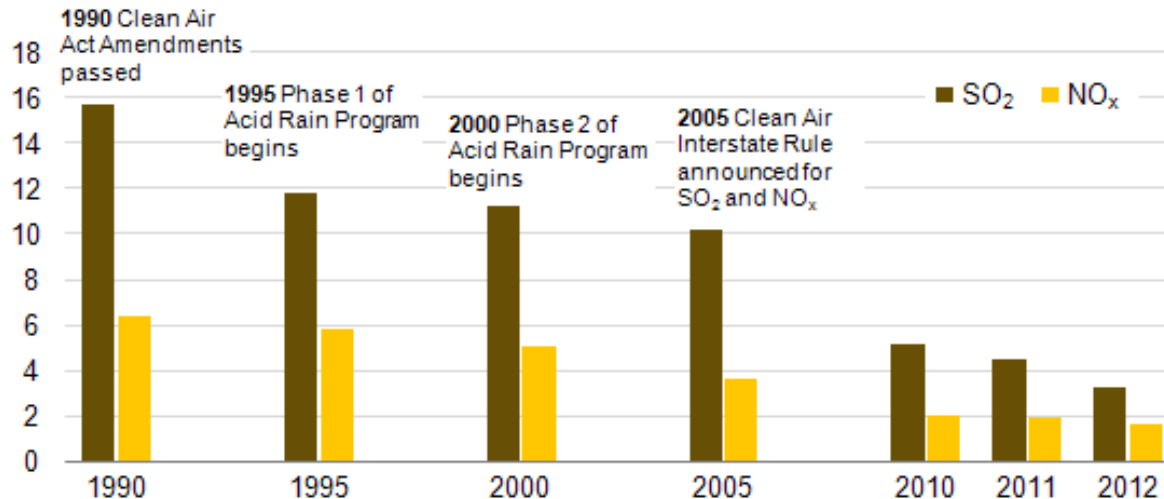


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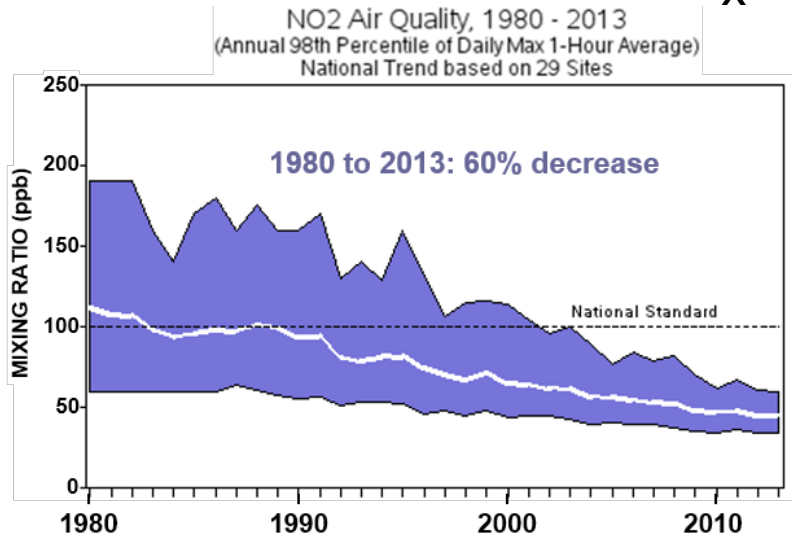
Figure 6.16, Chemistry in Context. US NO<sub>x</sub> emission sources, 2007

## SO<sub>2</sub> and NO<sub>x</sub> emissions from the electric power sector million short tons



<http://www.eia.gov/todayinenergy/detail.cfm?id=10151>

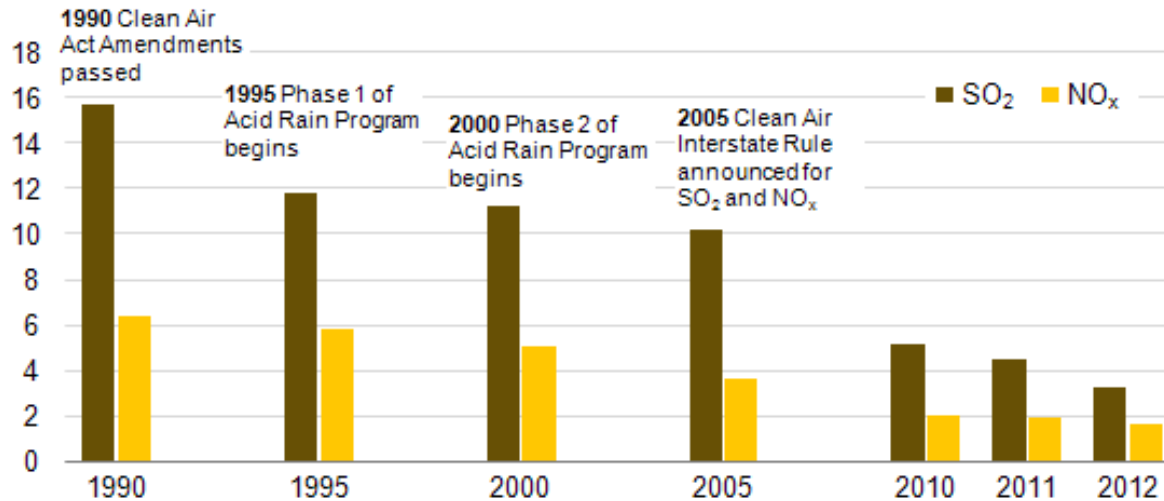
# NO<sub>x</sub> Sources (US)



Observed NO<sub>2</sub> dropping, largely in compliance with NAAQS 1 hr standard of 100 ppb

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SO<sub>2</sub> and NO<sub>x</sub> emissions from the electric power sector  
million short tons

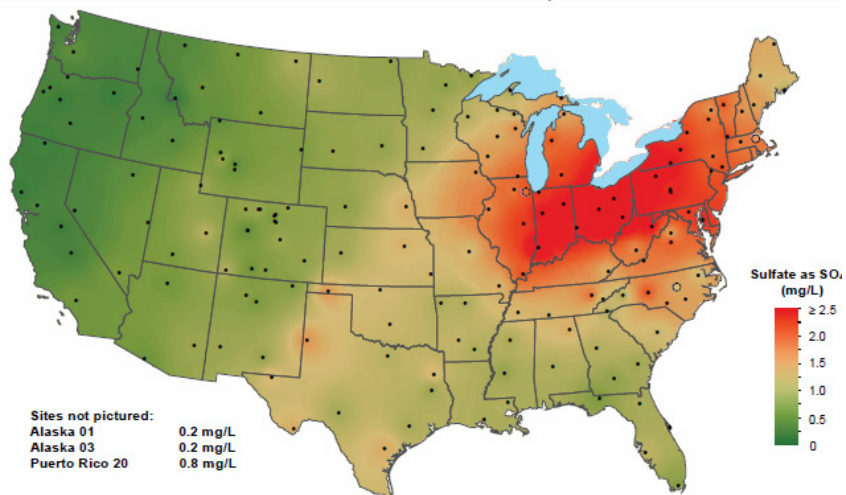


<http://www.eia.gov/todayinenergy/detail.cfm?id=10151>

# Sulfate Deposition

## 1994

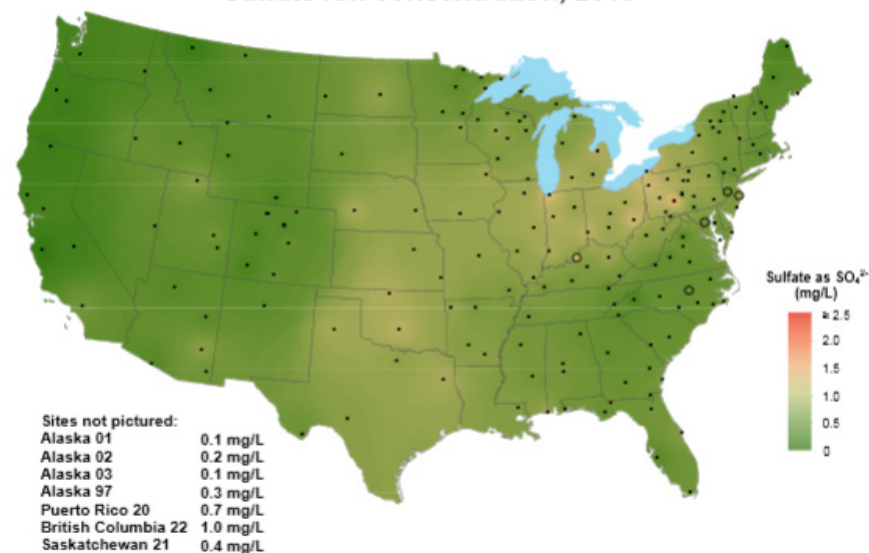
### Sulfate ion concentration, 1994



National Atmospheric Deposition Program/National Trends Network  
<http://nadp.isws.illinois.edu>

## 2013

### Sulfate ion concentration, 2013



National Atmospheric Deposition Program/National Trends Network  
<http://nadp.isws.illinois.edu>

<http://nadp.sws.uiuc.edu/>

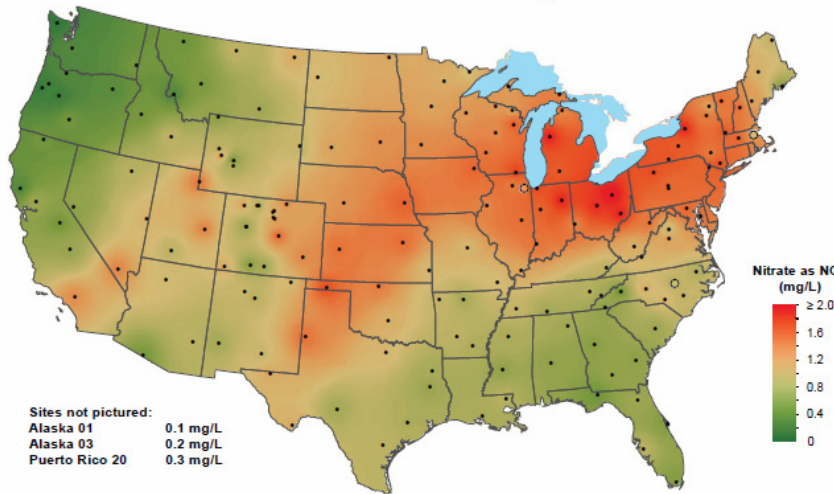


# Nitrate Deposition

1994

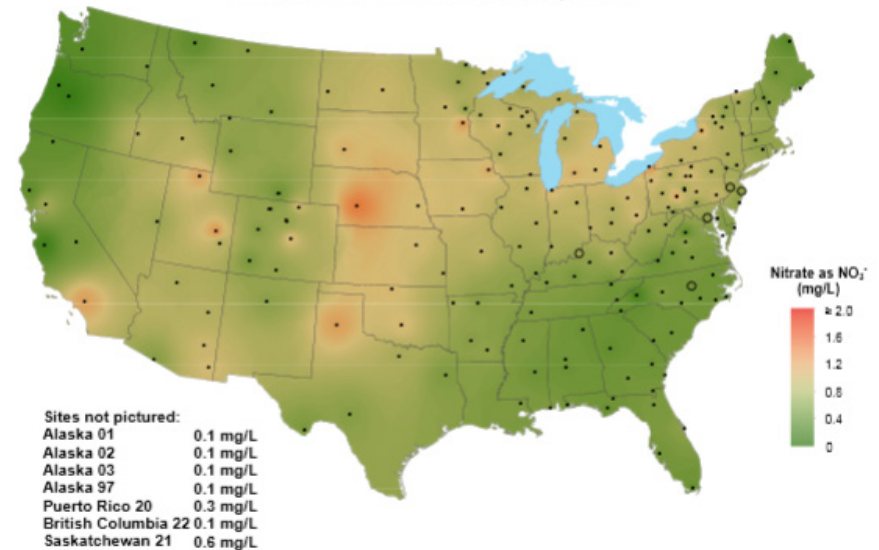
2013

Nitrate ion concentration, 1994



National Atmospheric Deposition Program/National Trends Network  
<http://nadp.isws.illinois.edu>

Nitrate ion concentration, 2013



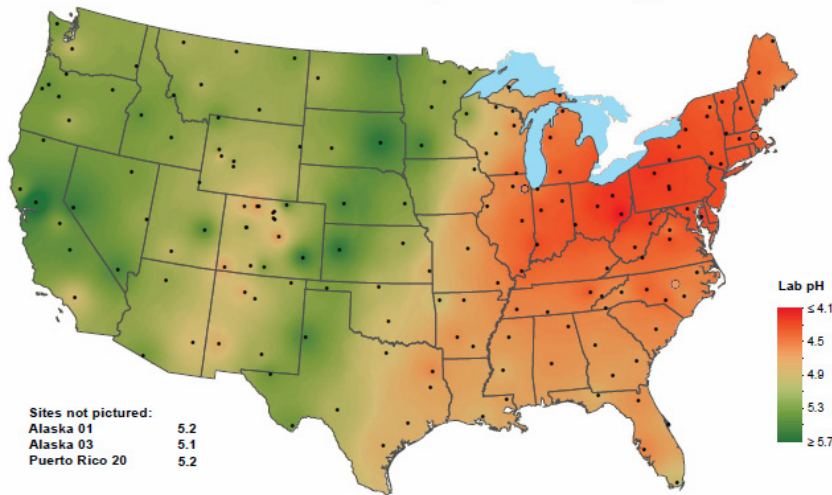
National Atmospheric Deposition Program/National Trends Network  
<http://nadp.isws.illinois.edu>

<http://nadp.sws.uiuc.edu/>

# pH

## 1994

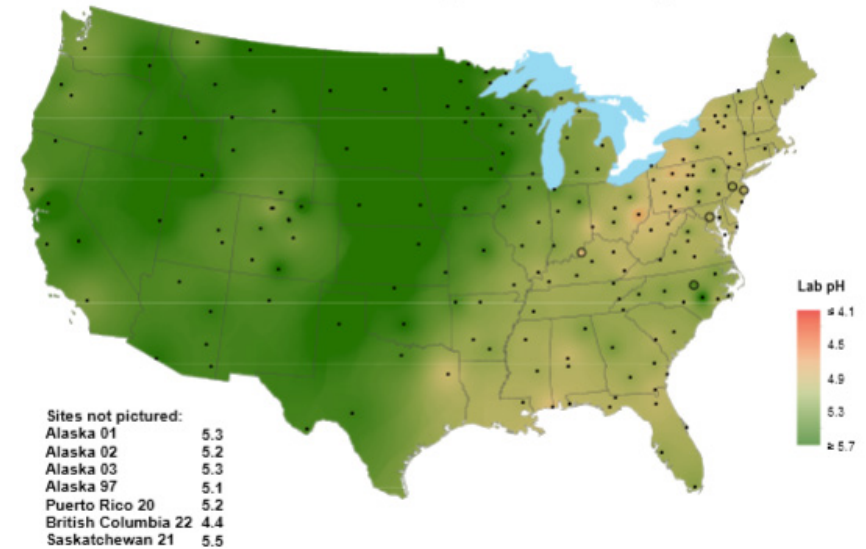
Hydrogen ion concentration as pH from measurements made at the Central Analytical Laboratory, 1994



National Atmospheric Deposition Program/National Trends Network  
<http://nadp.isws.illinois.edu>

## 2013

Hydrogen ion concentration as pH from measurements made at the Central Analytical Laboratory, 2013

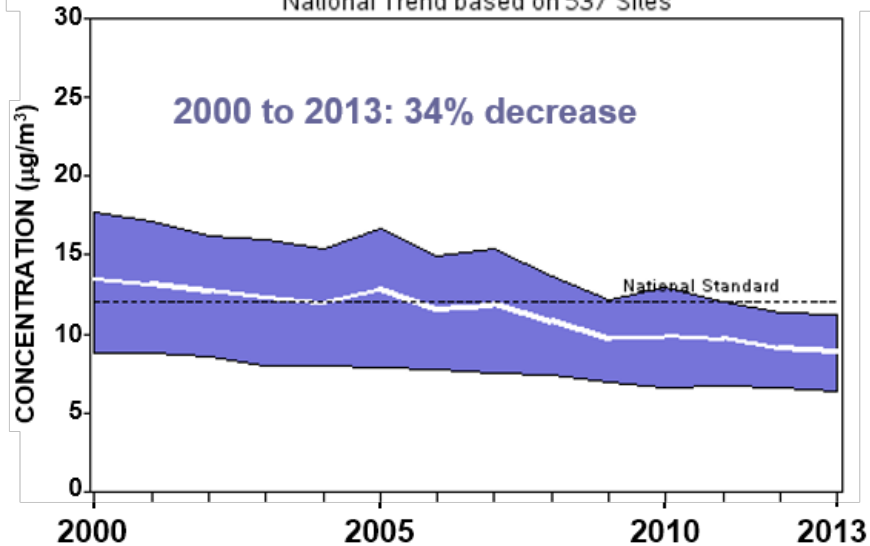


National Atmospheric Deposition Program/National Trends Network  
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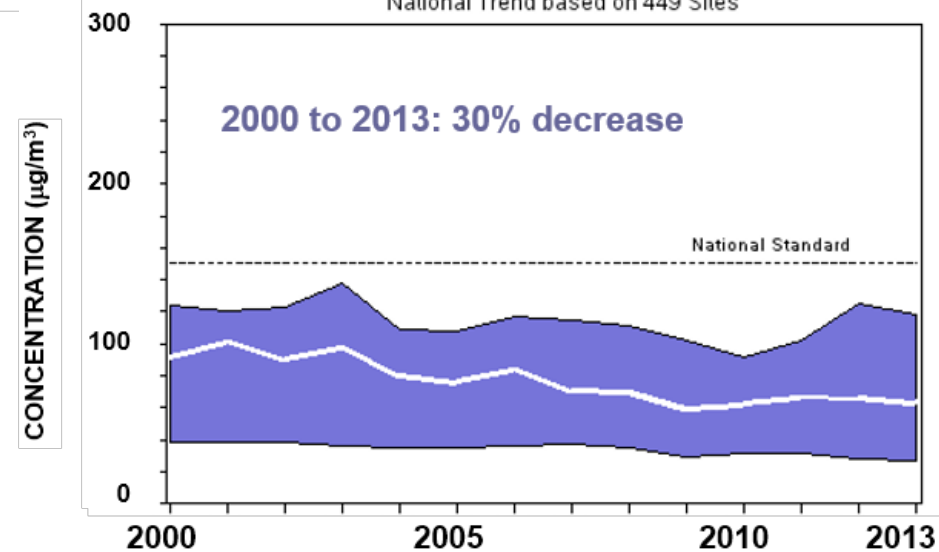
<http://nadp.sws.uiuc.edu/>

# PM Trends

PM2.5 Air Quality, 2000 - 2013  
(Seasonally-Weighted Annual Average)  
National Trend based on 537 Sites

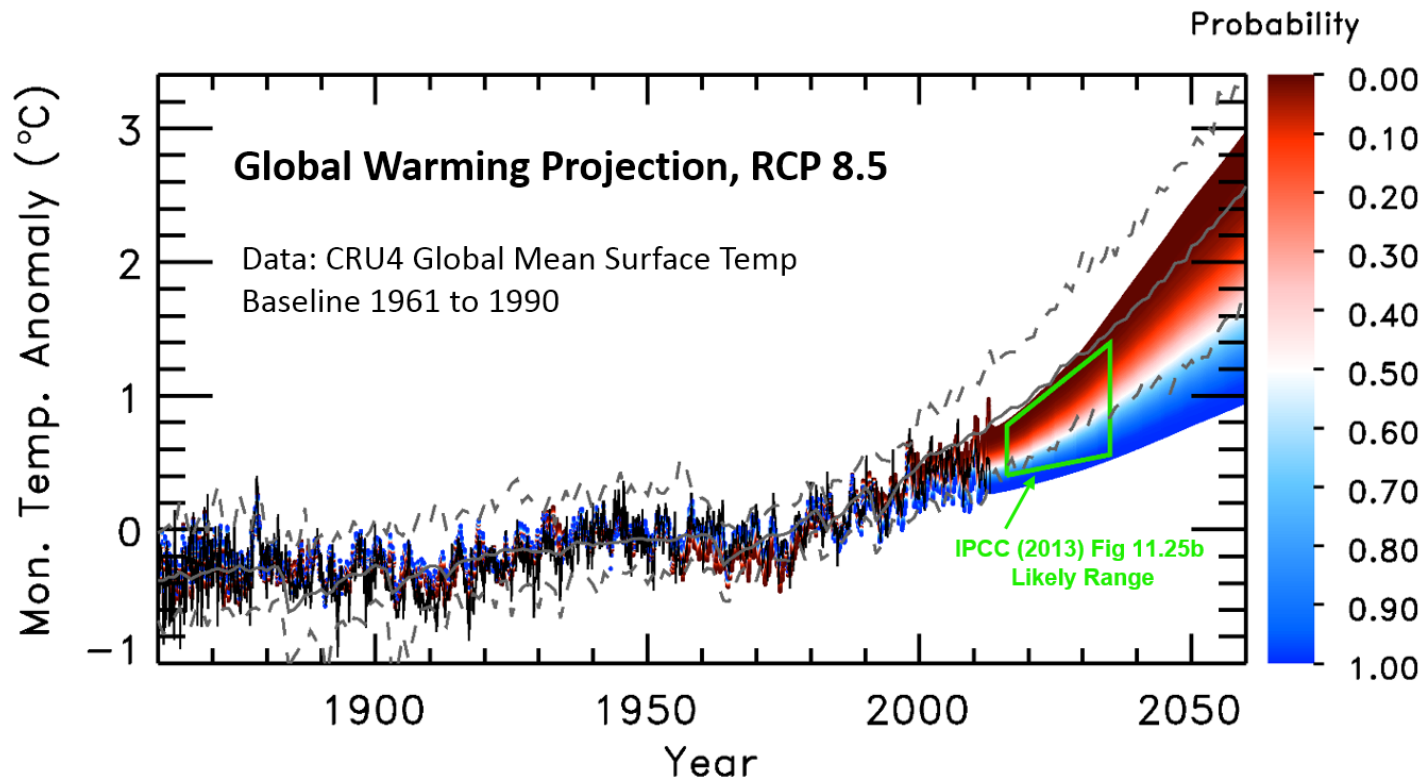


PM10 Air Quality, 2000 - 2013  
(Annual 2nd Maximum 24-Hour Average)  
National Trend based on 449 Sites



<http://www.epa.gov/airtrends/>

# Uncertainty of Aerosol RF Effects Future Climate



If tropospheric aerosols have offset a large fraction of GHG induced warming, then the actual warming that may occur could be considerably *larger* than “best estimate”

If tropospheric aerosols have offset only a tiny fraction of GHG induced warming, then the actual warming that may occur could be considerably *smaller* larger “best estimate”